

AN INVESTIGATION of the RISKS of EXPLOSION
in the OPERATION of FLAME-PROOF
MINING APPARATUS.

by

G. HIBBERD, A.R.T.C.

ProQuest Number: 13905580

All rights reserved

INFORMATION TO ALL USERS

The quality of this reproduction is dependent upon the quality of the copy submitted.

In the unlikely event that the author did not send a complete manuscript and there are missing pages, these will be noted. Also, if material had to be removed, a note will indicate the deletion.



ProQuest 13905580

Published by ProQuest LLC (2019). Copyright of the Dissertation is held by the Author.

All rights reserved.

This work is protected against unauthorized copying under Title 17, United States Code
Microform Edition © ProQuest LLC.

ProQuest LLC.
789 East Eisenhower Parkway
P.O. Box 1346
Ann Arbor, MI 48106 – 1346

Kirkcubright
Muirhead
27/12/29.

The Very Reverend Professor G. Milligan, D.D., D.C.L.,
Clerk of Senate,
The University,
Glasgow,

Dear Sir,

Adverting to your letter, in which I was instructed to amend my thesis, 'An Investigation of the Risks of Explosion in the Operation of Flame Proof Mining Apparatus' and submit it on or before 31st Dec. 1929, I have honestly endeavoured to satisfy the requirements, and now submit my work for adjudication, amended as follows:

- (1) An introduction is given explaining the reasons for undertaking the research, and showing the relationship it bears to the work of investigators who may have had a similar end in view.
See pages I-IV.
- (2) The tabulated statements of results have been arranged as indicated on page V.
- (3) A reprint of a paper published in the 4th Number

of 'The Journal' of The Royal Technical College, and
a mathematical investigation of 'The Drop in
Temperature Produced by the Passage of Hot
Gases through Apertures' are given in
Appendix i, pp. 84-88.

- (4) A bibliography is given in Appendix ii,
pp. 89-90.

Trusting that my work, thus amended,
meets with approval

I am, Dear Sir,

Yours Respectfully,

- i -

AN INVESTIGATION of the RISKS of EXPLOSION in the
OPERATION of FLAME-PROOF MINING APPARATUS.

By G. Hibberd, A.R.T.C.

- - - - -

INTRODUCTION: In May, 1924, the author was awarded the Walter Duncan Research Scholarship by the Governors of the Royal Technical College, Glasgow, and with the approval of the Board of Studies of the College decided to investigate certain problems connected with the safe working of electrical machinery in mines.

At that time the term "flame-proof" was applied somewhat loosely to certain types of machinery intended for use underground and many machines considered flame-proof were later shown to be unsafe. This was chiefly due to the lack of suitable references for the guidance of designers and manufacturers of this type of machinery.

The problem of protecting apparatus was attacked by many eminent investigators, and in searching for a solution, the character of flame and flame movement, the various methods of initiating by electrical means an explosion of gaseous mixtures and methods of preventing excessive rise of pressure in a gaseous explosion in a partially closed vessel had to be thoroughly investigated before the real problem of adequate protection could be approached.

In this preliminary work the experimenters used spherical bronze vessels, e.g., in February 1924, in a paper read before the Institution of Mining Engineers, Professor Douglas Hay and Mr. Ira C.F. Statham gave an account of experiments, carried out in such vessels, and also mentioned that they had tested a large number of switch boxes etc., intended for use in mines. A short time later the Safety in Mines Research Board Paper No. 5 appeared under the joint authorship of I.C.F. Statham and R.V. Wheeler. This paper

also gave results of investigations in which spherical bronze vessels were used, and in addition put forward certain recommendations which they considered necessary so that flame-proof conditions might obtain.

These papers were welcomed and it was recognised that a great step had been made towards the solution of the problem of safe working of electrical machinery in mines. It was felt, however, that the results obtained were only truly applicable to spherical vessels, and some difficulty arose with regard to the application of the recommendations to casings such as were already in use in mines. These casings had been evolved from long experience of the requirements having special regard to economy of space and material, and the ability to withstand the rough usage associated with mining operations. Naturally they differed greatly in material and shape from the spherical vessels used by Hay, Statham and Wheeler.

Many manufacturers had independently evolved machinery which they claimed to be flame-proof, but even after the publication of the above-mentioned papers some doubt existed among them with regard to the margin of safety of their products. This was clearly seen when Sheffield University announced that they were prepared to receive and test Flame-Proof Apparatus and issue certificates if the apparatus satisfactorily passed the tests to which they were submitted. A large number of manufacturers availed themselves of the opportunity and now possess certificates for the apparatus which passed the required tests. In the opinion of many, testing of such apparatus after application of recommendations should not be necessary. Testing might be resorted to as a precautionary measure but recommendations issued should be such that after their application the possibility of failure should not exist.

The author, therefore, decided, after careful

perusal of available literature (a bibliography is given in an appendix) and after consideration of the results and recommendations given by investigators to confine himself to a research in which the vessels used would be typical examples of modern mining switch-gear casings, and to endeavour to present results in such a form that designers and others interested would have no difficulty in deciding the dimensions of flanges, gaps, etc., for casings which had to be flame-proof.

Flanged casings were obtained and experiments performed as here described. The results obtained have been presented in the form of graphs from which a suitable design for the flange of any casing and a suitable depth for the flame-proof gap can readily be found.

Abnormal conditions such as the "blowing" of a fuse under large overload conditions, the presence of coal dust in addition to gas, the effect of increased temperature of casings due to working conditions or electrical defects have all been considered. In addition the possibility of passage of flame at the bearings of protected motors has been fully dealt with.

So far as the author is aware these problems have not been discussed in any paper published in this country, and he is confident that the present work will be a useful addition to the list of publications dealing with Flame-Proof Apparatus.

SUMMARY:

- I. The work was undertaken because of the necessity of presenting results in a manner that could easily be understood and applied with confidence to actual mining switch-gear casings, etc.
2. This could only be done by experimenting with casings such as were found most suitable (with regard to design and material) for use in mines.

3. The problems dealt with are entirely practical and practical solutions are offered.
4. The work bears no distinct relationship to other works which have the same object in view - viz: absolute safety in operation under all conditions likely to be encountered - but forms what should be a useful addition to the list of works on the subject of "Flame-Proof Apparatus", since it deals with problems about which very little was hitherto known.

In Appendix No. i a reprint of a paper by the author - published in the 4th Number of The Journal of The Royal Technical College - is presented together with a mathematical analysis of the drop in temperature produced by the passage of hot gases through apertures.

In Appendix No. ii a list is given of the more important works perused by the author. The writer is much indebted to their authors and freely acknowledges the assistance he received by studying their papers. In this connection it is desired to pay special tribute to the work of Professor W.M. Thornton, D.Sc., D. Eng., whose many excellent papers were a constant source of inspiration.

CONTENTS.

	<u>Page :</u>
Introduction...	1 - iv.
Theory of Protection.	2 - 9
Flame Proof Enclosure	9
Flame Proof Test	9 - 10
Explosion Chamber	10
Explosive Mixtures Used	10
Ignition of Gas Mixtures...	11
Apparatus Used	11
Junction Box with Grooved Flanges	11 - 13.
Gate - End Circuit Breaker	13 - 17
Critical Lengths	17 - 25
Variation Igniting Current.	26 - 28
Effect of Addition of Coal Dust) to Gas Mixtures	28 - 31
Measurement of Pressures Developed by Explosions of Gas Mixtures.	31 - 35
Webbed Covers	35 - 36
Heated Casings	36 - 41
Analysis of Gas Mixtures	42 - 47
Permissible Amount of Wear of Bearings	48 - 53
General Summary of Results and Conclusions ..	53 - 57
Tabulated Statements of Results of Experiments: Critical Lengths (Nos. 1 - 209)...	58 - 62
Variation of Igniting Current (1st Series: Nos. 209-340)...	62 - 65
Coal Dust added to Gas Mixtures (Nos. 340 - 444) ...	65 - 67
Variation of Igniting Current (2nd. Series. Nos. 444- 538)	67 - 69
Webbed Covers etc. (Nos. 538 - 608) ...	69 - 71
Heated Casings (Nos. 609 - 804) ...	72 - 79
Permissible Wear at Shaft and Bearing (1st Series Nos. 805 - 874) ...	80 - 82
Permissible Wear at Shaft and Bearing (2nd. Series Nos. 874 - 903) ...	82

CONTENTS (Continued).

	Page
Appendix (i) Introduction	84
Reprint of Paper on the Methods of Preventing Electrical Apparatus from Starting Explosions in Mines.....	INSERTED BETWEEN PAGES 84 AND 85
The drop in Temperature produced by the passage of Hot Gases through apertures.	85-88
Appendix ii	
Bibliography.....	89-90

- 1 -

AN INVESTIGATION OF THE RISKS OF EXPLOSIONS IN THE
OPERATION OF FLAME-PROOF MINING APPARATUS.

- - - - -

Electricity by reason of its adaptability is now used extensively in coal mines, but its introduction has been accompanied by many dangers, chief among which are the possibility of ignition of coal dust and inflammable mixtures of methane and air (firedamp). A parallel potential source of ignition may be said to exist in the flame safety lamp, but the protection afforded by gauzes is such that complete confidence may be placed in modern types of lamps, and the design is such that omission of parts during assembly is inexcusable.

Unfortunately absolute reliance, in electrical apparatus for use underground, is not justified at present, from a safety point of view. Many protective devices have been introduced, (quite a number displaying the main features of the safety lamp) to guard against the possibility of ignition of gas and dust, but with all devices, great care in inspection maintenance and assembly of parts must be exercised if the apparatus to which they are fitted is to remain safe.

Many vague ideas exist regarding the exact nature and function of these devices, consequently, protected apparatus may become unsafe through deterioration, neglect or lack of knowledge of proper methods of maintenance of these contrivances.

Few of the original problems of protection of apparatus remain to be solved, and progress in this direction is such that the day is not far distant when it will be possible to install Apparatus, safe in every respect, and be assured that it will retain its safety features as long as the apparatus is in commission.

The present Research was conducted under the direction of the Professors of Mining and Electrical Engineering in the Royal Technical College, Glasgow, and is chiefly concerned with such methods as might be applied to underground electrical Apparatus in order to afford adequate protection from explosive mixtures of gas and air.

Demonstrations were given annually to the Senior Engineering Students of the College.

THEORY OF PROTECTION:

Occurrence of Methane:-

The most common explosive gas mixture found in coal mines consists of methane and air, and is generally referred to as "firedamp." Methane is given off naturally as the coal is worked. The gas was probably entrapped during the early stages in the formation of coal, but dissociation of coal substance by great pressure and high temperatures acting over long periods of time, may partly account for its presence. If the gas were allowed to accumulate highly explosive mixtures would in all probability be formed so that to keep mixtures within safe limits dilution with air is necessary.

The "Coal Mines Act" requires that "a sufficient quantity of air shall be constantly circulated to dilute and render harmless all noxious and inflammable gases" to such an extent that the workings, shall be in a fit state for working and passing therein."

Under normal conditions every endeavour is made to maintain the required standards of ventilation, but abnormal occurrences may allow of accumulation of gas where it is least desired e.g. in the vicinity of electrical apparatus. If the system of ventilation is interrupted, say, by a fall of material in an airway, or a stoppage of the fan, gas will accumulate in dangerous proportions in "gassy" sections of the mine. The workings of the mine may also be flooded by the sudden liberation of large quantities of gas

from an "outburst." In addition to the above the presence of "blowers" of gas necessitate special attention to ventilation if the formation of explosive mixtures of gas and air is to be prevented.

PRESENCE OF GAS IN APPARATUS:-

Electrical apparatus is subjected to variation of temperature during and subsequent to working periods. Expansion and contraction of the metal casings and gaseous contents will accompany each rise and fall in temperature. It follows therefore that in the process of cooling after working periods, gas will be "drawn" into the box or casing, through the joints, to maintain equilibrium of pressure with the outside. If the surrounding atmosphere contains methane, a highly inflammable mixture may thus be introduced into the casing and thereby create the required conditions for an explosion. Openings in the casing may permit of natural diffusion of gases, so that independent of temperature variations gas may accumulate in electrical apparatus. If the above conditions exist all that is necessary to initiate an internal explosion is a spark or electrical arc, and in event of flames being communicated to the outside of the apparatus, an explosion having far reaching and serious effects may be the final result. Gas tight apparatus or hermetically sealed casings would seem to be the remedy, since accumulation of gases inside casings would be prevented, but difficulty of inspection, renewal and repair, together with rough usage, and deterioration of sealing material in the humid atmospheres of mines, render the application of this measure impracticable. In the absence of sealing methods we cannot prevent the accumulation of gas in casings, so that the danger of ignition is ever present. The real danger lies in the transmission of high temperature gases and flames to the outside, and all protective methods aim at the prevention of this.

IGNITION OF FIREDAMP:-

The mixtures of C.H.₄ (Methane) and air which on burning evolve the greatest amount of heat contain approximately 9.5 per cent of methane, and are highly explosive. In these mixtures the methane and oxygen are present in approximately the proportions necessary for complete chemical combination. If the percentage of methane be reduced below 9.5 per cent less heat will be generated by the burning mixtures since obviously the amount of heat will depend on the amount of inflammable gas present. In mixtures containing over 9.5 per cent of methane, there is insufficient oxygen present to effect complete combustion. If the percentage of methane be continually reduced, a point is reached at which only sufficient heat is generated by the burning portion to raise an adjacent portion to the ignition temperature. With further reduction flame propagation does not take place. A similar state of affairs is reached by increasing the percentage of methane above 9.5 per cent. The points at which self propagation of flame ceases are termed the "limits of inflammability" and are 5.3 per cent and 14.8 per cent respectively. It is evident that any variation in the condition of experiment which alters the amount of heat transferred from the burning portion to an adjacent portion will alter the limits of inflammability. Factors which affect the limits are:- material and dimensions of containing vessel, initial temperature and pressure of the mixture, and state of mixture i.e. whether turbulent or quiescent.

Wheeler has shown that the most explosive mixtures are not the most readily ignitable mixtures, and also that slow burning mixtures will transmit flame through small passages when more rapidly burning mixtures would not.

Firedamp has a peculiar property by virtue of which a source of ignition must be in contact with an inflammable mixture for a definite period of time before ignition takes place, e.g. if the temperature of the source be 650°C. 10 seconds elapse between the introduction of the source and ignition of mixture; if the

source is at a temperature of 1000°C ignition occurs in less than one second. The time is not occupied in bringing the mixture to the temperature of the source (this is probably effected in a small fraction of a second) but may be due to the rate at which chemical combination takes place. The rate of combination is quite measurable at 300°C and increases rapidly with temperature. This combination results in liberation of heat which raises the temperature of the surrounding mixture. Thus the rate of combination increases rapidly until finally the temperature is such that combination takes place throughout the mixture almost instantaneously. It follows therefore that the higher the temperature of the source the shorter will be the period necessary to raise the temperature of the gas to that at which instantaneous combination is effected throughout the mixture. This is important from a mining point of view, since if flame or hot gases from an explosion in a casing, can be cooled considerably before reaching the outside, the chances of external ignition will be slight. The flames from an explosion are of short duration, and if considerably cooled, would not be in contact with the external mixture for the period of time necessary to initiate explosion at the temperature of the "cooled" gases.

POSSIBLE SOURCES OF IGNITION:-

It has been shown that any sustained flame, however small, will cause ignition of firedamp mixtures. The arc produced at a switch or controller when a circuit is opened, is of short duration but the temperature is very high and is quite capable of causing ignition of gas mixtures. Gas surrounding a fuse will be ignited if the fuse is "blown." Sparking at commutators of D.C. motors or slip rings of A. C. motors must be regarded as a potential source of ignition. Broken coils in stators and rotors of induction motors must also be regarded as dangerous, when gas is present.

TRANSMISSION OF FLAMES FROM APPARATUS TO THE OUTSIDE:-

Under the conditions described above, if the motor or

switch gear is not protected external explosion with its accompany:
ing destructive effects will inevitably result. If the apparatus
is^{is} adequately protected the possibility of external ignition is very
great. If the casings are protected but constructed of material
not sufficiently strong to withstand the force of the explosion,
the conditions are equivalent to no protection. It is therefore
apparent that in addition to preventing passage of flame a casing
must be able to withstand the pressure developed by an explosion.

GAS EXPLOSIONS IN CLOSED VESSELS:-

When an explosive mixture of methane and air is present
in a metal casing and a source of ignition is applied (say an
electric spark), flame spreads rapidly from the point of ignition
throughout the mixture at a rate depending on the amount of methane
present. Actual measurement of the temperature of explosion have
recorded 1900°C - 2100°C as the highest temperature attained.
Theoretically, the temperature should be considerably higher, but
the proximity of the cold walls of the containing vessel and
probable dissociation of methane at high temperatures, each
absorbing heat, probably together account for difference between
the theoretical and recorded temperatures. Since excessive
temperature development is prevented, pressure development is
restrained in a like manner.

The maximum pressure recorded for methane-air explosions
in completely closed vessels, is 108 lbs per sq. inch. If the
joints of the vessel be separated by a small amount the pressure
is reduced considerably, since part of the mixture is expelled
unburnt through the space at the joints, when ignition takes
place. Hot gases produced by an explosion will also escape if a
gap is left at the joints of the casing.

METHODS OF PROTECTION ADOPTED:-

From the foregoing it will be evident that the provision
of a large cooling surface is essential, and in methods of
protection adopted, this aspect has been kept in view. In a few

types of devices, cooling of flame has apparently been the principal object aimed at, and the possibility of obtaining considerable reduction of pressure while effectively cooling the flame has not received the attention it merits. Methods of protection of apparatus have so far proceeded mainly along four lines. One of the earliest methods made use of gauze and perforated plates as protective devices. These devices are founded on sound physical principles and involve the provision of large cooling surfaces consisting of wire gauze. The gauzes were separated by distance pieces, but numerous layers were necessary before the desired effect was achieved. In early applications gauzes were used for cooling purposes only; the attainment of considerable reduction of pressure seems to have been lost sight of, with the result that one of the necessary requirements of adequate protection was not obtained. In addition coal dust was free to enter the apparatus and also lodged on the gauzes and in the interstices, interrupting ventilation. The gauzes would be readily corroded, so that frequent inspection and renewal would probably be necessary. Recent research has shown that if properly applied, and entry of coal dust prevented by layers of tin or lead foil, satisfactory results may be obtained and considerable lowering of pressure be produced in event of explosion.

Labyrinth and plate protection was also tried. The device consisted of large numbers of thin metal plates so arranged and spaced that escaping gases and flames had to traverse long tortuous passages before reaching the outside atmosphere. In passing through the labyrinth, effective cooling was obtained but such was the resistance offered to the escaping gases that only slight reduction of pressure resulted. This appliance requires accurate machining and fitting, and is easily damaged. Coal dust will lodge between the plates and reduce the efficiency of the device. This device has not been extensively applied for the above reasons.

Other contrivances adopted are known as relief valves. They consist of valves, normally held closed by springs, set to open a specified distance immediately the pressure rises above a certain amount. In passing through the valves the hot gases are brought into contact with large cooling surfaces. These devices are also most effective flame coolers, but fail to cause appreciable reduction of pressure.

Special devices have special disadvantages, but disadvantages common to all of them may be enumerated as follows.

1. The addition of devices to any apparatus means additional care in inspection and maintenance of apparatus, which of course increases cost of upkeep.
2. Machining of parts and application to casings tends to increasing price of product.
3. Special devices require expert handling, and ignorance or carelessness in handling may render the device useless and the apparatus unsafe.
4. Accumulation of dirt, grease, rust, etc., may hinder the proper functioning of the device.
5. Concentration of escaping gases at small area is not desirable.
6. Doubtful whether flame-cooling and pressure reduction can each be carried to safe limits in the same device.

FLANGE PROTECTION:-

This is perhaps the simplest and most effective method of cooling gases and lowering pressure of explosion.

It consists of providing broad flanges at all joints in casings, the flanges being so proportioned that the required degree of cooling is obtained, and by having a gap of predetermined depth between the flanges, excessive pressure development is prevented. The great advantages of flanges are simplicity, low cost and permanence. There is nothing that can go out of order, and maintenance

and renewal costs are nil. The U. S. Bureau of Mines advocates metal to metal contact of not less than one inch at all joints. This of course means high pressure in event of an explosion, and necessitates very strong and rugged casings. No advantage is taken of the property called "lag on ignition." All that is required is cooling of flame to 700°C or thereby, and as already pointed out the lag on ignition at this temperature is such that external ignition cannot take place since the temperature is not maintained for a sufficient length of time.

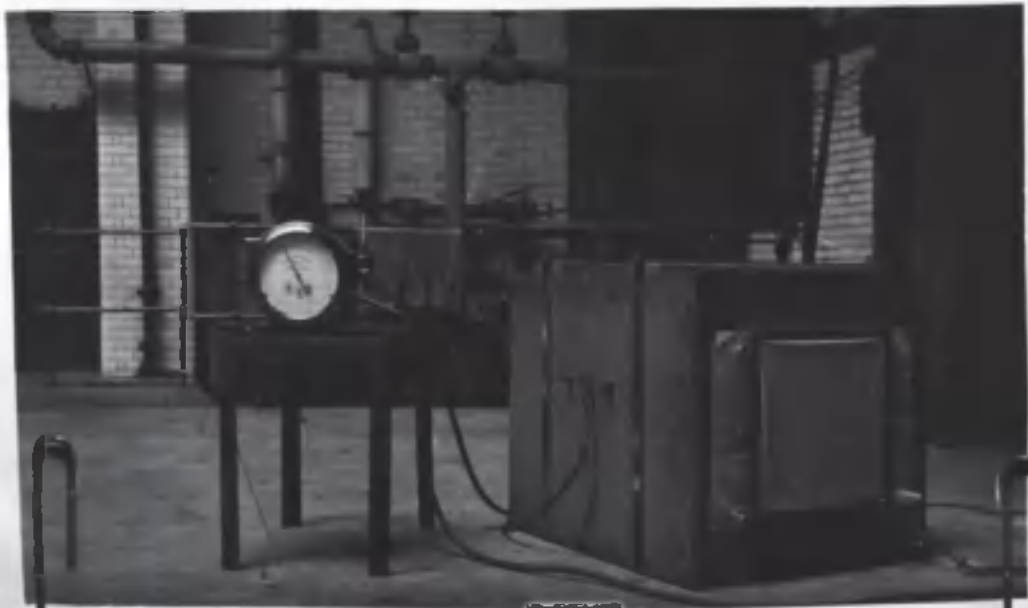
Since flange protection in our opinion offers the greatest possibilities for adequate protection of electrical apparatus for use underground, it was chosen for this particular investigation, and all casings used in the experiments were fitted with broad flanges.

FLAME PROOF ENCLOSURE:-

The "British Engineering Standards Association" give the following definition of a flame-proof enclosure:- "A flame proof enclosure is one that will withstand without injury any explosion that may occur within it under conditions of the rating of the apparatus (and recognised overloads, if any associated therewith) and will prevent the transmission of flames such as will ignite any inflammable mixture which may be present in the surrounding atmosphere."

THE FLAME-PROOF TEST:-

The following test is applied to casings to determine whether they satisfy the requirements of the above definition:- The apparatus to be tested is placed in an "Explosion Chamber" and an explosive gas mixture is introduced into it. The Explosion Chamber is now closed and filled with an explosive gas mixture, after which the mixture in the casing is ignited and the result noted. If the external mixture (i.e. the mixture in the chamber surrounding the apparatus) is ignited, flame has passed from the apparatus, and



under the conditions of the test, it is unsafe. If no external ignition occurs, the mixture is ignited by some means, to prove its explosibility. The tests are repeated until a consecutive number of similar results are obtained, and according to results the apparatus is declared safe or unsafe under the conditions of the test.

THE EXPLOSION CHAMBER:-

The "Explosion Chamber" used in the investigation consisted of a wooden box, fitted with terminals and a gas tap, and of the following dimensions:- Length 2 ft. 10 ins. Width 2 ft. 7 inches, Height 2 ft. 6 ins. To prevent damage in event of an external ignition, safety valves or windows were fitted into the ends of the box. These windows were 15 ins square and during the filling stage were blocked with paper to prevent leakage. When external ignition occurred the paper windows were burst, allowing of dissipation of energy without damage to the box. If external ignition did not occur a light on the end of a stick was thrust through the paper to ignite the mixture and prove its explosibility.

METHOD OF OBTAINING EXPLOSIVE MIXTURES:-

The explosive mixtures were obtained by mixing coal-gas and air in known proportions. Objection to this might be made on the ground that coal gas is not found in mines, and that firedamp, the explosive mixture generally found in mines, should have been used instead. It may be pointed out however that the surrounding gas mixture is really only an indicator the purpose of which is to indicate when flame has passed from the internal explosion. Evidently then a mixture of coal-gas and air will serve the purpose admirably and such a mixture has the additional advantage that it is more sensitive to ignition than methane and air (firedamp). Considered in this light results obtained, using coal gas and air, will not vary greatly from those obtained using methane and air and any difference will be in favour of coal gas since a greater margin of safety will be obtained.



METHOD OF IGNITING THE EXPLOSIVE MIXTURES:-

The gas mixtures in the casing under test were ignited electrically by "blowing" a fuse, power being taken from a 250 volt. D.C. supply with a variable resistance in circuit. The fuse wire was No. 24 S.W.G. Tin and from $1\frac{1}{2}$ to 2 ins.in length.

APPARATUS TESTED IN THE EXPERIMENTS:-

Each piece of apparatus tested in the experiments was fitted with fixed flanges at the joints of cover and casing. The casings were typical examples of modern mining switch-gear casings. These casings were readily adapted for experimental purposes, by tapping the sides or ends to receive a gas tap and a lead-in plug for the conductors. This adaptation did not in any way interfere with the protective features of the boxes. The use of typical switch gear casings is a step which commends itself, in that any weakness or defect in the casings could be readily detected and suggestions made for general improvement.

EXPERIMENTAL: JUNCTION BOX WITH GROOVED FLANGES:-

The first set of experiments was carried out on a junction box, detached from a gate-end circuit breaker. This box had a grooved flange 0.5 ins broad, into which fitted a cotton gasket, so that when the cover was bolted into position a very tight joint was secured. The ends of the box were blocked with pieces of wood, bored to receive terminals and a gas pipe. Leads were taken from the terminals inside the explosion chamber to those on the junction box, and short leads from the latter were bridged with a piece of fuse wire, inside the junction box. These connections having been made, the cover was bolted into position, an explosive gas mixture was introduced, and the gas pipe plugged to prevent leakage. The "windows" of the explosion chamber were now blocked with paper and the junction box surrounded with an explosive atmosphere. By closing the switch the fuse was "blown", igniting the gas in the junction box, the effect produced being noted. A large number of tests as above were made with the gasket

in position and then with it removed. Typical results are shown in the following table.

TABLE 1.

Tests Made on Junction Box with Grooved Flanges 0.5 ins.

Broad. Volume of Box.

0.08 Cubic ft.

Cubic feet of Gas.

Junction Box.	Explosion Chamber.	Current in Amps. (Igniting)	Breadth of Flange.	Ignition.		Remarks.
				Internal.	External.	
0.020.	3.400.	26	0.5	Yes.	No.	Tape in position in Groove.
"	"	"	"	"	"	"
"	"	"	"	"	"	"
"	"	"	"	"	"	Signs of Charring.
"	"	"	"	"	Yes.	Tape removed from Groove.
"	"	"	"	"	"	"
"	"	"	"	"	"	"
"	"	"	"	"	"	"

CONCLUSIONS:-

It was found that with the gasket in position flame did not pass between the flanges, but after a few ignitions distinct signs of charring were seen, showing that if the experiments were continued the gasket might possibly be burned out, and thereby provide a passage for flame to the outside mixture. With the gasket removed external ignition took place in every experiment, indicating that a flange 0.5" broad is not in itself an adequate protective device. The tape or gasket is mechanically weak and would soon rot



in the humid atmosphere of mines, and for this reason alone could scarcely be recommended.

If for any reason the gasket had to be removed great care would have to be exercised in replacing it correctly in position in the groove, to prevent risk of providing passage for flame. Above ground with abundant light this would require ordinary care only, but underground where lighting arrangements are often far from perfect it would be a more difficult matter.

It would be better to avoid the dangers which might arise, by removing the gasket altogether and increasing the breadth of flange to one inch or more if necessary. If this measure were adopted, we would have a piece of apparatus made flame proof at the beginning and always retaining its properties, unlike a device depending for its flame proof properties on a material which readily deteriorates and hence gradually loses its properties.

EXPERIMENTAL:- PRELIMINARY:-

GATE END CIRCUIT - BREAKER WITH ROUGH MACHINED FLANGES:-

The next piece of apparatus^{tested} was a gate-end box fitted with circuit breakers set to trip at 17.3 amperes. Part of the gear was removed to adapt the box for experimental purposes. As shown in the *photograph* the cover is hinged and held in position by means of eight $\frac{5}{8}$ inch. diameter bolts. The flanges were 1 $\frac{1}{2}$ inches broad and rough machined. The junction box used in the above experiments was re-attached to the gate-end box and tests similar to those already described were carried out.

After one or two tests in which external ignition did not occur, it was thought that the combined junction box and gate end box was quite safe under the conditions of the test. To confirm this, one more experiment under exactly the same conditions as the others was made. A violent external explosion resulted when the switch was closed, and on removing the junction box cover the cotton gasket was found to be completely burned through in parts, and badly charred in others. The flame proof properties of the

combined apparatus were thus entirely destroyed. The junction box was removed and a piece of wood, bored to receive a gas pipe was bolted into position in its place.

A number of experiments were now carried out with the flanges of the cover and box in contact, the only gaps being those formed by the rough machining, the largest of which did not exceed .01 inches. Under these conditions it was not found possible to initiate an external explosion. By placing pieces of metal between the flanges it was possible to obtain gaps of any desired depth all round the perimeter of the cover. Experiments were thus carried out at various depths of gap until external ignition was produced by flame passing between the flanges from an internal ignition. The following table shows the average results of a large number of experiments.

TABLE 2.

Tests on Gate End Circuit Breaker. Box. Flanges $1\frac{1}{2}$ ins.

Broad. Volume 1.55 Cub. ft.

Cubic feet of Gas.		Igniting Current in Amperes.	Breadth of Flange.	Depth of Gap.	Ignition.		Remarks
Gate- End Box.	Explosion Chamber.				Internal.	External	
0.350	3.400	28	1.75ins.	0.025 ins.	Yes.	No.	Ext. Mixture Exp.
0.400	3.600	"	"	"	"	"	"
0.400	3.666	"	"	"	"	"	"
0.419	4.043	"	"	0.038ins.	"	"	"
0.420	4.039	"	"	"	"	"	"
0.451	4.023	"	"	"	"	"	"
0.490	4.400	"	"	0.055ins.	"	Yes.	Very Violent Expl. Chamber Shattered

In the last experiment of this series the Explosion Chamber was shattered. Apparently the "safety valve" paper windows did not operate rapidly enough to prevent damage to the box. A gap of 0.055 inches all round the Gate End Box would provide a large release area, and at that particular depth of gap the flanges could not have a great cooling effect on the flame from the internal explosion. When the internal ignition occurred a large mass of intensely hot flame would be projected into the external mixture from all round the box. The pressure wave preceding the flame would create considerable mechanical disturbance in the external mixture and followed immediately by a large body of flame, ignition would take place under conditions approaching detonation. Under these abnormal conditions the pressure rise would be so rapid that maximum pressure would be attained before relief was obtained by rupturing of the paper safety valves. The energy of the explosion would therefore be expended on the sides of the explosion chamber instead of being dissipated through the "safety valves." The result was of course destruction of the explosion chamber since it was not designed to withstand such pressures.

COOLING EFFECT OF FLANGES ON FLAME:-

To obtain some information on the flame that causes ignition of the external mixtures, the following experiments were performed with the gate-end box in the "open", i.e. surrounded by air. An explosive mixture was admitted to "box" in the usual manner and ignited, the distance the flame extended beyond the flanges being noted approximately. These tests were performed with the same depths of gap as were used when the box was in the explosion chamber. With the exception of the first series, i.e. with flanges in contact with each other, flame was observed in every case, often blue in colour and extending only a short distance beyond the flanges. This was chiefly noticeable with gaps up to 0.025 inches in depth. In these cases cooling had been carried

out to a successful degree, and in some of the experiments in which weak mixtures were used the flame seemed to emerge at a relatively slow rate, and never exceeded one inch in length. At gaps greater than 0.025 inch the flame was brownish-red in colour and extended 3-4 inches. At gaps at which external ignition was known to occur bright red flames streaked with white, extended more than six inches beyond the flanges, when ignition took place. The flame was visible all round the casing.

IGNITION OF GAS MIXTURES AT "MAKING" and "BREAKING" CONTACT in the GATE.

END BOX:-

The gear removed from the box was now replaced and a number of experiments carried out to determine whether an explosion of gas could be initiated at "making" or breaking contact. A piece of strong twine was attached to the handle of the gate-end switch(which was in the OFF position), an explosive mixture was admitted to the box, then after closing the other switches, contact was made in the gate-end box by pulling the cord. The circuit-breakers were set to trip at 17.3 amperes, but as there were only 9 ohms. resistance in the circuit the short circuit current would be 27.8 amperes. It follows therefore that no sooner was contact made than it was broken by the circuit breakers operating. The flash thus produced ignited the gas mixtures in the casing. Extra resistance was introduced and the current lowered to 15 amperes. With this reduced current ignition occurred in each experiment at "break", but on one occasion ignition took place at "make". This was probably due to dirty contacts, because when they were cleaned no further ignitions occurred at "make."

CONCLUSIONS:-

The combined gate-end box and junction box may be represented as two separate chambers with a short inter-communicating passage. It has been shown that the operation of the overload releases causes ignition of a gas mixture, and in event of such an occurrence in the combined apparatus the mixture in the smaller



FIG 1 :- PLAN OF SMALL SWITCH BOX.



FIG 2 :- SECTION SHOWING LOOSE
LOOSE FLANGES IN
POSITION
FORMING A GAP.



FIG 3 :- LOOSE FLANGE.

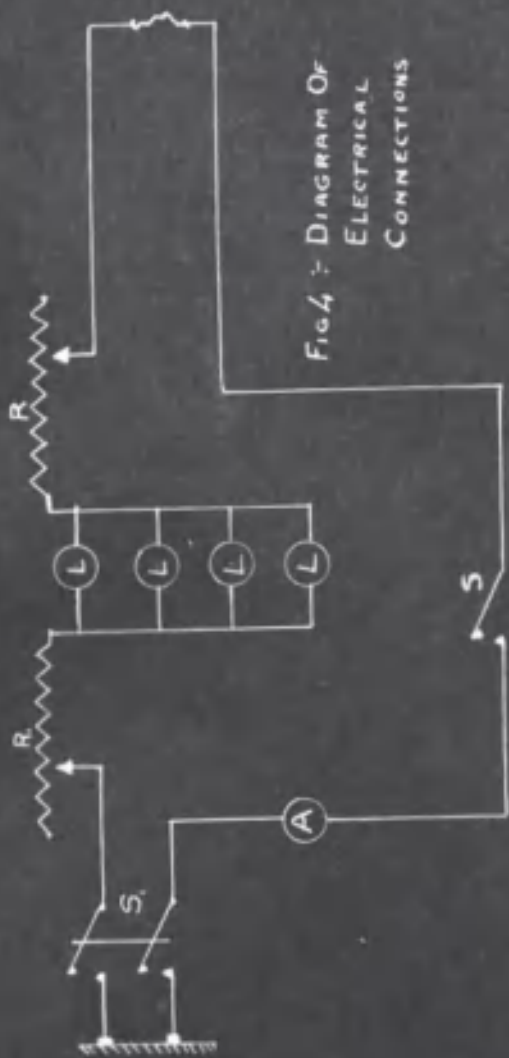


FIG 4 :- DIAGRAM OF
ELECTRICAL
CONNECTIONS



FIG 5 :- A, STUD HOLE AT PRESENT
B, STUD HOLE BOTTOMED



chamber would be compressed before ignition so that higher pressures would be developed. This probably accounts for the sudden rupturing and burning of the cotton gasket. With a piece of wood covering the opening over which the junction box is normally attached, the gate-end box is flame proof with gaps up to 0.038 ins. in depth between the flanges all round the perimeter. With the studs in position and the cover secured, there is little danger of an external explosion if the opening referred to above is well covered. In passing, it might be well to mention that the stud hole at the hinge communicates with the inside of the casing (the others do not). It therefore follows that if this stud is omitted the cooling properties of the flanges are of no avail because we have a direct passage from the inside of the box to the outside atmosphere. This could be prevented by bottoming the stud hole, i.e. by covering the bottom of hole with metal as shown in the fig. 5. If this alteration were made the value of the box from a flame-proof point of view would be considerably increased. The failure of the combined apparatus by the burning of the gasket is a good illustration of how adequate protection of a piece of apparatus (such as by broad flanges of gate-end box) may be rendered useless by an inadequately protected piece of apparatus (junction box with narrow flange and gasket) which is connected to it. The remedy is protection to the same degree of safety for all parts of same apparatus, always remembering that adequate protection implies sufficient strength to withstand the explosion and ability to cool the gases from the explosion.

CRITICAL LENGTH OF OPENING:-

In this series of experiments an effort was made to determine the maximum length of opening for any particular depth of gap between flanges of given breadth, which prevents passage of flame capable of igniting an explosive mixture of gases. The apparatus used consisted of two switch boxes of different volumes and fitted with flanges of different breadths. The volume of the smaller box was approximately 0.120 cub. ft. that of the larger

2.352 cub. ft., while the former was fitted with flanges 1 inch broad and the latter with flanges $1\frac{3}{4}$ inch broad. The flanges were smooth machined.

The smaller box was used in the first part of the investigation. Loose sheet metal flanges (0.010 inch thick) of shape shown in fig. 3. were prepared and placed between the flanges of the cover and the box, so that when the nuts were tightened, the only passage from the inside of the box to the outside atmosphere was through a small rectangular opening. The length of the gap could be fixed and the depth altered by adding or removing loose flanges. Fig 2. shows this clearly.

The initial depth of gap was 0.1 inch, and it was found that before flame passed at a temperature sufficiently high to ignite an explosive mixture the gap had to be $\frac{3}{4}$ " inch long.

Experiments such as above were carried out with gaps of 0.08 inch, 0.06 inch, 0.042 inch etc., until 0.030 inch was reached when it was found that this depth of gap could be left all round the box and external ignition could not be produced by the flame from an internal explosion.

Tests similar to those described above were carried out on the larger box special loose flanges being prepared for the purpose.

The gate-end box already described was now adapted so that experiments similar to those above might be made on it. By so doing a better comparison was obtained and the effect of breadth of flange and volume better understood.

The results from this series of experiments are shown numbered and tabulated at the end of the thesis, while in table III selected results are given to demonstrate the manner in which the critical length varies with depth of gap and breadth of flange.

CONCLUSIONS AND DEDUCTIONS:-

This series of experiments was most interesting especially when nearing what might be termed the "critical length of opening" i.e. the length of opening which, at any given depth

TABLE III-

CRITICAL LENGTHS OF OPENING FOR FIXED DEPTHS OF GAP.

No	COAL GAS IN CUBIC FEET EXHAUSTION CHAMBER	IGNITING CURRENT IN AMPERES	BREADTH OF FLANGE	GAP		IGNITION		REMARKS
				LENGTH	DEPTH	INTERNAL	EXTERNAL	
25	3.108	29	1	.375	.100	Yes	No	EXTERNAL MIXTURE EXPLOSIVE.
38	3.064	12	"	.875	.080	Yes	No	Do
47	3.028	"	"	1.750	.058	Yes	No	Do
67	3.270	"	"	6.50	.042	Yes	No	Do
72	3.241	"	"	9.00	.038	Yes	No	Do
87	3.110	"	"	29.00 (PERIMETER)	.035	Yes	Yes	EXTERNAL MIXTURE EXPLOSIVE.
91	3.000	"	"	"	.030	Yes	No	EXTERNAL MIXTURE EXPLOSIVE.
101	3.010	15	1.375	1.000	.100	Yes	No	Do
109	2.700	"	"	2.875	.070	Yes	No	Do
127	2.850	"	"	4.500	.059	Yes	No	Do
148	2.776	"	"	70.00 (PERIMETER)	.040	Yes	Yes	EXTERNAL MIXTURE EXPLOSIVE.
154	2.700	"	"	"	.036	Yes	No	EXTERNAL MIXTURE EXPLOSIVE.
156	2.600	"	"	"	.030	Yes	No	Do
204	2.000	"	1.750	4.500	.046	Yes	No	Do
166	2.600	"	"	6.250	.080	Yes	No	Do
181	2.400	"	"	12.125	.058	Yes	No	Do
193	2.000	"	"	15.625	.048	Yes	No	Do
168	2.600	"	"	66.000 (PERIMETER)	.038	Yes	No	Do

EXPLANATION OF TERMS USED.

- BREADTH OF FLANGE :- THE DISTANCE THE FLANGE EXTENDS OUTWARDS FROM THE INSIDE OF THE BOX
- DEPTH OF GAP :- THE DISTANCE BETWEEN THE FLANGE OF BOX AND FLANGE OF COVER AT THE OPENING LEFT FOR ESCAPE OF HOT GASES
- LENGTH OF GAP :- THE LENGTH OF OPENING [MEASURED ALONG THE FLANGE] LEFT FOR ESCAPE OF HOT GASES

of gap, just prevents passage of igniting flame. A small alteration of length of opening at this point decides whether external ignition will occur or not. We pass from a stage where there is quite definitely internal ignition only, to a stage where there is definite external ignition occurring apparently simultaneously with internal ignition, so short is the interval between them. Between these stages we have what might be termed a "retarded ignition" stage in which the "lag on ignition" effect is often observed. In this stage two distinct reports are heard, first the internal explosion followed by the external explosion. The area of opening has been slightly increased but cooling is still being carried to a considerable degree but not sufficient to lower the temperature of the gases below the ignition temperature of the surrounding mixture. In other words the "lag" at the degree of cooling is not of sufficient duration to prevent external ignition.

It is inconceivable that, at the short lengths of opening cooling is produced solely by contact with the flanges since only a small area is exposed to the flames and the time of contact is of short duration. The rapid expansion of the hot gases from the explosion through a small opening, will result in considerable cooling, and at these small lengths of opening must be regarded as the predominant factor in preventing ignition of the external mixture. At the critical lengths cooling is carried just sufficiently far to prevent ignition of the outside mixture, consequently any slight alteration of length will affect the degree of cooling. An increase in length increases the area through which expansion takes place. It follows therefore that expansion to atmospheric pressure will take place at a relatively less rate since the velocity of the issuing gases will be reduced as the area of opening is increased. The slower rate of expansion will maintain the temperature of the gases for a relatively longer period, so that they will issue at temperatures at which the "lag"

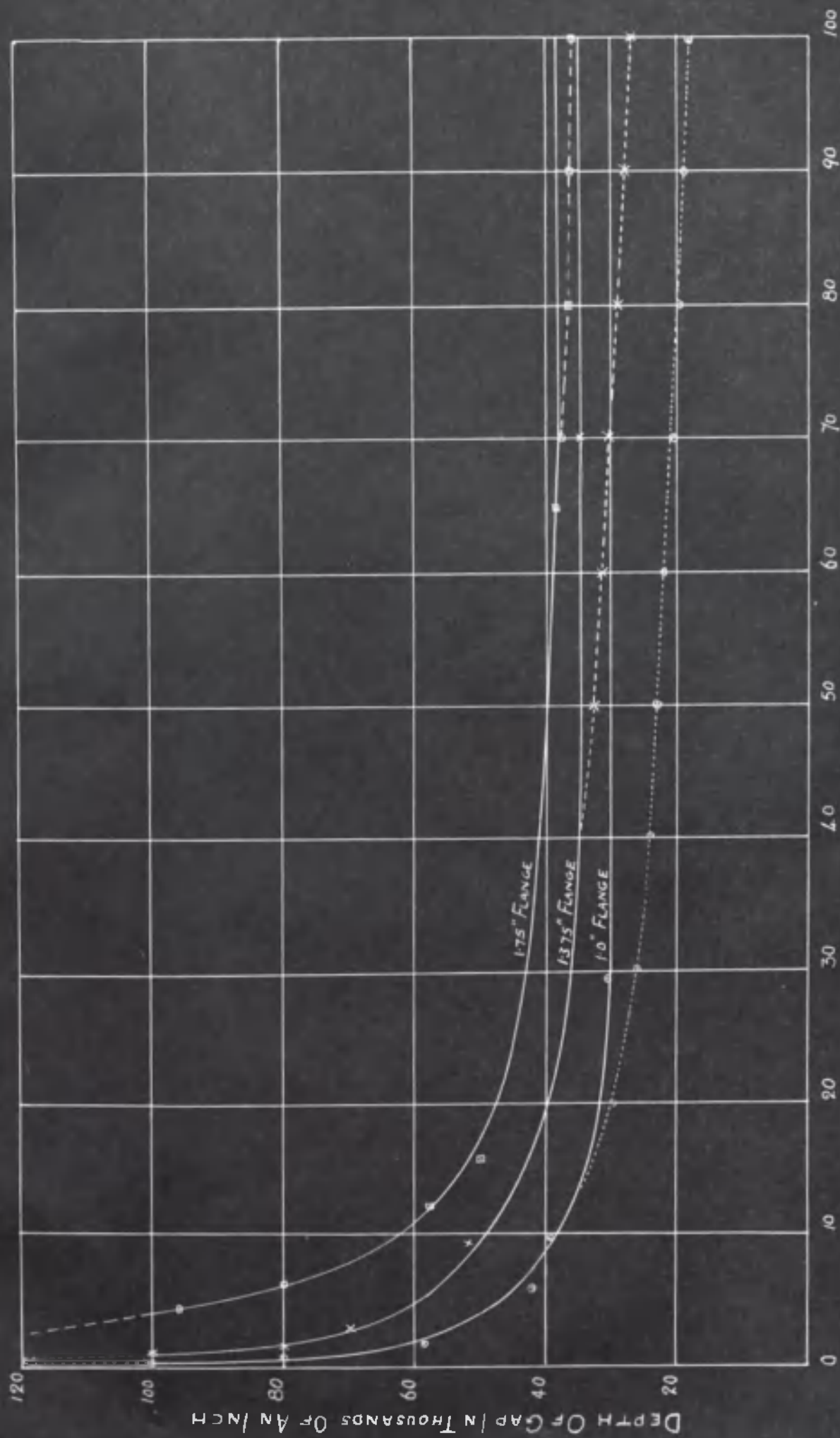


FIG. 6 - CRITICAL LENGTHS *i.e.* MAXIMUM LENGTH OF GAP WHICH, AT ANY GIVEN DEPTH OF GAP, PREVENTS PASSAGE OF FLAME CAPABLE OF IGNITING EXTERNAL EXPLOSIVE MIXTURES

is negligible and ignition of the external mixture will result. In the "retarded ignition" stage the temperature is maintained just sufficiently long to allow of ignition of the external mixture at the particular temperature, and the distinct pause between the two explosions indicates the effect of "lag on ignition."

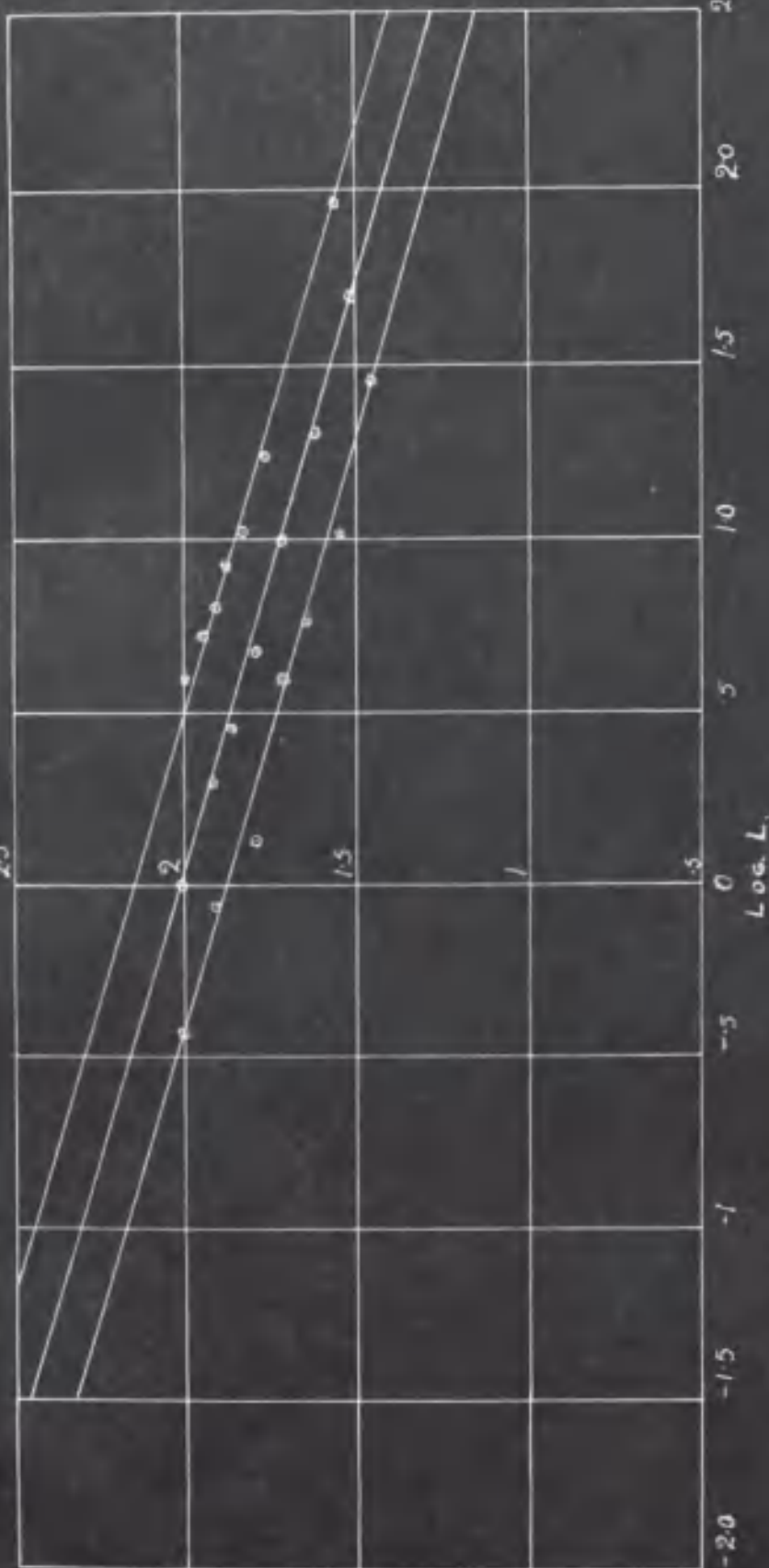
As the gap is reduced in depth and increased in length, a greater area of flange is exposed, so that at the smaller depths of gap cooling by contact with flanges will predominate. When we reach a depth of gap which can be safely left all round the casing, cooling of the gases from the internal explosion will be chiefly accomplished by the flanges and cooling by expansion will be negligible.

CURVES:-

From the tabulated results three curves were plotted with abscissae "Critical Length of Opening", and ordinates "Depth of Gap". (Fig. 6.) The curves are hyperbolae with the XY ordinate as asymptotes. From log. curves equations were obtained for each curve and by extrapolation the curves were extended beyond the points obtained in the experiments.

If we regard the perimeter of each box as a critical length, and, using the equations which satisfy the curves, calculate the depths of gaps which should prevent passage of flame at the particular perimeter, we find in each case that the calculated depth is much less than that found by experiment. The reason for this difference may be explained as follows. When a gap is left all round a box we have what might be termed an unrestricted passage to the outside. Under these conditions, when ignition occurs, the rapid expansion of gas at the source of ignition will expel unburned gas near the opening. This expulsion of unburned gas will occur at all lengths of opening, and when the length of opening is the perimeter of the box the loss must be considerable, since gas will escape in all directions. The heat generated depends on the amount of combustible gas present, so it naturally follows that under the conditions described the heat given off from the

2.5



Log D.

Log L.

BREADTH OF FLANGE 1"

DEPTH OF GAP OR GAP D.	LOG. L.	LOG. D.
100	2.00	-1.260
80	1.903	-1.055
63	1.793	-1.367
50	1.699	-1.532
42	1.632	-1.702
38	1.594	-1.822
30	1.477	-1.652

$$\log y = a \log x + \log b$$

$$2.14 = -1.26 + \log b$$

$$1.34 = 2.14 + \log b$$

$$1.34 = -1.26$$

$$\therefore a = -2.9$$

$$\log b = 1.47$$

$$\therefore \log y = -2.9 \log x + 1.47$$

$$y = \frac{2.9}{x^{2.9}}$$

$$\therefore y \cdot x^{2.9} = 7.4$$

$$\text{or } DL^{2.9} = 7.4$$

BREADTH OF FLANGE 1.375

DEPTH OF GAP OR GAP D.	LOG. L.	LOG. D.
100	2.00	-0.00
80	1.903	-1.310
70	1.875	-1.574
60	1.772	-1.690
50	1.699	-1.800
40	1.600	-1.910
36	1.535	-1.650

$$\log y = a \log x + \log b$$

$$2.3 = -1.26 + \log b$$

$$1.3 = 2.3 + \log b$$

$$1.0 = -1.0$$

$$\therefore a = -2.9$$

$$\log b = 2.01$$

$$\therefore \log y = -2.9 \log x + 2.01$$

$$y = \frac{10.23}{x^{2.9}}$$

$$\therefore y \cdot x^{2.9} = 10.23$$

$$DL^{2.9} = 10.23$$

BREADTH OF FLANGE 1.75

DEPTH OF GAP OR GAP D.	LOG. L.	LOG. D.
100	2.00	-1.021
90	1.955	-1.202
80	1.903	-1.379
70	1.851	-1.550
58	1.763	-1.742
48	1.682	-1.975
38	1.578	-1.772

$$\log y = a \log x + \log b$$

$$2.35 = -1.26 + \log b$$

$$1.40 = 2.35 + \log b$$

$$-0.95 = -3.26$$

$$\therefore a = -2.9$$

$$\log b = 2.125$$

$$\therefore \log y = -2.9 \log x + 2.125$$

$$y = \frac{133.4}{x^{2.9}}$$

$$\therefore y \cdot x^{2.9} = 133.4$$

$$DL^{2.9} = 133.4$$

FIG 6A.

burning mixture will be very much less than the potential heat of the original mixture, and the amount of heat to be dissipated will be considerably less than when the casing is completely closed or only partly open.

The large surface provided by the flanges will readily cool the escaping gases and it will be possible to increase the depth of gap up to a point beyond the calculated depth, and still prevent passage of flame capable of causing ignition of the external mixture. Increased depth of gap will increase the area of opening and tend to increase the loss of gas at ignition, but a "depth" will be reached when only the gas in contact with the flanges will be greatly cooled, and the temperature of the gas in the centre of the passage will not be appreciably lowered. In other words the flanges will be so far separated that self propagation of flame in the gas mixture between them will be possible. It follows therefore that when this point is reached external ignition will result. The 'depth' found by experiment therefore probably represents the maximum 'depth' to prevent self-propagation of flame between the flanges, and for the reasons stated above will be greater than the calculated value, when the perimeter is regarded as the critical length.

The maximum 'perimeter gap' (i.e. maximum gap that can be safely left all round the casing) for each box is indicated by a straight line parallel to the X-axis and intersecting the curve. It does not follow that these are the safe gaps for these flanges for all volumes and perimeters; e.g. taking the smallest vessel - volume .123 cub. ft. breadth of flange 1 inch perimeter 29 inches, safe perimeter gap .030 inch. - we cannot say, that this depth of gap is safe for any casing fitted with 1 inch flanges. Casings of larger volume than .123 cub. ft. will invariably have openings with perimeters, greater than 29 inches, since they will house larger gear. Greater volume means greater potential heat for same strength of gas mixture so more heat will have to be dissipated, and if breadth of flange is to remain at 1 inch the safe perimeter gap may have to be less than .030 inch. The curves

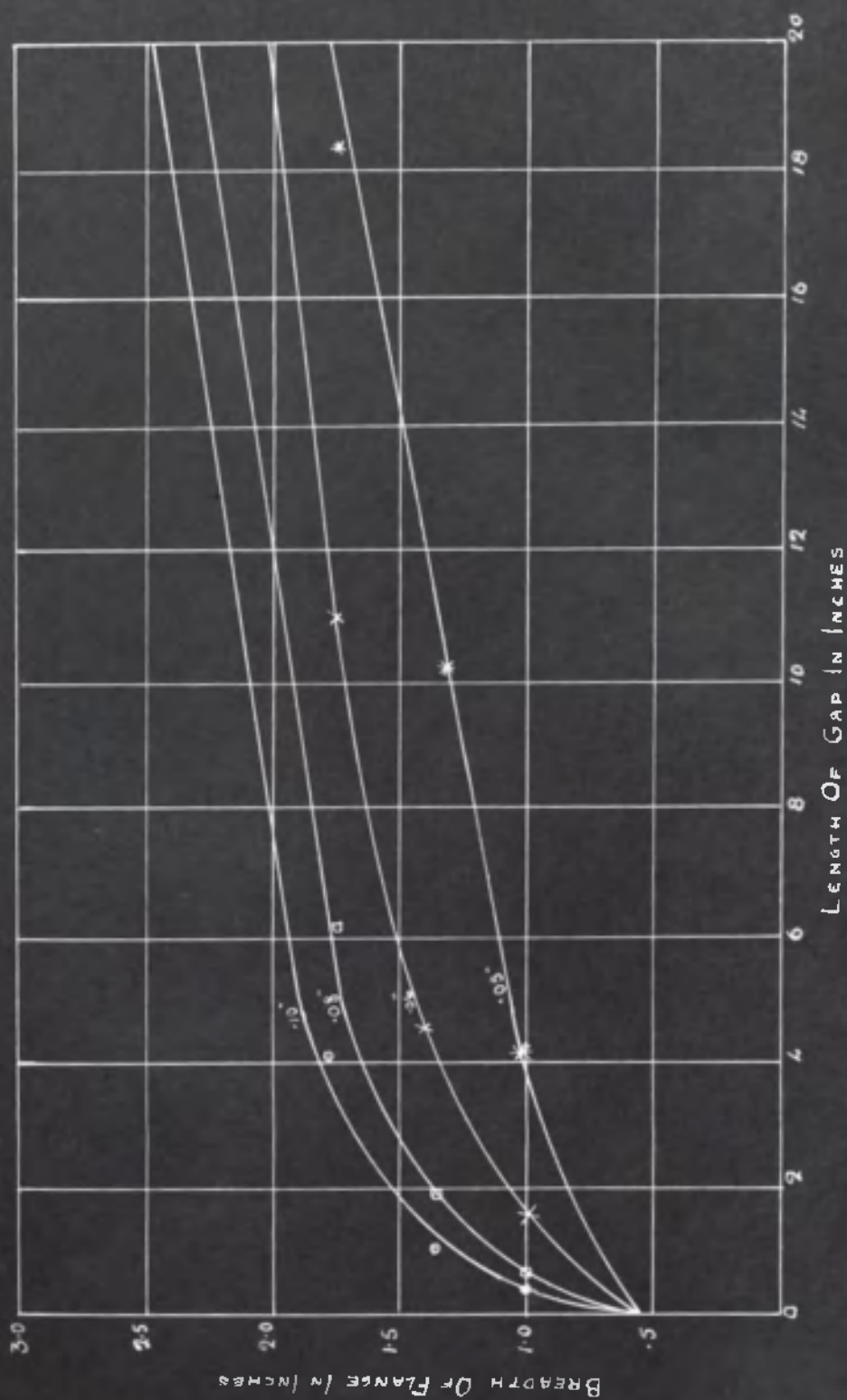


FIG. 7. BREADTH OF FLANGE REQUIRED FOR GIVEN LENGTH AND DEPTH OF GAP.
[EFFECT OF VOLUME NOT CONSIDERED]

show that increase of perimeter (where perimeter is regarded as a critical length) necessitate a decrease in depth of gap for same breadth of flange and volume, consequently if both volume and perimeter increase while the breadth of flange remains the same, a further reduction in depth of gap will be necessary.

The alternative to reduction of depth of gap is an increase in breadth of flanges, and is indeed the more satisfactory method of retaining flame-proofness. The increase in size and weight of casings will, in general, necessitate an increase in breadth of flanges at joints for reasons other than preventing passage of flame, e.g. increase in size of holes for studs, or bolts, to retain covers in position etc. However for casings to be used in mines the flame-proof factor is the most important. It would be interesting and useful if the relationship between volume and breadth of flange could be shown by a curve or expressed by an equation.

In fig. 7 a series of curves is shown, abscissae are "Critical Lengths of Opening," and ordinates are "Breadth of Flange." Readings are taken for same depths of gap from the curves in fig. 6. The curves indicate that breadth of flange has a great effect in preventing passage of flame. For example, if we take a gap of 0.1 inch depth, we find that with flanges 1 inch broad the critical length is 0.375 inch, while with flanges 1.75 inch broad the critical length for the same depth of gap is 4.0 inches an increase of 3.625 inches in critical length for an increase in breadth of flange of 0.75 inches. In this set of curves the effect of volume of casing has not been considered.

Fig. 8 shows another series of curves in which ordinates and abscissae are "Volume of Casing" and Critical Length respectively. Readings for equal depths of gap are again taken from fig. 6.

Although no account is taken of breadth of flange the influence of this factor is clearly seen, since the box with the broadest flange has the greatest length of gap for a given depth of gap.

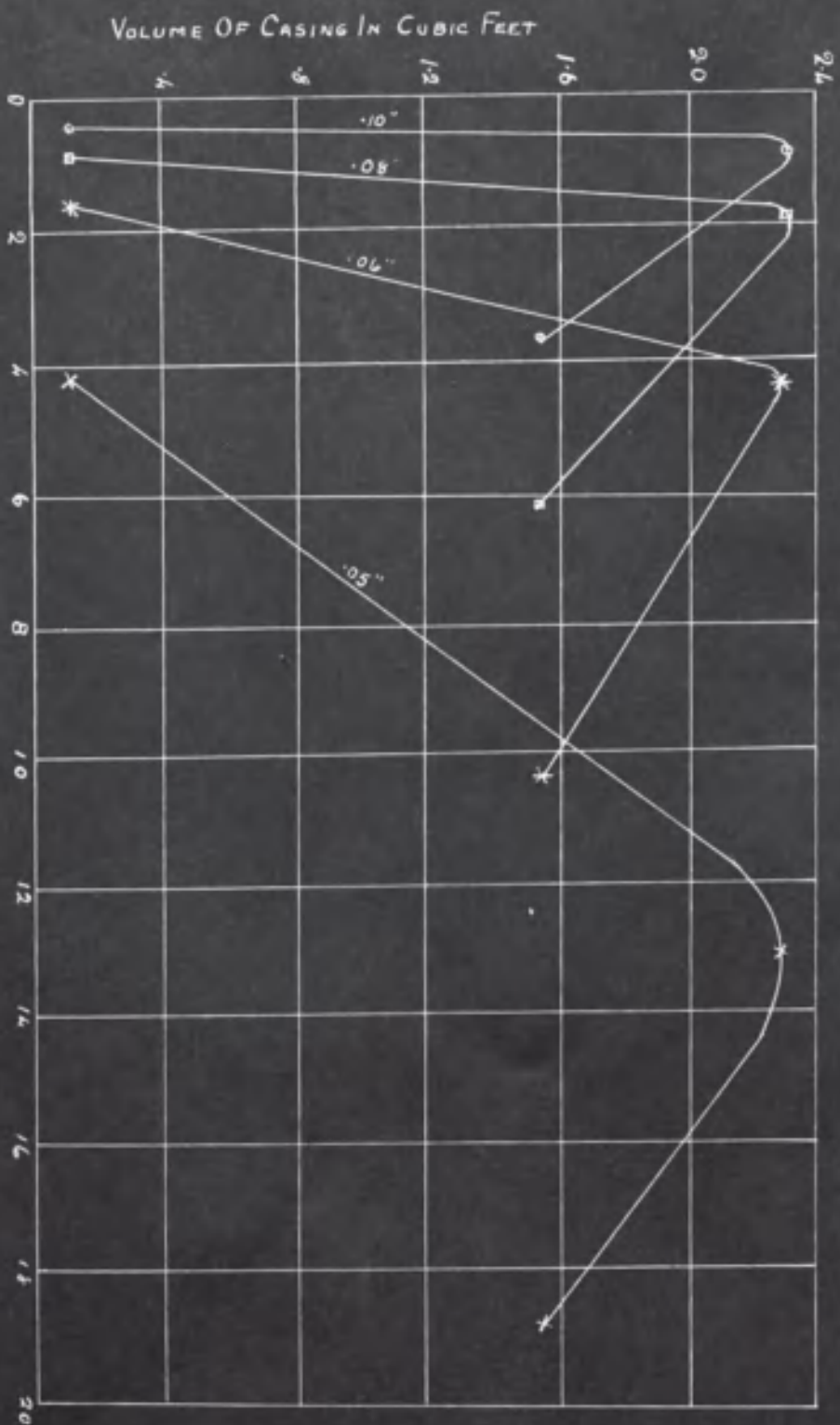


FIG 8 :- LENGTH OF GAP FOR GIVEN VOLUME OF CASING AND DEPTH OF GAP
[EFFECT OF BREADTH OF FLANGE NOT CONSIDERED]

We cannot draw any important conclusions from these two sets of curves, since in each, no account has been taken of factors having an important influence on prevention of passage of flame.

If we consider the volume left for passage of hot gases etc., (i.e. the product of length of gap, depth of gap, and breadth of flange) we have here a figure which includes the main determining factors in preventing passage of flame. If this were plotted against "volume of casing" a satisfactory relationship would not be shown, since the curves obtained would be somewhat similar to those shown in fig. 8. This of course is due to the fact that the largest casing used in the experiments was not provided with the broadest flange consequently its "volume of passage" was not the greatest for a given length of gap.

If, however we express the "volume of passage" as a ratio of "volume of casing", and plot this against volume of casing we obtain curves which will show in a satisfactory manner the relationship existing between these two factors for any given length of gap.

As the "volume of passage" is very small compared with the "volume of the casing", the figure representing the ratio will have a very small value.

We can overcome this difficulty by expressing the "volume of passage" in cubic inches and the "volume of casing" in cubic feet the result being the "volume of passage" in cubic inches, required for each cubic foot of "volume of casing," and any particular length of gap.

Calculations on this basis have been made and a table compiled (table IV), from which the curves shown in fig. 9 were plotted, showing relationship for lengths of gap between 1 inch and 100 inches.

In the calculations "lengths of gap" are regarded as "critical lengths", it follows therefore, that the "Volume of passage" given for any particular "length of gap", is that which just prevents passage of flame capable of igniting an external

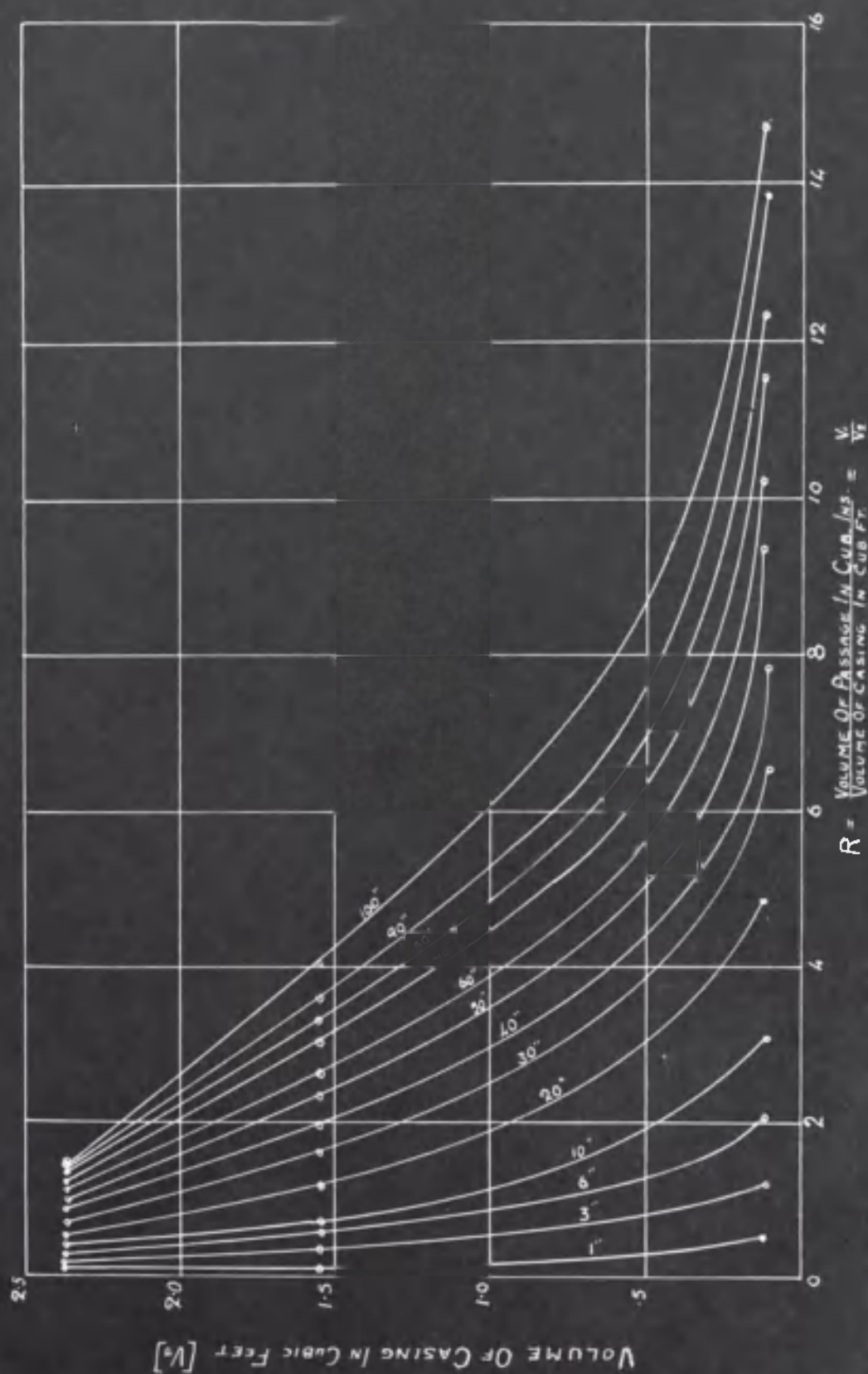


FIG. 9.— MAXIMUM VALUES OF 'R' FOR GIVEN VOLUME OF CASING AND LENGTHS OF GAP BETWEEN 1" INCH AND 100 INCHES



FIG. 10 :- BREADTH OF FLANGE REQUIRED FOR GIVEN VOLUME OF PASSAGE AT LENGTHS OF GAP BETWEEN 1 INCH AND 100 INCHES.

TABLE IV

CALCULATED VALUES OF 'V' AND R FOR VALUES OF 'L' BETWEEN 1" AND 100"

LENGTH OF GAP (L) IN INCHES	DEPTH OF GAP (D) IN INCHES	AREA OF GAP (A) IN SQ. INCHES	BREADTH OF FLANGE (F) IN INCHES	VOLUME OF PASSAGE (V) (A x F) INS ³	VOLUME OF CASING (V ₂) IN CUBIC FEET	$R = \frac{\text{VOLUME OF PASSAGE}}{\text{VOLUME OF CASING}} = \frac{V}{V_2}$
1	.074	.074	1.000	.074	.123	.577
"	.103	.103	1.375	.143	2.352	.060
"	.133	.135	1.750	.221	1.550	.142
3	.251	.153	1.000	.153	.123	1.243
"	.075	.225	1.375	.309	2.352	.127
"	.097	.291	1.750	.509	1.550	.328
6	.043	.258	1.000	.258	.123	2.097
"	.061	.366	1.375	.503	2.352	.209
"	.079	.474	1.750	.829	1.550	.534
10	.037	.37	1.000	.370	.123	3.008
"	.053	.53	1.375	.730	2.352	.314
"	.068	.68	1.750	1.190	1.550	.767
20	.030	.60	1.000	.60	.123	4.878
"	.043	.86	1.375	1.18	2.352	.501
"	.056	1.12	1.750	1.96	1.550	1.264
30	.027	.81	1.000	.810	.123	6.585
"	.038	1.14	1.375	1.560	2.352	.662
"	.050	1.50	1.750	2.600	1.550	1.677
40	.024	.96	1.000	.960	.123	7.804
"	.035	1.40	1.375	1.920	2.352	.802
"	.045	1.80	1.750	3.100	1.550	2.000
50	.023	1.15	1.000	1.150	.123	9.352
"	.033	1.65	1.375	2.270	2.352	.965
"	.043	2.15	1.750	3.760	1.550	2.425
60	.021	1.26	1.000	1.260	.123	10.245
"	.031	1.86	1.375	2.550	2.352	1.084
"	.040	2.40	1.750	4.200	1.550	2.709
70	.020	1.40	1.000	1.400	.123	11.463
"	.030	2.10	1.375	2.900	2.352	1.232
"	.039	2.73	1.750	4.770	1.550	3.077
80	.019	1.52	1.000	1.520	.123	12.357
"	.028	2.24	1.375	3.080	2.352	1.309
"	.037	2.96	1.750	5.180	1.550	3.377
90	.019	1.71	1.000	1.71	.123	13.902
"	.028	2.52	1.375	3.46	2.352	1.471
"	.036	3.24	1.750	5.67	1.550	3.658
100	.018	1.80	1.000	1.800	.123	14.634
"	.027	2.70	1.375	3.700	2.352	1.573
"	.035	3.50	1.750	6.120	1.550	3.947

inflammable gaseous mixture. Consequently if we fix the perimeter of a casing of given volume (and regard the perimeter as a "critical length"), reference to the curves will give the "volume of passage" in cubic inches per cubic foot of "volume of casing," required at the length of gap, equivalent to the perimeter, if flame-proof conditions are to hold. Having obtained this figure, the total "volume of passage" for the casing under consideration is readily found.

In fig. 10 a family of curves is shown, abscissae being total "Volume of passage," and ordinates "Breadth of flange." The volume of passage is shown for lengths of opening between 1 inch and 100 inches.

It naturally follows that if we obtain the "volume of passage" required for any particular perimeter and volume from fig. 9. reference to fig. 10 will indicate the breadth of flange necessary under the circumstances, and knowing the perimeter, breadth of flange, and volume of passage, the permissible depth of "perimeter gap" (i.e. gap left between flanges all round opening) may be easily found.

Figs. 9 and 10, should prove of great use to designers of flame-proof casings, when some doubt exists regarding the proportioning of breadth of flange, and depth of gap, for any casing, the volume and perimeter of which have been fixed, by the electrical gear it is intended to house.

Three worked examples are given to illustrate the use of the curves:-

Casings of the following dimensions are required to house certain electrical gear:-

- | | | | | | | | | | | | | |
|-----|--------|-----|-------|-------|---------|----|--------|----|--------|----|----|--------|
| (1) | Volume | 0.2 | cubic | feet, | opening | at | cover, | 9 | inches | by | 6 | inches |
| (2) | " | 0.6 | " | , | " | " | " | 12 | " | " | 8 | " |
| (3) | " | 1.4 | " | , | " | " | " | 18 | " | " | 12 | " |

Find for each the breadth of flange necessary and the maximum permissible depth of perimeter gap in order that the

casing may be flame proof.

Let Volume of passage in cub.ins. = V_1	Now $R = \frac{V_1}{V_2} \therefore V_1 = RV_2$
" " casing " ft. = V_2	And $V_1 = F \times L \times D \therefore D = \frac{V_1}{F \times L}$
Let $\frac{V_1}{V_2} = R$	V_2 and L are given.
Let breadth of flange in inches. = F	R and F are found from figs. 9 and 10.
" perimeter of casing in inches = L	V_1 and D are calculated.
" gap " " = D	

(1) $V_2 = 0.2 \text{ ft.}^3$, $L = 2(9 + 6) = 30 \text{ ins.}$

From fig. 9., R , for $L = 30 \text{ ins.}$ and $V_2 = 0.2 \text{ ft.}^3$ is given as $5 \text{ in.}^3/\text{ft.}^3$

$$V_1 = RV_2 = 5 \text{ in.}^3/\text{ft.}^3 \times 0.2 \text{ ft.}^3 = 1.0 \text{ in.}^3$$

From fig. 10, F , for $L = 30 \text{ ins.}$ and $V_1 = 1.0 \text{ in.}^3$ is given as 1.1 inch.

$$D = \frac{V_1}{L \times F} = \frac{1.0 \text{ in.}^3}{30 \text{ in.} \times 1.1 \text{ ins.}} = .030 \text{ ins.}$$

\therefore breadth of flange required is 1.1 inch and permissible perimeter gap is $0.030 \text{ inch. (max.)}$

(2) $V_2 = 0.6 \text{ ft.}^3$, $L = 2(12 + 8) = 40 \text{ ins.}$

From fig. 9., R , for $L = 40 \text{ in.}$ and $V_2 = 0.5 \text{ ft.}^3$ is given as $4.05 \text{ in.}^3/\text{ft.}^3$

$$V_1 = RV_2 = 4.05 \text{ in.}^3/\text{ft.}^3 \times 0.6 \text{ ft.}^3 = 2.43 \text{ in.}^3$$

From fig. 10, F , for $L = 40 \text{ ins.}$ and $V_1 = 2.43 \text{ in.}^3$ is given as 1.55 ins.

$$D = \frac{V_1}{L \times F} = \frac{2.43 \text{ in.}^3}{40 \text{ ins.} \times 1.55 \text{ ins.}} = .039 \text{ ins.}$$

\therefore breadth of flange required is 1.55 and permissible perimeter gap is $0.039 \text{ ins. (max.)}$

(3) $V_2 = 1.4 \text{ ft.}^3$, $L = 2(18 + 12) = 60 \text{ ins.}$

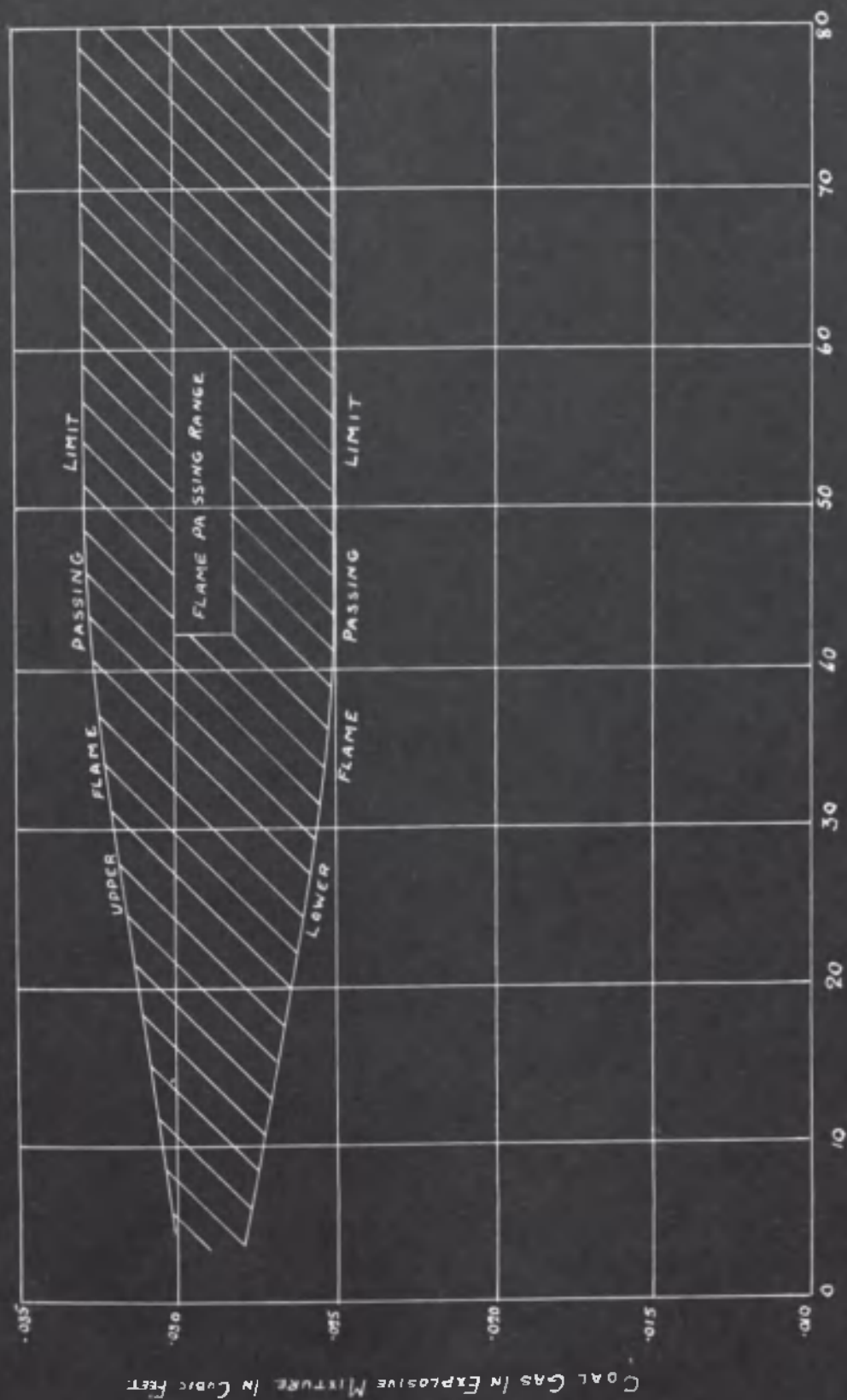
From fig. 9, R , for $L = 60 \text{ ins.}$ and $V_2 = 1.4 \text{ ft.}^3$ is given as $3.0 \text{ in.}^3/\text{ft.}^3$

$$V_1 = RV_2 = 3.0 \text{ in.}^3/\text{ft.}^3 \times 1.4 \text{ ft.}^3 = 4.2 \text{ in.}^3$$

From fig. 10, F , for $L = 60 \text{ ins.}$ and $V_1 = 4.2 \text{ in.}^3$ is given as 1.75 ins.

$$D = \frac{V_1}{L \times F} = \frac{4.2 \text{ in.}^3}{60 \times 1.75} = .040 \text{ ins.}$$

\therefore breadth of flange required is 1.75 inches and permissible perimeter gap is $0.040 \text{ ins. (max.)}$



IGNITING CURRENT IN AMPERES

FIG. 11. - VARIATION OF FLAME PASSING RANGE WITH VARIATION OF IGNITING CURRENT.

IGNITING CURRENT AND PASSAGE OF FLAME:-

This part of the "research" was devoted to investigating the effect of igniting currents of various strengths, on mixtures of gas near the limits of inflammability.

The small switch box (volume 0.125 cub .ft. flanges 1 inch broad) was used in the experiments. A gap 0.058 inch in depth was chosen, the critical length being 1.625 inches.

The procedure was exactly similar to that already described; an explosive mixture was introduced into the small box, the latter was now placed in the Explosion Chamber and surrounded with an explosive atmosphere, after which the switches were closed, thereby igniting the internal mixture. The results were noted.

By keeping the current constant throughout a series of experiments, in which the gas mixtures were varied from the lower to the upper explosive limits, points were obtained above and below the most explosive mixture, at which external ignition could not be produced, although the mixture ignited was within the explosive range.

On repeating the above experiments, using a higher value of igniting current other points were obtained at which flame capable of causing external ignition no longer passed.

The points so obtained might be conveniently referred to as the "flame passing limits", and the range of mixtures between the points as the "flame passing range."

The currents used in the experiments varied from 3 amperes to 78 amperes, and it was found that the "flame passing range" increased until igniting current had a value of 30 amperes. The "range" did not increase above this value of igniting current, and was found to be nearly coincident with the "explosive range".

Two sets of points, representing the "lower" and "upper" limits of the "flame passing range", were thus obtained and are shown plotted in fig. II.

A few of the results are given in the following table.

TABLE V.

The Effect of Strength of Igniting Current on Passage of Flame from Explosions of Gas and Air Mixtures.

Gas in cubic ft.		Ig. Cur. Amps.	Bdth. of Fl.	Gap.		Ignition.		Remarks
S.B.	E.C.			Depth.	Length.	Internal.	External	
.027	2.700	3.5	1"	.058	1.625 +	Yes.	No.	
.028	"	3.5	"	"	"	"	Yes.	
.027	"	14.5	"	"	"	"	Yes	
.025	"	14.5	"	"	"	"	No	
.024	"	33	"	"	"	"	No	Lower Limit.
.025	"	33	"	"	"	"	Yes	
.025	"	60	"	"	"	"	Yes	
.024	"	78	"	"	"	"	No	
.031	2.700	16	1"	.058	1.625 +	Yes	Yes	
.032	"	16	"	"	"	"	No	
.032	"	23	"	"	"	"	Yes	
.033	"	23	"	"	"	"	No	Upper Limit.
.0325	"	40	"	"	"	"	Yes	
.034	"	40	"	"	"	"	No	
.033	"	60	"	"	"	"	No	
.034	"	78	"	"	"	"	No	

CONCLUSIONS & DEDUCTIONS:-

The opening between the flanges was the minimum length of opening for passage of flame at the depth of gap used, consequently increase or reduction of the amount of coal gas in the mixture, by reducing the potential heat of the mixture would in itself be sufficient to prevent passage of flame.

The existence of the "flame passing range" is not surprising. As already pointed out any alteration of the conditions of experiment when working with mixtures near the explosive limits,

affects these limits.

Berthelot in his experiment observed that the explosibility of mixtures near the limits was increased, when ignition was accompanied by shock which produced mechanical disturbance of the mixture. It follows therefore that as the igniting current is increased the disturbance set up when the fuse is blown will increase, so the rate of propagation of flame will increase and time rate of development of pressure will be relatively greater. Weak mixtures will therefore burn with greater rapidity as the igniting current is increased, and conditions may be such that flame capable of causing ignition will be projected through the opening between the flanges. Points will be reached above and below the most explosive mixture where the mechanical disturbance is no longer capable of producing such marked effects, and these points will be very near, if not coincident with, the explosive limits of the mixtures used.

In the experiments the range was a maximum when the igniting current reached a value of 40 amperes, and increase of value of current beyond this did not widen the range. In Fig. 11 the results obtained are shown in the form of a graph.

Since the phenomenon is observable only at limit mixtures under abnormal igniting conditions and is apparent only when the length is increased beyond the critical length for the particular gap, it follows that if the gap is such that flame from the most explosive mixtures is effectively cooled in passing through the gap no fear need be entertained regarding the possibility of external ignition by fuses blowing under considerable overload.

A few experiments similar to those described above were carried out with gaps of 0.078 inch and 0.040 inch, and the results, in general, confirmed those already obtained.

COAL DUST:-

The effect produced by addition of coal dust to the gas mixtures.

The question of ignition of coal dust in addition to

gas mixtures is one that must be considered. Where designed openings are left in casings, entry of fine coal dust will inevitably occur, if the apparatus is used in a coal mine. In all electrical apparatus, its presence is undesirable, since faulty electrical contact etc. will be produced, if the quantity of dust is excessive. The dust will also lodge in the opening and might prevent, or hinder, the effective operation of protective measures.

In order to investigate the effect of the presence of coal dust in the casing and between the flanges, in addition to explosive gas mixtures, and to determine whether the conclusions arrived at using gas mixtures alone, would have to be modified to embrace the effect of added dust, a large number of experiments were carried out.

A sample of coal was obtained and ground to pass through a 200 mesh sieve.

Proximate analysis performed in duplicate gave the following average results.

Volatile Matter.....	32%	{ Volatile Hydrocarbons....	24.2%
Coke.....	68%	{ Moisture.....	7.8%
		{ Ash.....	5.5%
		{ Fixed Carbon.....	62.5%

QUANTITY OF DUST ADDED:-

It was decided to add dust in the proportion of one ounce per cubic foot of air space, a quantity much in excess of that necessary for propagation of flame in a dusty atmosphere. The reason was, of course, to make the test as severe as possible.

The box used in the experiments was the small box already described, having volume of 0.12 cub. ft., and flanges one inch broad. The gap chosen was 0.058 inch deep and 1.625+ inches long. This gap produces external ignition with gas mixtures.

In the initial experiments dust was placed between the flanges at the gap. It was found to have no appreciable effect, because when ignition of dust and gas took place inside, the casing, the dust between the flanges was blown out and was not

ignited. The placing of dust between the flanges was therefore abandoned on this account.

The most severe conditions possible under the circumstances, were imposed on the box but only in a few cases out of a large number of experiments was it possible to obtain external ignition. In two of the cases external ignition was effected through the bursting of the box by the internal ignition, direct communication with the outside mixture being thus afforded. The bolts securing the cover of the casing were occasionally sheared also.

When the cover was removed after an experiment, dense yellowish fumes, the product of partial distillation of the coal, were found. It would seem therefore that this evolution of distillation products etc., had some effect in preventing passage of flame. If however we consider the bursting of the two boxes, and the number of bolts that were sheared, we might be inclined to suggest that dust added to the gas mixture increases the pressure attained by the explosion.

We might visualise what takes place in the box when coal dust and coal gas present in explosive proportion are ignited. Here we have two combustible substances and a certain amount of air. When the fuse "blows", ignition of the gas mixture occurs and a dust cloud is raised. Part of this dust must be consumed directly, part will be volatilised, while a certain amount will be expelled, probably unburned, from the casing. The products of this partial distillation cannot be consumed since the oxygen has been used up in the explosion of gas and burning of dust, so that the box will be filled with spent gases and volatile matter - not a very suitable medium for flame propagation. The mixture that issues from the box will therefore consist of coal dust, distillation products and gases from the explosion. Condensation and cooling of the contents of the box immediately follows, and part of the surrounding mixture will rush in through the opening. The mixture in the box may now be inflammable but the temperature

TABLE VI.

PRESSURES DEVELOPED BY EXPLOSIVE GAS MIXTURES UNDER VARIOUS CONDITIONS.

CUBIC FEET OF COAL GAS	IGNITING CURRENT IN AMPERES	GAP		PRESSURES RECORDED IN []			AVERAGE PRESSURES lbs/sq"
		LENGTH	DEPTH				
.015	18	.375"	.10"	5	5	5	5
.020	"	"	"	20	22	22	21.5
.025	"	"	"	39	39	38	39
.028	"	"	"	41.5	42	41	41.5
.030	"	"	"	40	44	43	42
.032	"	"	"	41	40	40	40
.035	"	"	"	32	32	33	32.3
.040	"	"	"	23	20	22	22
.042	"	"	"	11	12	11	11.3
.015	"	.475	.060	5	5	5	5
.020	"	"	"	18	23	22	21
.025	"	"	"	34	34	33	34
.027	"	"	"	34	35	38	36
.030	"	"	"	36	37	40	38
.035	"	"	"	32	33	33	32.6
.040	"	"	"	24	18	25	22.3
.045	"	"	"	5	5	5	5
.020	"	1.625	.060	21	20	20	20
.025	"	"	"	33	32	31	32
.030	"	"	"	36	35	35	35.3
.035	"	"	"	29	32	30	30.3
.040	"	"	"	21	19	20	20
.020	"	6.0	.040	18	18	18	18
.025	"	"	"	26	30	28	28
.030	"	"	"	34	33	32	33
.035	"	"	"	24	32	27	28
.040	"	"	"	18	18	18	18
.020	"	9.0	.036	3	3	4	3.3
.025	"	"	"	14	15	18	16
.030	"	"	"	21	21	18	20
.035	"	"	"	12	14	15	14
.040	"	"	"	3	3	4	3.3
.030	"	29" [DIAMETER]	.030"	6	6	5	6
.030	"	1.625"	.010"	39			39
.030	"	"	.030"	37			35
.030	"	"	.040"	34			34
.030	"	"	.058"	33			33
.030	"	"	.080"	29			29

has fallen below the ignition temperature, so further ignition does not occur.

It is therefore evident that the dust cloud raised has a damping effect on the flame of the gas explosion, and the partial distillation will cause a further fall in temperature, so that gases issuing from an explosion will have a temperature considerably lower than those issuing from an explosion of a gas mixture only.

The quantity of coal dust equivalent to one ounce per cubic foot of air space, amounted to 3.5 gms. This amount of dust was practically all consumed by ten internal ignitions. Of the ten ignitions, only the last one, in general produced external ignition, being practically a gas ignition.

CONCLUSIONS:-

To sum up, we might say that the results obtained indicate that the addition of coal dust does not increase the danger of passage of flame, but rather retards it, while the pressure developed by the explosion is increased.

MEASUREMENT OF PRESSURE DEVELOPED BY EXPLOSIONS:-

An important part of this investigation lies in the measurement of pressures developed by explosions of gas and air mixtures. So far, the only point considered has been the prevention of passage of flame capable of igniting the external mixture. It is apparent however that results are incomplete without some knowledge of the pressures developed at the various 'critical lengths.'

We have already mentioned that adequate pressure release and prevention of passage of flame are the aims of all protective measures. From a "flame-proof" point of view these two factors directly oppose each other, for, while a large escape area lowers pressure it increases danger of passage of flame, and a small escape area reduces possibility of passage of flame but tends to development of high pressures. Both factors must therefore be considered when we intend to leave designed gaps between flanges, since obviously, the most suitable gap will be that

TABLE VII
PRESSURES DEVELOPED BY GAS MIXTURES CONTAINING COAL DUST.

COAL GAS IN CUBIC FEET	COAL DUST [200 MESH] IN GRAMS.	IGNITING CURRENT IN AMPERES	GAP		PRESSURES RECORDED IN LBS./SQ. IN.			AVERAGE PRESSURES lbs./sq. in.
			LENGTH	DEPTH				
.030	NIL	21	6	.040	35	31	35	33
.030	"	26	"	"	32	33	32	32.3
.030	"	35	"	"	34	33	32	33
.030	"	42	"	"	33	32.5	32	32.5
.030	"	50	"	"	34	32	33	33
.030	"	60	"	"	33	34	32	33
.015	3.5	18	1.625	.055	9	8	8	8.3
.020	3.5	"	"	"	21	20	19	20
.025	3.5	"	"	"	29	32	31	31
.028	3.5	"	"	"	36	37	37	36.6
.029	3.5	"	"	"	37	38	39	38
.030	3.5	"	"	"	39	39	39	39
.031	"	"	"	"	38	38	38	38
.035	3.5	"	"	"	32	33	34	33
.040	3.5	"	"	"	21	21	21	21
.045	3.5	"	"	"	8	9	9	8.6
.015	NIL	"	"	"	7	7	7	7
.020	"	"	"	"	20	20	19	19.6
.025	"	"	"	"	30	32	33	31.6
.028	"	"	"	"	34	34	33	33.6
.029	"	"	"	"	34	35	35	34.6
.030	"	"	"	"	36	35	35	35.3
.031	"	"	"	"	36	34	35	35
.035	"	"	"	"	29	32	30	30.3
.040	"	"	"	"	21	19	20	20
.045	"	"	"	"	7	7	7	7

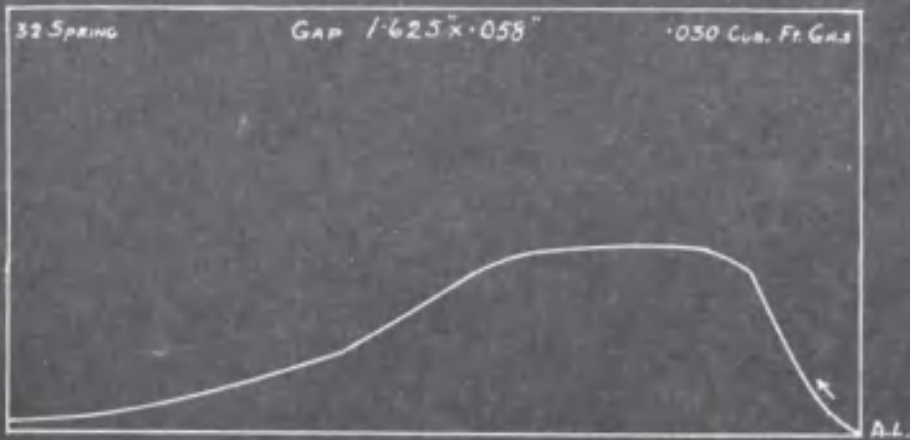


FIG. 15: REPRODUCTION OF TYPICAL TIME-PRESSURE INDICATOR CARD (READ FROM RIGHT TO LEFT)

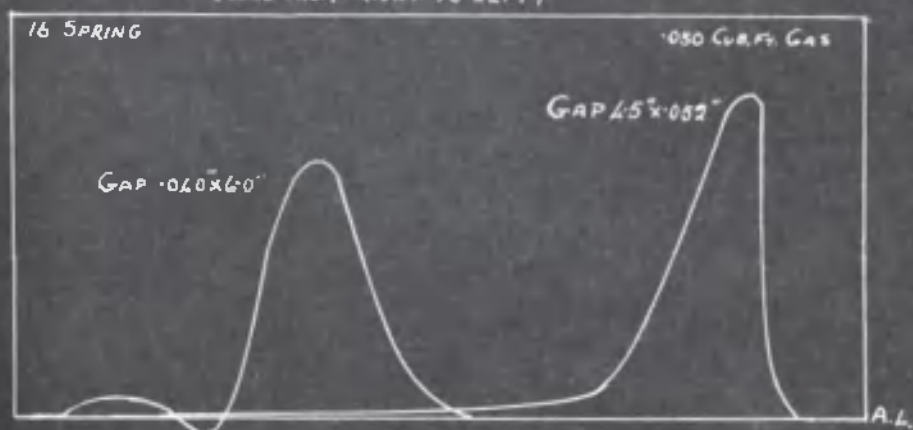


FIG. 16: REPRODUCTIONS OF TIME-PRESSURE CURVES (READ FROM RIGHT TO LEFT)

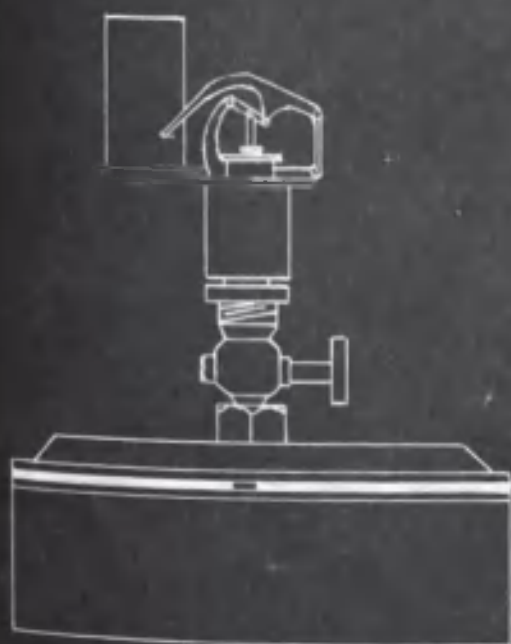


FIG. 17: SWITCH BOX WITH INDICATOR ATTACHED

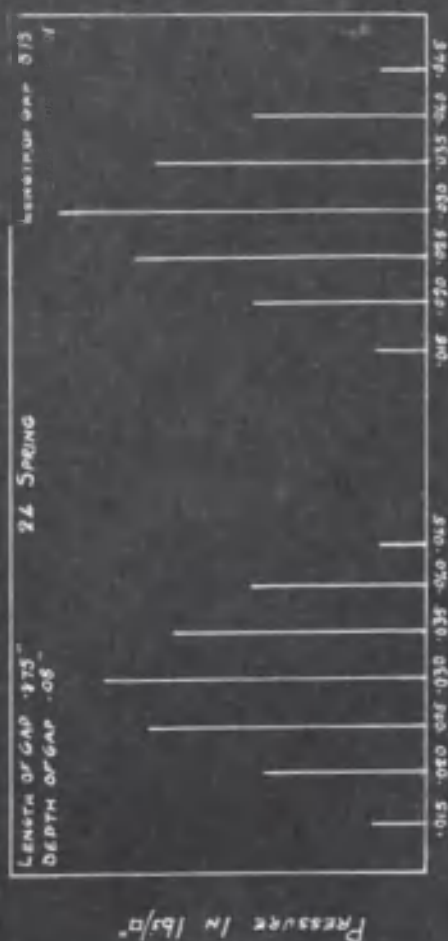


FIG. 18: REPRODUCTION OF INDICATOR CARD SHOWING METHOD OF RECORDING PRESSURES

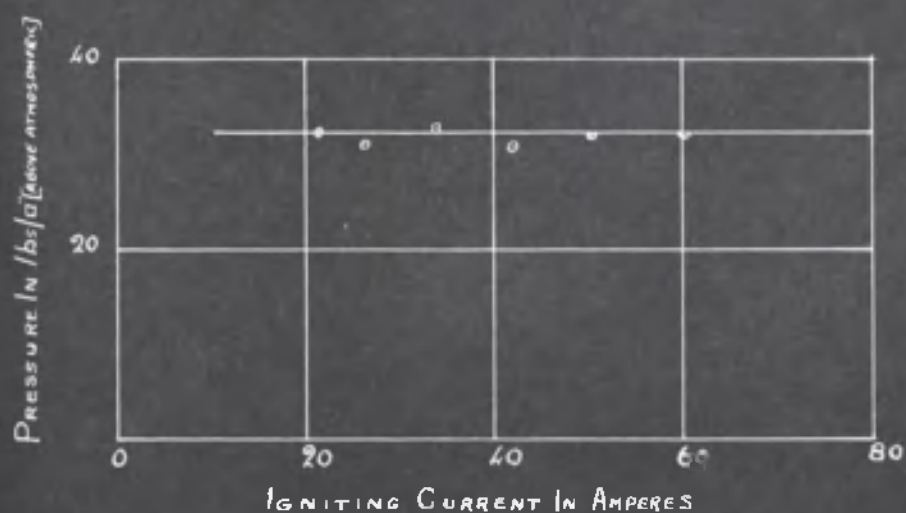


FIG. 12 :- GAS QUANTITY, DEPTH AND LENGTH OF GAP
ALL KEPT CONSTANT

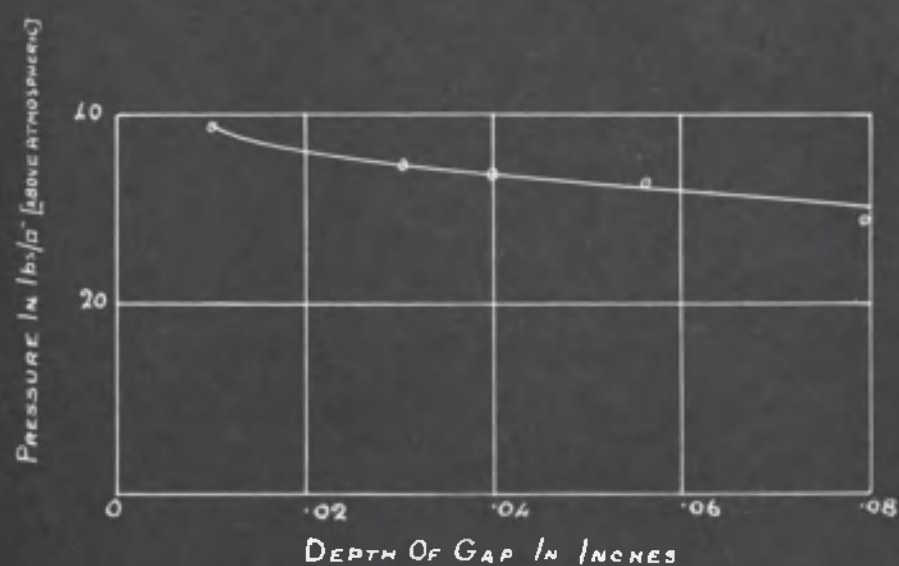


FIG. 13 :- GAS QUANTITY, LENGTH OF GAP, AND IGNITING
ALL KEPT CONSTANT

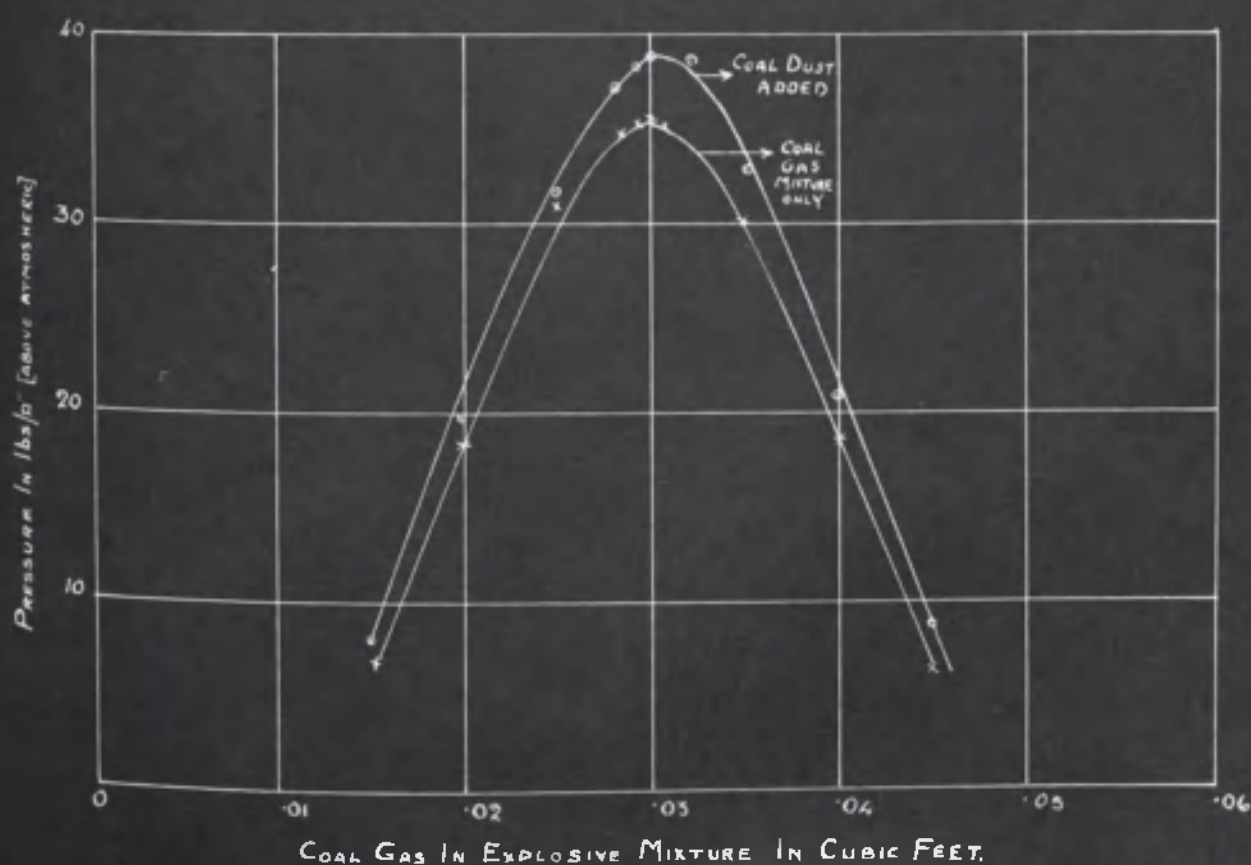


FIG. 14 :- LENGTH AND DEPTH OF GAP, AND IGNITING CURRENT, ALL KEPT
CONSTANT.

which gives the lowest possible pressure, consistent with preservation of "flame proof" properties.

Now, as the critical length increases, the "escape area" (i.e. product of "critical length and "depth of gap) increases so that we may expect lowering of the pressure developed by the explosion of the internal mixture. It has already been shown that the flange area exposed, increases, and that cooling by contact with the flanges becomes predominant, as the "critical length" increases.

Apparently then, when cooling is brought about by contact with large metallic surfaces, we may expect greatly reduced pressures, whereas when the contact area between the flanges is small, higher pressures may be expected.

It has been pointed out that when the "escape area" is small, the flange area exposed is small, and that cooling is effected mainly by expansion.

Under these circumstances, we must expect relatively high pressures.

In the measurement of the pressures three factors were considered viz. gap left between the flanges, gas quantity present, and value of igniting current. Each factor can be varied at will, and in the following experiments each has been varied in turn while the other two remained constant.

Thus, three sets of results were obtained.

- (a) Pressures developed with gas quantity and gap constant, and igniting current varied.
- (b) Pressures developed with gap and igniting current constant, and gas quantity varied.
- (c) Pressures developed with igniting current and gas quantity constant, and gap varied.

The apparatus used was similar to the small switch box already referred to, and pressures were measured by means of a steam-engine indicator (see fig. 17.) No attempt was made to obtain time-pressure records in all the experiments, but in special

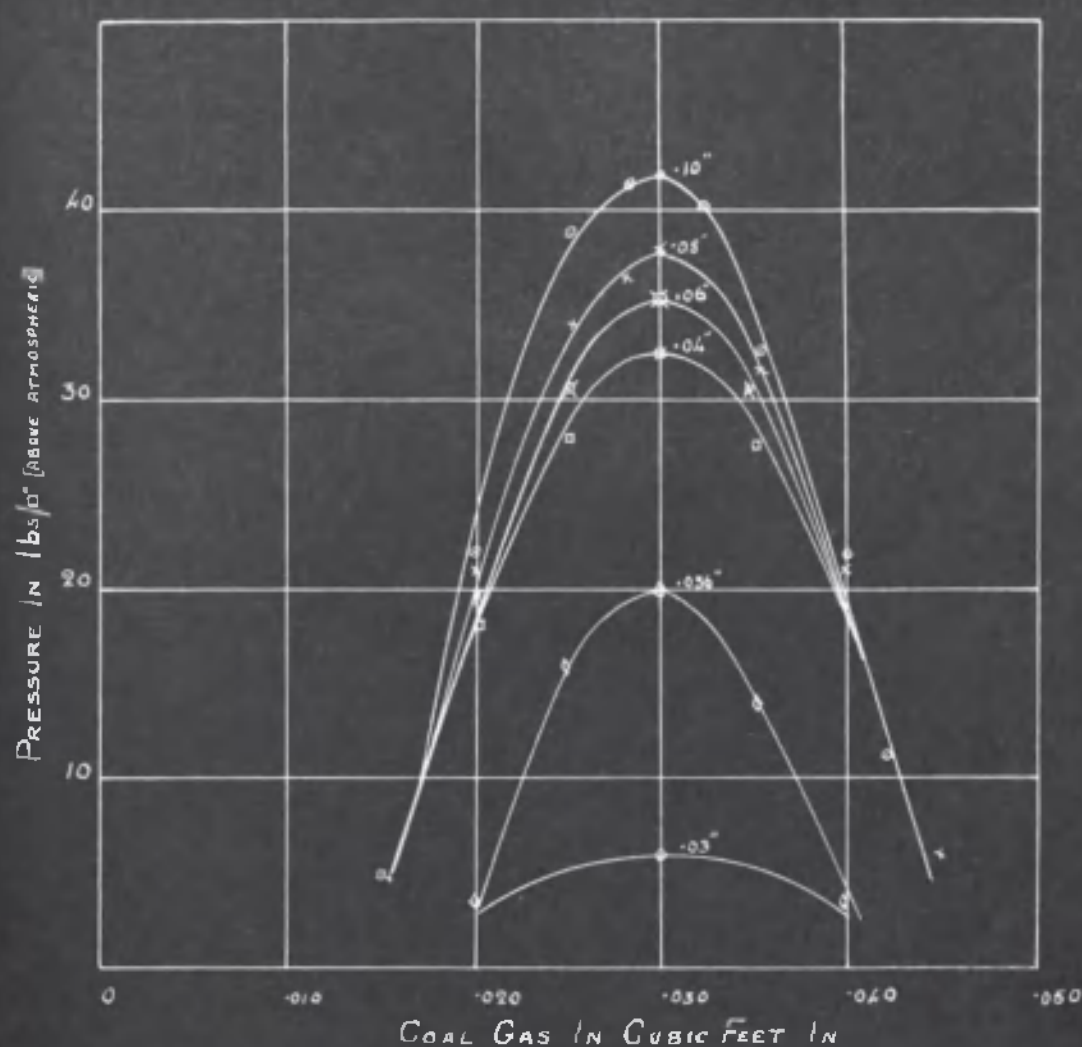


FIG 19: VARIATION OF PRESSURE WITH VARIATION OF GAS MIXTURES AT CRITICAL LENGTHS IGNITING CURRENT CONSTANT

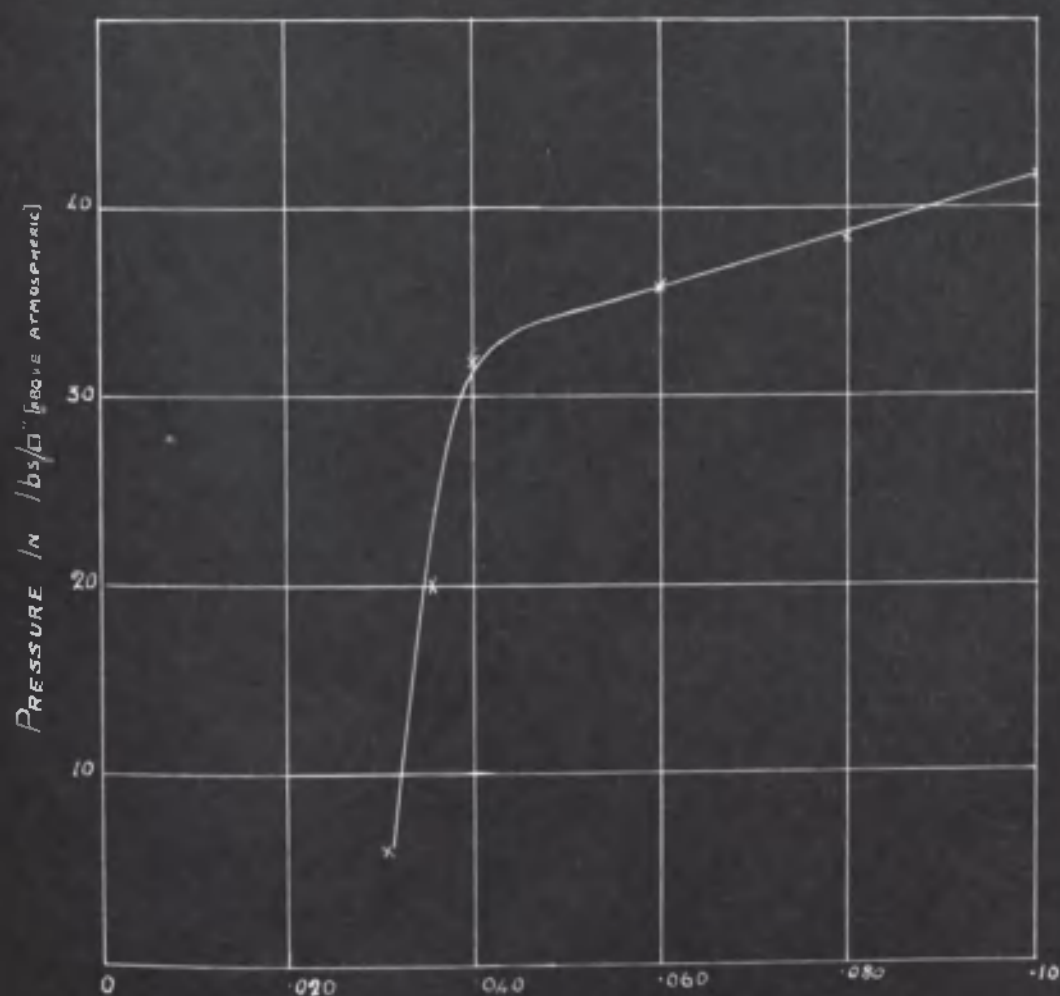


FIG 20: MAXIMUM PRESSURES DEVELOPED AT CRITICAL LENGTHS

experiments a few were obtained by releasing the 'drum' of the indicator, at the moment of ignition. Typical indicator cards are reproduced in figs. 15 and 16.

The results of these experiments are best shown by means of graphs and in addition to the tabulated results curves are shown in figs. 12, 13 and 14, in which the effects of the various factors on pressure development can be readily seen.

When igniting current is the variable factor the graph of pressure is a straight line parallel to the abscissa, indicating that the value of the igniting current has not any effect on the ultimate maximum pressure of the explosion. This, of course, is what we might expect. The only effect we might look for, would be, that the pressure rise might be more rapid with a large current than with a small igniting current.

With gap and current constant and gas quantity varied the effect on pressure is naturally, very marked. We traverse the entire explosive range of gas mixtures in this series, and observe how the pressure rises rapidly from the lower limit of explosibility to the most explosive mixture, and then, as the gas quantity is increased beyond this point the pressure falls rapidly until the upper limit is reached at which pressure is very low.

By keeping the gap length, current and gas quantity constant, and varying the depth of gap from 0.10 inch to 0.01 inch, a set of pressure records were obtained from which a curve was plotted. In this graph we observe particularly the effect of decreasing the depth of gap, or in this case continually reducing the "escape area" between the flanges. As the "escape area" is reduced we gradually approach the totally enclosed state in which no provision is made for escape of gases or pressure release, and hence pressure developed is a maximum. It follows therefore that as this state is approached the pressure must increase.

PRESSURES DEVELOPED AT CRITICAL LENGTHS:-

The pressures developed by different gas mixtures at critical lengths were also measured and curves plotted from the

result (see fig. 19.) The curves show that the pressure falls by small amounts until the gap is 0.04 inch. At depths of gap less than 0.04 inch the pressure fall is very rapid until we can leave a gap all round the perimeter. The pressure recorded under the latter conditions is only about 6 lbs per square inch.

We have referred to cooling by contact, and expansion, and described the conditions under which each predominates. There must, however, be a certain point at which each is equally effective and above and below this point some characteristic must become apparent to indicate the factor which is chiefly responsible in effecting cooling of the gases from the explosion.

We have shown that pressure is a minimum when cooling by contact predominates, consequently when a point is reached at which pressures begin to fall rapidly it must be evident that the expansion effect no longer controls cooling.

The position of this point is readily detected in fig. 20 in which ordinates are "maximum pressures at critical lengths" and abscissae are depths of gap.

The point at which change occurs is 0.04 inch depth of gap. We may reasonably assume then, that cooling by contact begins to assert itself from .04 inch gap downwards and that at gaps over 0.04 inch, expansion is the chief cause of cooling, while the effect of each is apparently equal at this particular depth of gap.

PRESSURES DEVELOPED WHEN COAL DUST IS ADDED TO THE GAS MIXTURES:-

Coal dust, in the proportion used in previous experiments, was added to the box and a set of pressure records obtained. The results are shown plotted in fig. 14, and indicate that the pressure is higher than when gas mixtures alone are used. Under favourable conditions very much higher pressures than those indicated may be recorded.

GENERAL CONCLUSIONS:-

It has been shown that strength of igniting current does not affect the ultimate pressure attained by an explosion in

a partially closed vessel.. We have also seen that the addition of coal dust increases the pressure developed. The most important results however are those showing the pressures developed at 'critical lengths'. These results help materially in deciding which method of protection to adopt. It is clearly evident that cooling by contact with large cooling surfaces is preferable to cooling by expansion, or any method in which resistance is offered to the escaping gases, as, for example, in passing through a special device. It is of importance to note that large cooling surfaces, from which the 'escape area' is small, do not completely satisfy the requirements, because pressure developed varies inversely as the 'escape area'. In employing flanges with the proper perimeter gap between them, we provide a large cooling surface and a large 'escape area', and, consequently combine effective cooling with adequate pressure release. In this respect flange protection is superior to all protective devices, because, as we have already pointed out, it is doubtful whether the two factors can be satisfactorily combined in the same appliance.

WEBBED COVERS:-

*
While carrying out a series of experiments in which a stronger type of box was used it was observed that critical lengths were slightly greater than in the other casings of same dimensions. The breadth of flange was the same in each case, and the volumes were also approximately equal, so that we might have expected similar critical lengths. The material of which each casing was made was cast iron, therefore possibility of different rates of conduction of heat cannot be entertained.

In the stronger type of box there were several internal webs. This, while strengthening the box increased the internal surface area by approximately 30 sq. inches. It follows therefore that a larger cooling surface was presented to the hot gases of the explosion so that the temperature reached would be less than in the box in which there were no webs. The gases issuing at the opening would be at a lower temperature than under the old

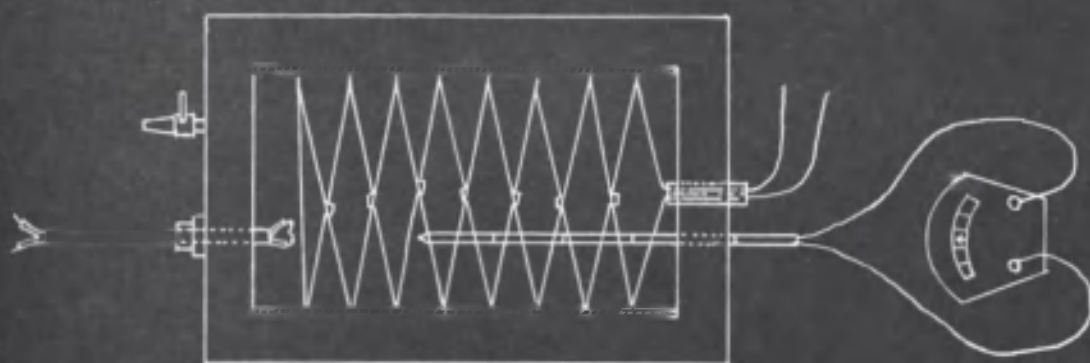


FIG 21: SMALL SWITCH-BOX WITH HEATING COIL
AND THERMO-COUPLE IN POSITION AS USED
IN HEATED CASING EXPERIMENTS

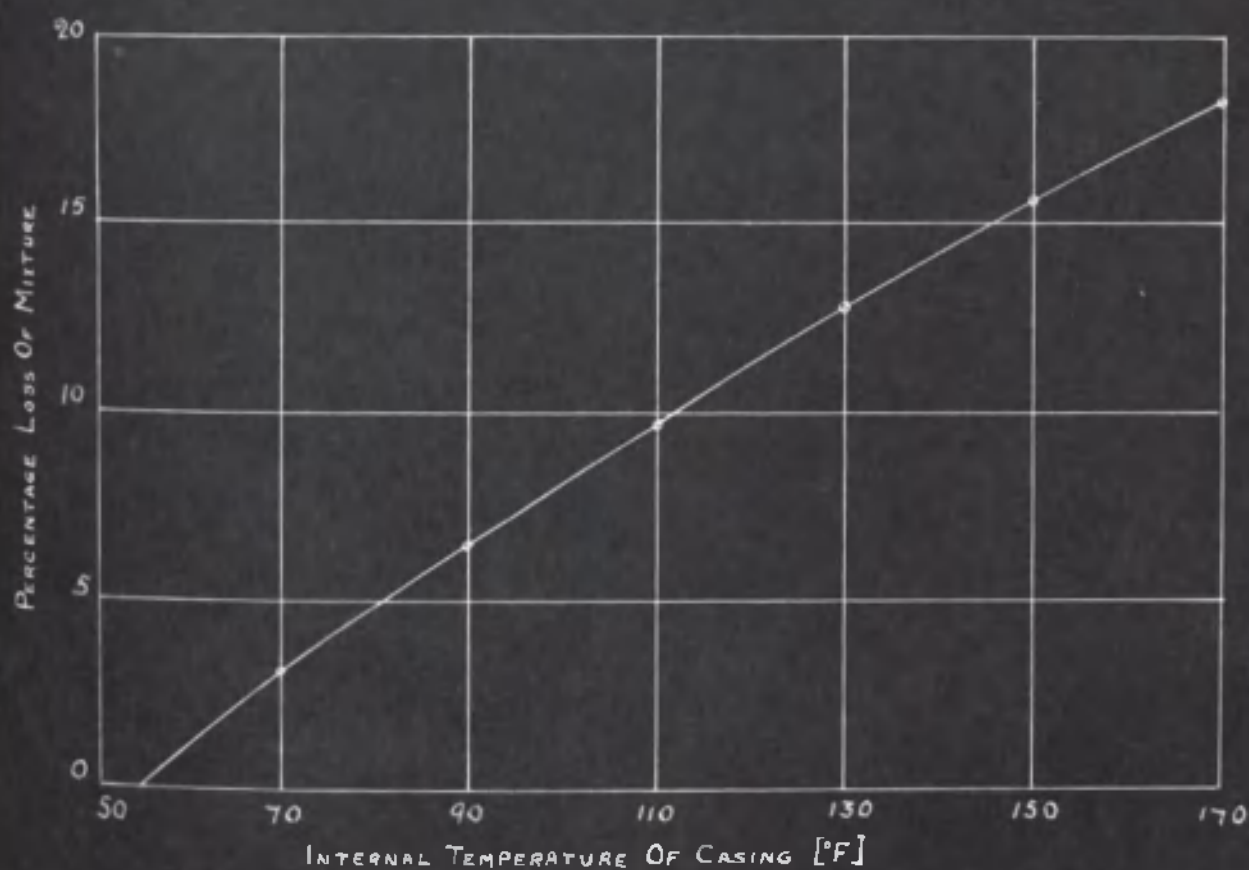


FIG 22: CURVE SHOWING REDUCTION IN "NET WEIGHT" OF MIXTURE WITH RISE IN
TEMPERATURE OF CASING
INITIAL TEMPERATURE OF CASING 55°F

conditions, therefore an increase in the 'escape area" would be necessary before flame passed to ignite the external mixture.

The critical lengths would therefore be greater in the stronger box.

The increase in length was slight but very marked. Further investigation is necessary before definite conclusions can be arrived at, but the results are certainly indicative of what might be done in the direction of cooling gases before they pass between flanges or through protective devices.

Internal, and external webs, are already applied to casings of certain machinery used in mines, the purpose of the application being to prevent excessive rise in temperature under working conditions. The idea could be readily extended to switch-gear casings etc., to increase the cooling surface presented to hot gases in event of an explosion occurring, and thereby considerably assist in maintaining the flame-proof features of the apparatus.

HEATED CASINGS:-

Up to the present the experiments described have all been performed with the casings at the atmospheric temperatures prevailing at the time of experiment. Strictly speaking therefore, the results obtained, and the conclusions arrived at, apply only to ordinary atmospheric conditions.

The rise in temperature of electrical apparatus depends on the working conditions, and may be such that temperature of 100°C and over may be reached.

Electrical defects in the apparatus, will of course give rise to abnormally high temperatures at certain parts of casings, and in the following discussion are considered apart from temperature rise due to working conditions.

The problem that confronts us, is, to determine whether apparatus, flame-proof at atmospheric temperatures, still retains its properties at higher temperatures. If the properties are not

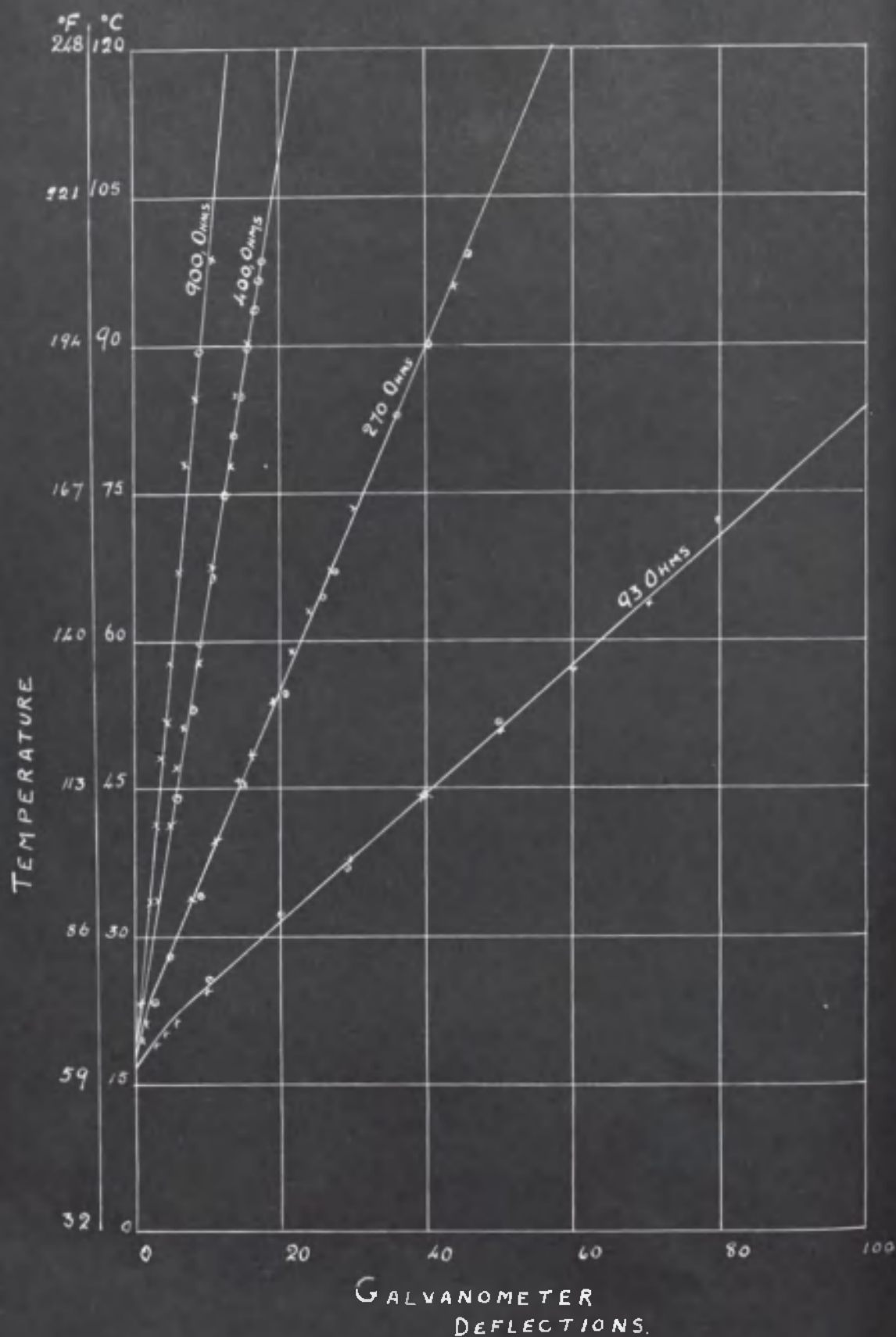


FIG. 23: THERMO-COUPLE CALIBRATION CURVES

retained, we must consider what modifications are necessary to counteract the effects of the increase in temperature.

Now, ample cooling surface is essential in all protective measures, and the lower the temperature of the surface, the more efficiently will it function.

Each degree rise in temperature, will therefore reduce the efficiency of the cooling surface, and a point will ultimately be reached at which the cooling power will be negligible, so far as the cooling of hot gases is concerned.

As the temperature of the walls of the casing increase, their cooling effect on the gases, at ignition will not be so marked, and higher temperatures may be reached by the gases, if the vessel is totally closed.

The behaviour of the gaseous contents of the casing, under an increase in temperature, must also be taken into account.

Let us consider a totally closed vessel, into which a gaseous mixture at atmospheric temperature and pressure, has been introduced. According to Charles' Law if the temperature of the casing be raised, the pressure of the gases will rise above atmospheric pressure, while the volume remains practically constant. If the contents of the vessel were explosive, higher pressures would be developed on ignition, under the latter conditions.

Now, if the vessel were only partly closed, and communication with the outside was effected, say through a small opening between flanges, totally different results would be obtained.

Free expansion is possible in this case and pressure remains constant. It follows therefore that with each degree rise in temperature a certain volume of gas, will leave the vessel. The volume of gas remaining in the vessel will be the same (approximately) at all temperatures, but the nett weight of gas will be reduced as the temperature rises. We see therefore that if the original mixture was combustible, it follows that as

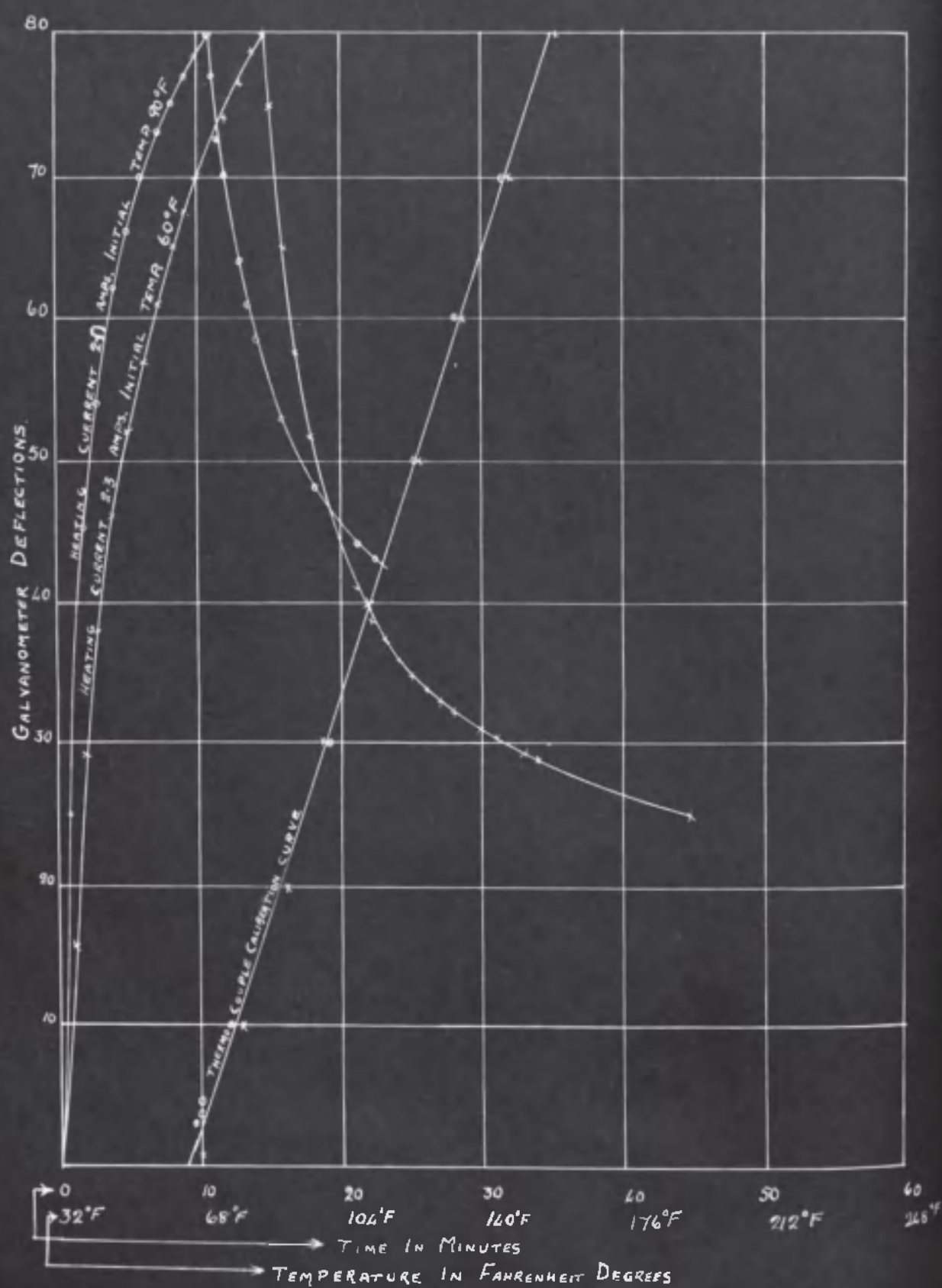


FIG. 24. HEATING AND COOLING CURVES. NOTE EFFECT OF INITIAL TEMPERATURE OF CASING ON RATE OF HEATING

temperature rises, the potential heat of the mixture remaining, decreases. (See fig. 22.)

It is thus evident that while the efficiency of the cooling surface is reduced with rise in temperature, in the event of an explosion occurring in the vessel, less heat has to be dissipated, so that it may still be capable of preserving the flame-proof properties so essential to safeworking.

In the case of cooling by expansion through a small opening or gap, we must consider the state of the mixture rather than the state of the vessel containing it.

We have seen that the potential heat of the mixture decreases as the temperature of the casing rises, therefore pressure developed by the explosion will be less, consequently the rate of expansion will be reduced.

The decrease in rate of expansion will of course be reflected in the range of cooling effected, and were it not for the fact that the potential heat of the mixture is less than normal, we might expect the temperatures of the expelled gases to be higher than when the initial temperature was that of the surrounding atmosphere.

In the actual experimental work the greatest difficulty was to devise a suitable method of heating the casing to the desired degree and maintaining it at that temperature.

Obviously the most suitable means was electrical, and a ni-chrome coil similar to those fitted in electrical fires was used. Temperature records were taken with a Copper-Constantan thermo-couple, calibration curves are shown in fig. 23.

The small box already referred to was adapted for the experiments, being suitably bored to receive the heating coil and the thermo-couple. The arrangement is shown in Fig. 21 and was found to be the most suitable under the circumstances.

In the initial experiments the usual explosive mixture was admitted to the box at atmospheric temperature, then heating operations were commenced, and continued until the required degree

of temperature was reached. During the heating stage gas was admitted to the explosion chamber till the correct explosive mixture was obtained, the condition being adjusted so that the required temperature was reached when the correct quantity of gas had been admitted.

The internal explosions produced under the above conditions, were generally very feeble, and in a number of experiments no internal ignition was obtained.

The current passing in the coil was adjusted by means of a rheostat and in determining the value to be used for heating purposes, the coil was first brought to red heat, then the resistance was increased until visible light was no longer emitted. The coil must therefore have been at a temperature of over 300°C .

At this temperature prolonged contact with the gaseous mixture, would tend to oxidation of the combustible constituents, with production of a non-explosive or feebly explosive mass.

It is therefore evident that gas cannot be admitted to the box while current is passing through the heating coil.

The casing of course never attained the temperature of the coil, and to determine its temperature at anytime during the period of heating, and after the heating current was cut off, a number of time-temperature observations were made for different values of heating current. Time-temperature curves were plotted from the observations, and by referring to them it was a simple matter to select a rate of heating and cooling so that the required temperature of casing would be attained at a certain specified time after the heating current was cut off. See fig. 24.

During the cooling stage gas was admitted to the casing and then to the explosion chamber, the operation being completed when the casing had reached the required temperature.

By adopting this procedure the possibility of oxidation of the internal mixture was eliminated, but another undesirable feature was introduced.

As the box cooled, contraction of its gaseous contents

would occur, and gas would enter from the outside to maintain equilibrium of pressure. It follows therefore that the internal mixture would gradually alter in composition as the casing cooled.

Somewhat similar results may be expected when the external mixture is admitted, before gas is passed into the small box under test.

In this case there will be a mixture containing a small quantity of coal gas, introduced by diffusion and contraction into the casing, before the real explosive mixture is admitted.

We therefore do not know the exact composition of the mixture in the casing, on account of the changes continually taking place, so that while every effort was made to have the internal and external mixtures of same composition it is doubtful if this was ever attained.

The experiments were carried out at critical lengths for gaps from .108 inch to 0.030 inch, so that any change would be readily detected, because the margin of safety is necessarily very small at critical lengths, and any slight alteration in the conditions of experiment, might produce large differences in results.

RESULTS:-

The experiment showed very clearly that the casings remained "flame-proof" until temperatures from 103°F to 110°F were reached. At temperatures above 110°F external ignition could be produced on every occasion until by continual heating the casing had reached a temperature of 155°F - 160°F, when flame capable of causing internal ignition no longer passed.

Further heating of the casing did not seem to alter the results obtained at 160°F.

CONCLUSIONS:-

It has been pointed out that the higher the initial temperature of the casing, the less will be the cooling when ignition takes place, so that the maximum temperature attained by the explosion will probably be higher than normally.

We have also observed that decrease in potential heat of mixture accompanies rise in temperature of the casing, if the latter contains an opening communicating with the outside.

Now, these two factors operate at the same time, the former increasing in power as the temperatures rises above atmospheric, while the latter becomes more apparent as temperature increases.

There is therefore every possibility that we may reach a temperature when "temperature increase of explosions" has a more powerful influence than "the reduction in potential heat effect."

When this state of affairs is reached, external ignition will occur, and will continue until the potential heat of the mixture is reduced to such an extent that the maximum temperature attained (taking into account the increase due to initial temperature) is relatively low.

Under these conditions cooling by contact or expansion, will once more assert itself and thereby prevent external ignition.

In figure 22, percentage loss of explosive mixture is plotted on a base of temperature in °F., the initial temperature being taken as 55°F, and a barometric pressure of 29.8 ins is assumed throughout.

From this graph we observe that the percentage loss of mixture is 9.75 per cent at 110°F and 16.74 per cent at 160°F and if the above hypothesis is correct these points probably represent the upper and lower limits of a flame-passing zone.

TABLE IX.

Loss of Mixture with
Rise in Temperature.

Temp. in °F.	Loss in Cub. ft.	Per Centage Loss.
70	.0035	2.84
90	.0079	6.42
110	.0120	9.75
130	.0157	12.76
150	.0192	15.60
170	.0225	18.29

CALCULATION OF LOSS.

Vol. of Casing is 0.123 cub. ft.
 e.g. Find volume at atmospheric temp. of 55°F that will occupy 0.123 cub. ft. at 110°F.

$$V = \frac{V_1 T_1}{T_2} = \frac{0.123 \times 515}{570} = 0.1111 \text{ cub. ft. at } 55^\circ\text{F and } 29.8''\text{Bar.}$$
 occupies 0.123 cub. ft. at 110°F and 29.8" Bar.
 ∴ volume (at atmos. pressure and temp.) that is lost = $0.123 - 0.1110 = 0.012 \text{ cub. ft.}$
 ∴ Per Cent. Loss = $\frac{0.012}{0.123} \times 100 = 9.75 \text{ per cent.}$

The margin existing between ignition and non-ignition of the external mixture must be very small before the effect described above can be shown. Under the conditions of experiment stated above the effect was clearly observed.

To determine whether the effect was observable when the casing contained explosive firedamp mixtures a quantity of methane was generated from Aluminium Carbide, purified and stored in a gas tank. The necessary amount of methane was drawn from the tank as required.

The results from this series of experiments closely resembled those obtained with the coal gas explosive mixtures, as will be seen from the tabulated results at the end of the thesis.

ANALYSES OF GAS MIXTURES USED IN THE EXPERIMENTS:-

Samples drawn from the casing were analysed periodically. It is of interest to compare the results of analyses of samples, drawn when the casing was at Atmospheric temperature, with those drawn after the gas had been in contact with the heated surface for a considerable time.

ANALYSES OF SAMPLES OF GAS DRAWN FROM THE CASING.

<u>No. 1.</u>		(a)	
<u>Barometric Pressure</u>	30.24"	Sample	45.8 ccs.
<u>Atmospheric Temperature.</u>	60°F.	K.O.H.	45.4 ccs....4ccsCO ₂
<u>Temperature of casing when gas was admitted.</u>	60°F.	Br.+K.O.H	45.3 ccs....1 ccs. C _m H _n .
<u>Temperature of casing when sample was withdrawn.</u>	60 F.	Pyro.	38.05 ccs...7.25ccs. O ₂
<u>Quantity of Coal Gas admitted to casing.</u>	.030cub.ft.	<u>RESULT:-</u>	0.87% CO ₂ 0.21% C _m H _n 15.80% O ₂
<u>Time between admission of gas and drawing of sample.</u>	1 minute.		

(b)			(c)		
Sample	40.00ccs.		Sample	63.40ccs.	
KOH	39.66 ccs.34ccs.CO ₂	KOH	62.90ccs.	.5ccsCO ₂
Br.+ KOH	39.56 ccs.10ccsC _m H _n	Br.+KOH	62.70ccs.	.2ccsC _m H _n
Comb.	29.77 ccs.	..9.79ccs.Cont.	Comb.	47.00ccs.	15.7ccsCont.
KOH	26.99 ccs.	..2.78ccs.CO ₂	KOH	43.00ccs.	4.00ccsCO ₂
Pyro	26.36 ccs.	.. .63ccs.O ₂ left.	Pyro.	42.00ccs.	1.0ccs. O ₂ left.

O ₂ in Sample	=	6.32 ccs.	O ₂ in Sample	=	10.02 ccs.
O ₂ after comb.	=	.63 ccs.	O ₂ after comb.	=	1.00 ccs.
. . O ₂ used in comb.	=	5.69 ccs.	. . O ₂ used in comb.	=	9.02 ccs.

H ₂	=	Total Contraction - O ₂ used in comb.	H ₂	=	Total Contraction - O ₂ used in Comb.
H ₂	=	9.79 ccs. - 5.69 ccs.	H ₂	=	15.70 ccs. - 9.02 ccs.
H ₂	=	4.10 ccs.	H ₂	=	6.68 ccs.

CH ₄	=	$\frac{2 \times \text{Cont.} - \text{CO}_2 \text{ formed}}{3} - \text{H}_2$	CH ₄	=	$\frac{2 \times \text{Cont.} - \text{CO}_2 \text{ formed}}{3} - \text{H}_2$
CH ₄	=	$\frac{2 \times 9.79 - 2.78}{3} - 4.1$	CH ₄	=	$\frac{2 \times 15.7 - 4}{3} - 6.68$
CH ₄	=	5.6 - 4.1 = 1.5 ccs.	CH ₄	=	9.13 - 6.68 = 2.44 ccs.

CO	=	CO ₂ - CH ₄	CO	=	CO ₂ - CH ₄
CO	=	2.78 - 1.5	CO	=	4 - 2.44
CO	=	1.28 ccs.	CO	=	1.55 ccs.

RESULT.

0.85% CO₂
 0.25% C_mH_n.
 10.25% H₂
 3.75% CH₄
 3.20% CO
 15.80% O₂
65.90% H₂
100.00

RESULT.

0.79% CO₂
 0.31% C_mH_n
 10.53% H₂
 3.86% CH₄
 2.44% CO
 15.80% O₂
66.27% H₂
100.00

<u>No. 2.</u>		(a)	
<u>Barometric Pressure</u>	30.54"	Sample	68.50 ccs.
<u>Atmospheric Temperature.</u>	57°F.	KOH	67.00 ccs. 1.50 ccs. CO ₂
<u>Temperature of Casing when gas was admitted.</u>	300°F.	Br. + KOH	66.80 ccs. 0.2 ccs. C _m H _n .
<u>Temperature of casing when sample was withdrawn.</u>	147°F.	Pyro.	57.00 ccs. 9.60 ccs. O ₂
<u>Quantity of Coal Gas admitted to casing.</u>	.0295 cub. ft.	<u>RESULT:-</u>	2.2% CO ₂
<u>Time between admission of gas and drawing of sample.</u>	12 mins.		0.29% C _m H _n .
			14.49% O ₂

(b)		(c)	
Sample	41.00 ccs.	Sample	40.10 ccs.
KOH	40.10 ccs. 0.90 ccs. CO ₂	KOH	39.30 ccs. ... 0.80 ccs. CO ₂
Br. + KOH	40.00 ccs. ... 0.10 ccs. C _m H _n	Br. + KOH	39.20 ccs. ... 0.10 ccs. CO ₂
Comb.	30.40 ccs. ... 9.60 ccs. Cont.	Comb.	29.80 ccs. ... 9.40 ccs. Cont.
KOH	28.00 ccs. ... 2.40 ccs. CO ₂	KOH	27.40 ccs. ... 2.40 ccs. CO ₂
Pyro.	27.60 ccs. ... 0.40 ccs. O ₂ left.	Pyro.	27.10 ccs. ... 0.30 ccs. O ₂ left.
O ₂ in Sample	= 5.98 ccs.	O ₂ in sample	= 5.8 ccs.
O ₂ after comb.	= 0.40 ccs.	O ₂ after comb.	= 0.3 ccs.
∴ O ₂ used in comb. =	5.58 ccs.	∴ O ₂ used in comb. =	5.5 ccs.
H ₂ = Total Contraction - O ₂ used in comb.		H ₂ = Total Contraction - O ₂ used in Comb.	
H ₂ = 9.60 ccs. - 5.58 ccs.		H ₂ = 9.40 ccs. - 5.5 ccs.	
H ₂ = 4.02 ccs.		H ₂ = 3.90 ccs.	
CH ₄ = $\frac{2 \times \text{Total Cont.} - \text{CO}_2 \text{ formed}}{3}$		CH ₄ = $\frac{2 \times \text{Total Cont.} - \text{CO}_2 \text{ formed}}{3}$	
CH ₄ = $\frac{2 \times 9.6 - 2.40}{3}$ = 4.02		CH ₄ = $\frac{2 \times 9.40 - 2.40}{3}$ = 3.9	
CH ₄ = 5.6 - 4.02 = 1.58 ccs.		CH ₄ = 5.46 - 3.9 = 1.56 ccs.	
CO = CO ₂ - CH ₄		CO = CO ₂ - CH ₄	
CO = 2.40 - 1.58		CO = 2.4 - 1.56	
CO = .82 ccs.		CO = 0.84 ccs.	

RESULT:-

2.15% CO ₂
0.26% C _m H _n
9.73% H ₂
3.82% CH ₄
1.98% CO
14.49% O ₂
67.57% N ₂
<u>100.00</u>

RESULT:-

2.00% CO ₂
0.25% C _m H _n
9.75% H ₂
3.90% CH ₄
2.10% CO
14.49% O ₂
67.51% N ₂
<u>100.00</u>

COMPARISON OF AVERAGE RESULTS OF ANALYSES OF
GAS MIXTURES.

I. Bar. 30.24. At. Temp. 60° F.
Casing at Atmospheric temp.

CO ₂	0.83%	
C _m H _n	0.25%	
H ₂	10.39%	18.09%
CH ₄	3.80%	
CO	2.82%	
O ₂	15.80%	
N ₂	66.11%	81.91%

Bar. 30.54. At Temp. 57° F.
Casing at 300° F when gas ad:
mitted.
147° F when sample with:
drawn.

CO ₂	2.11%	
C _m H _n	.26%	
H ₂	9.74%	18.01%
CH ₄	3.86%	
CO	2.04%	
O ₂	14.49%	
N ₂	67.50%	81.99%

EXPLOSIVE MIXTURES CONTAINING METHANE AND TRACES OF
HYDROGEN.

I. Barometric Pressure. 29.42"
Atmospheric Temperature. 59.0 F.
Temperature of Casing °
when gas was admitted. 59.0 F.
Temperature of Casing °
when sample was with: °
drawn. 59.0 F.
Time between admission
of gas and drawing of
sample. 1 minute.

(a)
Sample 32.8 ccs.
KOH 32.8 ccs.
Br. + KOH 32.8 ccs.
Pyre. 26.93 ccs. 5.87cc
O₂

RESULT:-

17.92% O₂

(b)

Sample 25.20 ccs.

Comb. 20.50 ccs.

KOH 19.00 ccs. .. 6.20 ccs. Tot. Cont. KOH

Pyro. 18.60 ccs. .. 0.4 ccs. O₂ left. Pyro.

O₂ present in sample = 4.51 ccs.

O₂ after combustion. = 0.40 ccs.

∴ O₂ used in com:
bustion. = 4.11 ccs.

Let T = total contraction due to
combustion and absorption of
CO₂ formed.

Let O = Oxygen used in combustion.

Then $H_2 = \frac{4}{3} T - 2O$

$CH_4 = O - \frac{1}{3} T$

∴ $H_2 = \frac{4}{3} \times \frac{6.2}{1} = 8.22$
= 8.26 - 8.22

$H_2 = .04$ ccs.

$CH_4 = 4.11 - \frac{6.2}{3} = 4.11 - 2.06$

$CH_4 = 2.05$ ccs.

RESULT:-

0.16% H₂ 17.92% O₂
8.13% CH₄ 73.79% N₂

(a)

Sample 21.60 ccs.

Comb. 17.60 ccs.

KOH 16.30 ccs. ... 5.30 ccs.
Tot. Cont. KOH
Pyro. 16.00 ccs. .. 0.3 ccs.
O₂ left.

O₂ in sample = 3.82 ccs.

O₂ after com:
bustion. = .30 ccs.

O₂ used in combustion = 3.52 ccs

$H_2 = \frac{4}{3} T - 2O$

$H_2 = \frac{4}{3} \times \frac{5.3}{1} = 2 \times 3.52$

$H_2 = \frac{21.2}{3} - 7.04 = 7.06 - 7.04$

$H_2 = 0.04$ ccs.

$CH_4 = O - \frac{1}{3} T$

$CH_4 = 3.52 - \frac{5.3}{3} = 3.52 - 1.76$

$CH_4 = 1.76$ ccs.

RESULT:-

0.09% H₂
8.14% CH₄
17.92% O₂
73.85% N₂

2.

Barometric Pressure 29.34"

Atmospheric Temperature 58.0°F.

Temperature of casing when
gas was admitted. 300°F.

Temperature of casing when
sample was withdrawn. 182°F.

Time between admission of
gas and drawing of sample 9 mins.

(a)

Sample 34.2 ccs.

Pyro. 28.4 ccs. 5.80 ccs. O₂

Result 17.0% O₂

(b)

Sample 25.0 ccs.

Pyro. 20.7 ccs. 4.3 ccs. O₂

Result 17.2% O₂

Average Result 17.1% O₂

(c)

Sample 29.4 ccs.
Comb. 24.2 ccs.
KOH 22.0 ccs. 7.4ccs.Tot.Cent.
Pyro. 21.9 ccs. 0.1 ccs.O₂ left.

O₂ present in Sample 5.02 ccs.
O₂ left after comb. 0.10 ccs.
O₂ used in comb. 4.92 ccs.

$$\begin{aligned} H_2 &= \frac{4}{3} T - 2.0 \\ H_2 &= \frac{4}{3} \times 7.4 - 2 \times 4.92 \\ H_2 &= \frac{29.6}{3} - 9.84 = 9.86 - 9.84 \\ H_2 &= .02 \text{ ccs.} \end{aligned}$$

$$\begin{aligned} CH_4 &= 0 - \frac{1}{3} T \\ CH_4 &= 4.92 - \frac{7.4}{3} = 4.92 - 2.46 \\ CH_4 &= 2.46 \text{ ccs.} \end{aligned}$$

Result:- 0.08% H₂
8.36% CH₄
17.10% O₂
74.46% N₂

(d)

Sample 64.4 ccs.
Comb. 52.8 ccs.
KOH 48.25 ccs.16.15ccs.Tot.
Con.
Pyro. 48.05 ccs. 0.2 ccs. O₂ left.

O₂ present in Sample 10.94ccs.
O₂ left after comb. .20ccs.
O₂ used in comb. 10.74ccs.

$$\begin{aligned} H_2 &= \frac{4}{3} T - 20 \\ H_2 &= \frac{4}{3} \times 16.15 - 2 \times 10.74 \\ &= \frac{64.6}{3} - 21.48 \\ H_2 &= 21.55 - 21.48 = .05 \text{ ccs.} \end{aligned}$$

$$\begin{aligned} CH_4 &= 0 - \frac{1}{3} T \\ CH_4 &= 10.74 - \frac{16.15}{3} \\ &= 10.74 - 5.38 \\ CH_4 &= 4.36 \end{aligned}$$

Result:- 0.07% H₂
8.32% CH₄
17.10% O₂
74.51% N₂

COMPARISON OF AVERAGE RESULTS.

I. Bar 29.42. At temp.59°F.
Casing at Atmospheric Temp.

H₂ = 0.12%
CH₄ = 8.14%
O₂ = 17.92%
N₂ = 73.82%

Bar. 29.34. At Temp. 58.0°F.
Casing at 300°F when gas ad:
mitted.
Casing at 182°F when Sample
drawn.

H₂ = 0.08%
CH₄ = 8.34%
O₂ = 17.10%
N₂ = 74.48%

Where heating is primarily due to some electrical defect, very high temperatures may be reached and result in destruction of insulation etc.

If temperature of 300^o C, are reached in the presence of an explosive mixture of gases, prolonged contact with the heated source will tend to oxidation of the combustible constituents with probable production of a non-explosive mass.

With higher temperatures the rate of heat production due to oxidation, may be greater than the rate of heat dissipation, and if this abnormal state of affairs be allowed to continue in: :flammation of the gas mixture will occur.

However, the smell, smoke, etc., which must be produced under such circumstances, will direct attention to the cause of the disturbance and should result in closing down of machinery etc., until the defect is remedied.

PERMISSIBLE AMOUNT OF WEAR AT BEARINGS:-

With "protected" motors we have to consider the possibility of passage of flame between the shaft and bearing, in event of excessive wear.

From consideration of results already obtained we are of the opinion that with plain bearings or ball bearings, the length of bearing necessary to support the shaft and armature or rotor, is more than actually required to prevent passage of flame.

The wear of bearing and shaft will produce a crescent shaped gap, the maximum depth of which will be the difference in diameters of the bearing and shaft.

If the length of shaft were say, 4 inches, the gap necessary for passage of flame along the shaft, to produce external ignition would be from $\frac{3}{16}$ th." to $\frac{1}{4}$ ".

In actual practice such wear would never be permitted, since serious electrical disturbances or magnetic unbalance etc., would be produced, and necessitate renewal or repair of bearing.

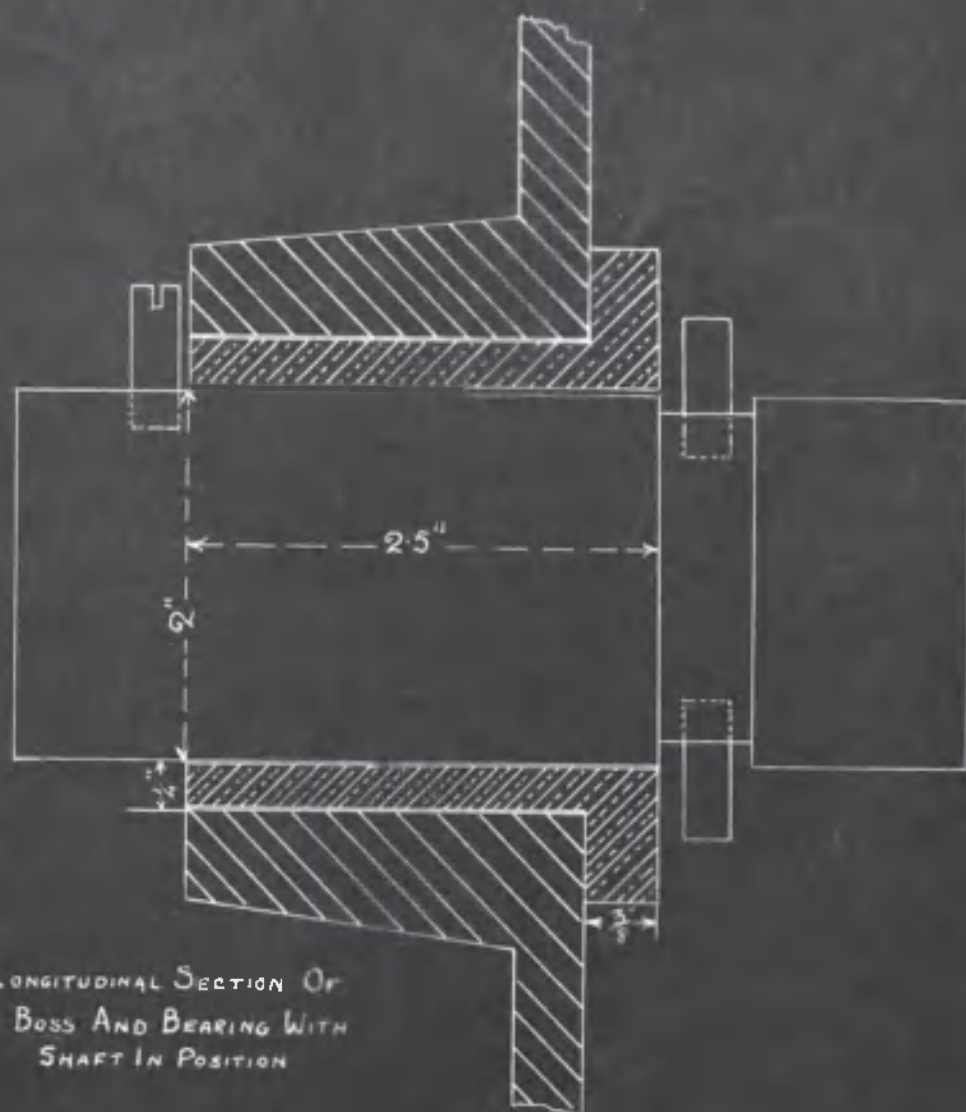


FIG. 25 :- LONGITUDINAL SECTION OF
BOSS AND BEARING WITH
SHAFT IN POSITION

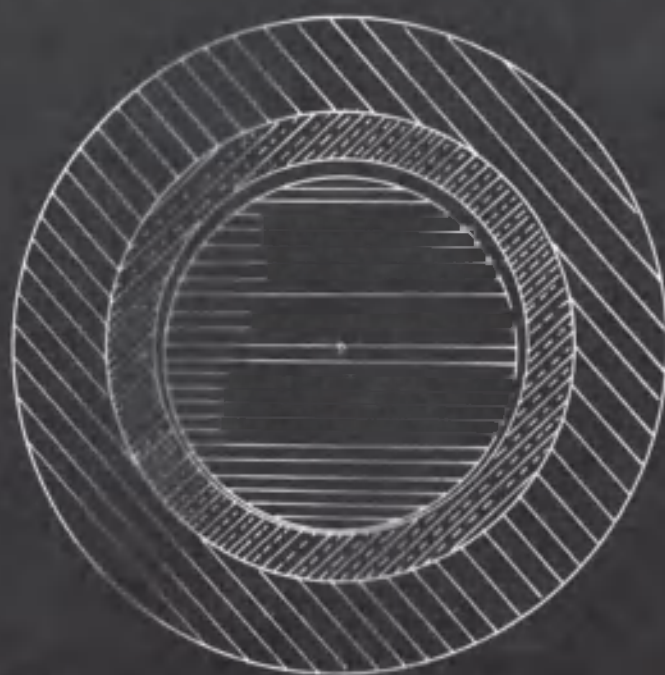


FIG. 26 :- CROSS SECTION OF BOSS,
BEARING, AND SHAFT. NOTE
GAP BETWEEN SHAFT AND
BEARING, SUCH AS MIGHT BE
FORMED BY EXCESSIVE WEAR.

Experiments were carried out on a box of volume 2.55 cub. ft., fitted with machined flanges 1.375 inches broad. The box was made of cast iron.

A white-metal bearing was cast into an opening in the side of the casing, the dimensions of the bearing being 2 ins. internal diameter, and 2.5 ins. external diameter, and 2.5 ins. long.

The steel shaft which fitted into this bearing, was 7 ins. long and projected 2 ins. from the bearing.

A groove 0.125 ins. deep was cut in the shaft at the inside end of the bearing, and in this groove four studs were set, to prevent the shaft from being forced out during an internal explosion.

After an explosion in a partially closed vessel such as the above, condensation and cooling occur, and pressure falls below atmospheric.

It follows therefore that the shaft would be drawn into the casing after an explosion, if some means were not adopted to prevent such an occurrence.

A screw-stud was therefore set in the shaft at the outside of the bearing, and had the desired effect.

The arrangement is clearly shown in fig. 25. One end of the casing was tapped to receive a gas tap and a plug for carrying the igniting current cable, and a gap 0.01 ins. in depth was left at the flanges.

The procedure was to admit an explosive mixture of coal gas and air to the casing and ignite it by blowing a "tin" fuse.

The length of fuse used was 1.5 ins. (No. 24. S.W.G.), the igniting current being 18 amperes at 250 Volts D.C.

In the initial experiments the bearing and shaft were a "tight fit", but as the work proceeded, the shaft was gradually reduced in diameter by predetermined amounts, strict observation

being kept to ascertain when flame first appeared at the bearing.

At each reduction in diameter of shaft, a number of tests were carried out in the "open", followed immediately by similar tests with the casing surrounded by an explosive atmosphere.

In none of the latter experiments was it possible to produce external ignition.

Occasionally when 'proving' the external mixture after an internal explosion, it was possible to see the flame being drawn into the casing, between the flanges and also at the bearing, the action producing a whistling sound.

In a large number of the tests the screw stud was withdrawn to see whether ignition of gas would occur when the external mixture entered the casing in the wake of the shaft, after the internal explosion.

Rapid cooling must occur since ignition of the external mixture could not be produced in this manner, but often when proving the external mixture a slight explosion could be heard within the casing, proving the existence of an explosive mixture which must have entered when the shaft was "drawn" in.

In the first series of experiments in the "open", a white emanation was discernible at the bearing during ignition.

As the shaft was reduced in diameter, this "emanation" gradually gave place to distinct brownish flames, with wisps of blue and a few "sparks", then as the gap became greater bright red flames and brilliant sparks were observed during ignition.

When a shaft diameter of 1.945 ins. (giving a gap of 0.055 ins.) was reached reddish-blue flames extending to the end of the shaft were apparent, and as the flames increased in luminosity the white emanation was no longer visible.

After experimenting with the shaft dry, a film of oil was introduced. The immediate effect was considerable shortening of the flame, and in many experiments complete quenching of flame resulted.

Considerable quantities of smoke were expelled during

the explosion, and the shaft on the outside of the bearing was covered with globules of oil.

The burning of oil between the shaft and the bearing would account for the volume of smoke at the bearing, while the globules of oil found on the shaft would be deposited there from spray of oil ejected by the gases from the explosion.

After this observation, it might be suggested that the white emanation spoken of above, was merely products of combustion expelled by the explosion. On examining the side of the "explosion chamber" a deposit of grey matter was found directly opposite the bearing indicating that the above view is probably correct.

The fifth experiment with the shaft reduced to 1.945 inches ended disastrously. The casing burst under the strain of the explosion and damaged the explosion chamber.

Part of the casing containing the shaft was carried through the end of the chamber and came to rest 10 feet from it.

On examination it was found that the casing had been sheared along the junction of the sides and bottom.

The gap at the bearing (0.055 ins) would never have been allowed to form in an actual machine, since disturbances etc., already mentioned would have necessitated renewal and repair before the wear had become so marked.

The results obtained are shown tabulated at the end of the thesis.

A smaller box made of cast steel, was now procured and adapted for experimental purposes in the manner already described. The volume of this box was approximately 0.123 cub. ft., with flanges 1 inch broad. The bearing was of brass and of the following dimensions, length 1.625 ins., internal diameter 0.625 inch, external diameter 0.875 inch.

The shaft was made of mild steel, and was 4 inches long, and had a diameter of 0.625 inch.

The shaft and bearing were intended to represent, a

controller shaft and bearing.

In the first series of experiments carried out on this box the shaft and bearing were a "tight fit."

The bearing was increased in internal diameter by degrees until flame capable of causing ignition finally passed.

Sparks and flame were first observed with a gap of 0.047 in. At the bearing, the flame being blue in colour and extending only a short distance.

As the gap at the bearing was increased, the flame from the explosions increased in length, and changed in colour from blue to red, while at very large gaps of the order of 0.10 inch the flame at the gap was a vivid white colour.

External ignition was ultimately produced when the difference in diameter between the shaft and bearing was 0.125 ins.

The flame produced in this case extended approximately 3 inches beyond the bearing.

Such excessive wear would never be permitted in a controller shaft and bearing, since faulty contact of studs, fingers etc., would be produced giving endless trouble, and necessitating attention before a gap of such dimensions had been formed. The results will be found tabulated at the end.

CONCLUSIONS:-

Although the first series of experiments could not be completed, sufficient data was obtained to indicate that in general the length of bearing required to support the shaft etc., is more than sufficient to prevent passage of flame between the shaft and bearing.

The danger of passage of flame need not be the deciding factor in determining the permissible amount of wear.

The factor of greatest importance is the prevention of electrical or magnetic disturbances, and if strict attention be directed to this phase of the question, no fear need be entertained regarding the possibility of ignition of gas by leakage at

the bearing.

It was found possible to complete the second series of experiments and the results are an interesting addition to those of the first series. The production of faulty contact, due to excessive wear at the bearing, has been mentioned. Acting thereby produced would increase the possibility of ignition of explosive gas mixtures present in the controller casing. It follows that if the casing was not adequately^{protected}/external ignition might be produced, but the likelihood of passage of flame between the shaft and bearing is remote, since it is hardly conceivable that apparatus would be allowed to fall into such a state of disrepair that a difference of 0.125 inch in diameters of shaft and bearing existed.

We may conclude therefore that if bearings are renewed when trouble in making contact is experienced, there will be no danger of passage of flame between the shaft and bearing, capable of igniting an external mixture.

GENERAL SUMMARY OF RESULTS AND CONCLUSIONS:-

1. Cotton gaskets do not constitute a permanent flame proof joint, because they deteriorate rapidly in humid atmosphere.
2. Flanges 0.5 inch broad do not adequately protect a casing. They do not prevent passage of flame capable of causing ignition of external gas mixture, unless they are machined and in metallic contact. Even under these conditions the margin of safety is small.
3. An adequately protected casing, maybe rendered unsafe by attaching an inadequately protected piece of apparatus to it. All apparatus should be protected to the same degree.

4. Where flanges are internal (i.e. extend inwards from the side of the casing), all stud holes should be bottomed, since omission of a stud would provide direct communication with the outside atmosphere.

5. Ignition may be produced by the operation of overload releases.

6. Cooling of gases from an internal explosion may be effected by either of two methods, or a combination of these two methods. The two Methods are (a) Cooling by contact with a large metallic surface. (b) Cooling by expansion through a small orifice. Cooling by expansion predominates where the release area is small, consequently pressures developed by the explosion of gas mixtures are relatively high.

Cooling by contact occurs when a large release area is provided. In this case high pressure development is prevented.

Where a gap is left all round the perimeter of an opening which is protected by broad flanges, large cooling surface and large release area are combined. This state of affairs satisfies the definition of the "flame-proof casing definition of the B.E.S.A., since cooling of flame is adequate and pressures developed are low.

Devices which provide a large cooling surface and a small release area do not completely satisfy the requirements.

7. There is a distinct "flame-passing range" in gas mixtures, and for a given depth and length of gap between flanges of given breadth, the higher the value of the igniting current (till 40 amps. is reached) the

more nearly does the limits of this range approach the limits of inflammability of the mixture.

The phenomenon is observable when the "margin of safety" is small.

8. Coal dust present in switch gear casings will probably interfere with the electrical continuity, and in event of an explosion of gas occurring in the casing, will tend to produce slightly higher pressures, but the flame-produced is not so intense as from an explosion of gas alone.

If the correct proportions of coal dust and gas are present very high pressures will be developed.

9. Increase of value of igniting current has no effect on ultimate pressure developed, but probably affects the rate of pressure development.

10. Pressure developed is dependent upon the release area provided. Where a large release area is provided a large mass of gas is expelled unburned when ignition occurs. This tends to reduce temperature and pressure attained, since these depend on amount of gas consumed.

11. WEBBED COVERS:- The provision of internal webs in casings is worth considering. These in addition to strengthening the casing increase the cooling surface presented to the flame when ignition occurs. This tends to lower the temperatures attained.

12. HEATED CASINGS:- If casings are adequately protected (i.e. have a good margin of safety) no doubt need be entertained regarding the ability of protective measures to function properly at temperatures considerably in excess of that of the surrounding atmosphere.

Increase in temperature of the casing, whilst reducing the efficiency of the cooling surface also

causes reduction of nett weight of gas contained in the casing.

This effect more than compensates for loss in efficiency of protective measures.

13. PERMISSIBLE WEAR OF BEARINGS:- With plain bearings the length required to support the load is more than necessary to prevent passage of flame along the shaft through a gap formed by reasonable wear between shaft and bearing.

Before conditions are such that flame capable of causing external ignition can pass, the electrical and magnetic disturbances will be of such a magnitude that the apparatus will have to be closed down for renewal of bearings, etc.

With controller shaft bearings the length should be from 2-3 times the diameter of the shaft, and renewals should be made when the amount of wear is such that trouble is experienced in making contact.

In each case, if the remainder of the apparatus is adequately protected there will be no danger of external ignition.

RECOMMENDATIONS:-

Broad flanges with a gap between them should be provided at all joints.

In deciding the depth of gap to be left between flanges the volume of the casing, the perimeter of the opening and the breadth of flange should all be considered..

Figs. 9 and 10 have been drawn from results obtained by experiment, and by referring to the curves the maximum depth of gap and necessary breadth of flange will be found for volumes and perimeter within the limits plotted.

The gap may be provided in a variety of ways, e.g. by means of spring washers, or studs projecting from the surface of the casing flange, but an allowance should be made for ~~working~~^{walking} at the ends of covers and casings, otherwise the gaps at the ends may be considerably in excess of the safe perimeter gap.

Cast steel casings are preferable to cast-iron casings, unless strengthening webs are provided in the latter.

In the experiments described in the foregoing pages, and in the interpretation of results etc., the practical aspect of the question has been considered of primary importance.

The conclusions arrived at and the recommendations and suggestions advanced are therefore made on this basis.

At all times every effort was made to reproduce a practical atmosphere, and have conditions resembling as closely as possible, those met with daily in mines.

The author desires to place on record his indebtedness to Professor Burns and Professor Parker-Smith of the Royal Technical College, Glasgow, under whose direction he worked. The kindly guidance and advice of these two gentlemen were much appreciated.

No	GAS METER READING	Gas in CUBIC Ft.		BURNING CURRENT IN AMPERES	BRASS OF FLANGE	G A P		IGNITION		REMARKS
		E.C.	S.B.			LENGTH IN INCHES	DEPTH IN INCHES	INTERNAL	EXTERNAL	
1	99.7-99.98 99.917	.36	.043	29	1"	.5	.023	YES	YES	Violent EXPLOSION RINGS BOLT SHEARED FALLING COVER TO LIFT AND ALLOW FLAME TO PASS.
2	99.98-99.99 99.985	.356	.029	"	"	"	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
3	100.00-100.01 100.002	.392	.030	"	"	"	"	YES	NO	Do
4	100.00-100.01 100.007	.413	.022	"	"	"	"	YES	NO	Do
5	100.00-100.01 100.007	.476	.046	"	"	"	"	YES	NO	Do
6	100.00-100.01 100.009	.423	.049	"	"	"	"	YES	NO	Do
7	100.00-100.01 100.009	.436	.044	"	"	"	"	YES	NO	Do
8	100.00-100.01 100.009	.446	.044	"	"	.375	.020	YES	NO	Do
9	100.00-100.01 100.009	.310	.035	"	"	"	"	YES	NO	Do Do
10	100.00-100.01 100.009	.355	.038	"	"	"	"	YES	NO	Do Do
11	100.00-100.01 100.009	.476	.057	"	"	"	"	NO	NO	FUSE CONNECTION BROKEN
12	100.00-100.01 100.009	.390	.074	"	"	.5	.054	YES	YES	THREAD OF WING NUT STRIPPED CAUSING COVER TO LIFT
13	100.00-100.01 100.009	.305	.071	"	"	"	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
14	100.00-100.01 100.009	.310	.061	"	"	"	"	YES	NO	Do Do
15	100.00-100.01 100.009	.360	.039	"	"	"	"	YES	NO	Do Do
16	100.00-100.01 100.009	.322	.034	"	"	"	"	YES	NO	Do Do
17	100.00-100.01 100.009	.300	.037	"	"	.375	.072	YES	NO	Do Do
18	100.00-100.01 100.009	.367	.049	"	"	"	"	YES	NO	Do Do
19	100.00-100.01 100.009	.326	.046	"	"	.50	.100	YES	NO	Do Do
20	100.00-100.01 100.009	.306	.043	"	"	"	"	YES	NO	Do Do
21	100.00-100.01 100.009	.346	.063	"	"	"	"	YES	YES	WOOD EXPLOSION CHAMBER USED VOLUME 16.6 CUBIC FEET
22	100.00-100.01 100.009	.345	.064	"	"	"	"	YES	YES	Do Do
23	100.00-100.01 100.009	.301	.067	"	"	"	.046	YES	YES	TWO DISTINCT EXPLOSIONS
24	100.00-100.01 100.009	.367	.035	"	"	.625	.086	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
25	100.00-100.01 100.009	.340	.036	"	"	"	.100	YES	NO	Do Do
26	100.00-100.01 100.009	.344	.039	"	"	.75	.073	YES	NO	Do Do
27	100.00-100.01 100.009	.312	.080	"	"	"	"	YES	NO	Do Do
28	100.00-100.01 100.009	.310	.038	"	"	.60	.080	YES	YES	
29	100.00-100.01 100.009	.367	.031	"	"	"	"	YES	YES	
30	100.00-100.01 100.009	.340	.030	"	"	"	.100	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
31	100.00-100.01 100.009	.324	.030	25	"	"	.080	YES	YES	TWO DISTINCT EXPLOSIONS
32	100.00-100.01 100.009	.331	.027	"	"	"	"	YES	YES	EXTERNAL MIXTURE EXPLOSIVE
33	100.00-100.01 100.009	.312	.040	20	"	"	.060	YES	NO	Do Do
34	100.00-100.01 100.009	.309	.030	"	"	"	"	YES	NO	Do Do
35	100.00-100.01 100.009	.327	.038	"	"	"	"	YES	NO	Do Do
36	100.00-100.01 100.009	.322	.036	12	"	.375	.080	YES	NO	Do Do
37	100.00-100.01 100.009	.321	.038	"	"	.625	"	YES	NO	Do Do
38	100.00-100.01 100.009	.306	.035	"	"	.475	"	YES	NO	Do Do
39	100.00-100.01 100.009	.303	.033	"	"	"	"	YES	NO	Do Do
40	100.00-100.01 100.009	.320	.037	"	"	.70	"	NO	NO	FUSE CONNECTION BROKEN
41	100.00-100.01 100.009	.315	.034	"	"	"	"	YES	YES	
42	100.00-100.01 100.009	.335	.033	"	"	.75	.058	YES	YES	
43	100.00-100.01 100.009	.335	.036	"	"	"	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
44	100.00-100.01 100.009	.337	.039	"	"	"	"	YES	YES	TWO DISTINCT EXPLOSIONS
45	100.00-100.01 100.009	.326	.035	"	"	"	"	YES	YES	Do Do
46	100.00-100.01 100.009	.346	.040	"	"	"	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE.

No.	GAS METER READING	GAS IN CUBIC FEET		CURRENT IN AMPERES	BURNING OF FLAME	GAS		IGNITION		REMARKS
		EC	SB			LENGTH IN FEET	DIA. IN INCHES	INTERNAL	EXTERNAL	
47	12 632-12 669 31.00	3 322	037	12	1"	1.75	.056	Yes	No	INTERNAL MIXTURE TOO RICH EXTERNAL " EXPLOSIVE
48	12 669-12 699 31.00	3 327	037	"	"	"	"	Yes	Yes	
49	12 699-12 729 31.00	3 332	037	"	"	"	"	Yes	Yes	
50	12 729-12 759 31.00	3 337	037	"	"	1.5	"	Yes	Yes	
51	12 759-12 789 31.00	3 342	037	"	"	1.5	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
52	12 789-12 819 31.00	3 347	037	"	"	1.375	"	Yes	Yes	TWO DISTINCT EXPLOSIONS
53	12 819-12 849 31.00	3 352	037	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
54	12 849-12 879 31.00	3 357	037	"	"	"	"	Yes	Yes	TWO DISTINCT EXPLOSIONS
55	12 879-12 909 31.00	3 362	037	"	"	"	"	Yes	Yes	
56	12 909-12 939 31.00	3 367	037	"	"	1.25	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
57	12 939-12 969 31.00	3 372	037	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
58	12 969-12 999 31.00	3 377	037	"	"	1.75	.040	Yes	No	D ₀ D ₀ D ₀
59	13 000-13 030 31.00	3 382	037	"	"	2.25	.040	Yes	No	D ₀ D ₀ D ₀
60	13 030-13 060 31.00	3 387	037	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
61	13 060-13 090 31.00	3 392	037	"	"	2.75	"	Yes	No	D ₀ D ₀ D ₀
62	13 090-13 120 31.00	3 397	037	"	"	3.5	"	Yes	No	D ₀ D ₀ D ₀
63	13 120-13 150 31.00	3 402	037	"	"	4.0	"	Yes	No	D ₀ D ₀ D ₀
64	13 150-13 180 31.00	3 407	037	"	"	4.6	.038	Yes	No	D ₀ D ₀ D ₀
65	13 180-13 210 31.00	3 412	037	"	"	4.75	"	Yes	No	D ₀ D ₀ D ₀
66	13 210-13 240 31.00	3 417	037	"	"	6.5	.040	Yes	No	D ₀ D ₀ D ₀
67	13 240-13 270 31.00	3 422	037	"	"	6.5	.042	Yes	No	D ₀ D ₀ D ₀
68	13 270-13 300 31.00	3 427	037	"	"	6.5	"	Yes	No	D ₀ D ₀ D ₀
69	13 300-13 330 31.00	3 432	037	"	"	16.0	.038	Yes	Yes	VIOLENT EXPLOSION
70	13 330-13 360 31.00	3 437	037	"	"	10.75	"	Yes	Yes	
71	13 360-13 390 31.00	3 442	037	"	"	5.75	"	Yes	Yes	
72	13 390-13 420 31.00	3 447	037	"	"	9.0	"	Yes	No	FEEBLE INTERNAL EXPLOSION EXTERNAL MIXTURE EXPLOSIVE
73	13 420-13 450 31.00	3 452	037	"	"	9.0	"	Yes	Yes	VIOLENT EXPLOSION
74	13 450-13 480 31.00	3 457	037	"	"	5.0	"	Yes	Yes	TWO DISTINCT EXPLOSIONS
75	13 480-13 510 31.00	3 462	037	"	"	7.0	.045	Yes	Yes	
76	13 510-13 540 31.00	3 467	037	"	"	6.75	"	Yes	Yes	
77	13 540-13 570 31.00	3 472	037	"	"	6.0	"	Yes	Yes	
78	13 570-13 600 31.00	3 477	037	"	"	5.75	"	Yes	Yes	
79	13 600-13 630 31.00	3 482	037	"	"	5.25	"	Yes	Yes	
80	13 630-13 660 31.00	3 487	037	"	"	4.125	"	Yes	Yes	TWO DISTINCT EXPLOSIONS
81	13 660-13 690 31.00	3 492	037	"	"	4.00	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
82	13 690-13 720 31.00	3 497	037	"	"	4.125	"	Yes	Yes	PARSE KNOCK DOWN BY TRAIL BRIDGE SMALL MIXTURE, EXTERNAL MIXTURE FLEW
83	13 720-13 750 31.00	3 502	037	"	"	2.9	.021	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
84	13 750-13 780 31.00	3 507	037	"	"	"	"	Yes	No	D ₀ D ₀
85	13 780-13 810 31.00	3 512	037	"	"	"	"	Yes	No	D ₀ D ₀
86	13 810-13 840 31.00	3 517	037	"	"	"	"	Yes	No	D ₀
87	13 840-13 870 31.00	3 522	037	"	"	"	.035	Yes	Yes	I
88	13 870-13 900 31.00	3 527	037	"	"	"	.030	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
89	13 900-13 930 31.00	3 532	037	"	"	"	.030	Yes	No	D ₀ D ₀
90	13 930-13 960 31.00	3 537	037	"	"	"	.030	Yes	No	D ₀ D ₀
91	13 960-13 990 31.00	3 542	037	"	"	"	.030	Yes	No	D ₀ D ₀
92	14 000-14 030 31.00	3 547	037	"	"	"	.030	Yes	No	D ₀ D ₀

No	GAS METER READING	GAS IN CUBIC FT EC 5B	CURRENT IN AMPERE	BURNING OF FLAME	GAP		IGNITION		REMARKS
					LENGTH IN INCHES	DEPTH IN INCHES	INTERNAL	EXTERNAL	
93	25-2457	2-905	316	"	5	100	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
94	27-212	2-905	312	"	6.25	100	Yes	No	Do Do
95	30-226	2-900	312	"	11/16	100	Yes	No	Do Do
96	36-156	2-900	312	"	6.25	100	Yes	No	Do Do
97	40-245	2-900	312	"	8.75	"	Yes	No	Do Do
98	42-275	2-900	316	"	"	"	Yes	No	Do Do
99	46-575	2-900	312	"	11.25	"	Yes	Yes	
100	48-645	2-900	320	"	"	"	Yes	Yes	
101	49-650	2-900	320	"	1-0	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
102	53-537	2-900	312	"	"	"	Yes	No	Do Do
103	59-675	2-900	312	"	2-25	1070	Yes	No	Do Do
104	62-696	2-900	312	"	2.375	"	Yes	No	Do Do
105	62-696	2-900	312	"	"	"	Yes	No	Do Do
106	65-706	2-900	312	"	2-75	"	Yes	No	Do Do
107	68-720	2-900	300	"	2-25	"	Yes	No	Do Do
108	69-725	2-900	300	"	2-625	"	Yes	No	Do Do
109	72-765	2-900	300	"	2-875	"	Yes	No	Do Do
110	75-805	2-900	300	"	3-00	"	Yes	Yes	
111	78-812	2-900	300	"	3-00	"	Yes	Yes	
112	81-812	2-900	300	"	6-00	1052	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
113	81-812	2-900	300	"	7-0	1052	Yes	No	Do Do
114	81-812	2-900	300	"	7-5	"	Yes	No	Do Do
115	81-812	2-900	312	"	8-0	"	Yes	Yes	
116	81-812	2-900	312	"	7-75	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
117	81-812	2-900	312	"	7-875	"	Yes	No	Do Do
118	81-812	2-900	312	"	7-875	"	Yes	No	Do Do
119	81-812	2-900	312	"	8-00	"	Yes	No	Do Do
120	81-812	2-900	312	"	8-25	"	Yes	No	Do Do
121	81-812	2-900	312	"	"	"	Yes	Yes	TWO DISTINCT EXPLOSIONS.
122	81-812	2-900	312	"	"	"	Yes	No	
123	81-812	2-900	312	"	8-50	"	Yes		
124	81-812	2-900	312	"	8-75	"	Yes		Do Do
125	81-812	2-900	312	"	9-00	"	Yes		Do Do
126	81-812	2-900	312	"	9-25	"	Yes		Do Do
127	81-812	2-900	312	"	9-50	"	Yes		Do Do
128	81-812	2-900	312	"	"	"	Yes		Do Do
129	81-812	2-900	312	"	"	"	Yes	No	Do Do
130	81-812	2-900	312	"	10-25	"	Yes	Yes	Do Do
131	81-812	2-900	312	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
132	81-812	2-900	312	"	12-5	"	Yes	No	Do Do
133	81-812	2-900	312	"	"	"	Yes	No	Do Do
134	81-812	2-900	312	"	14-5	"	Yes	No	Do Do
135	81-812	2-900	312	"	"	"	Yes	Yes	WOODEN PLUG BURNED OUT
136	81-812	2-900	312	"	"	"	Yes	Yes	TWO DISTINCT EXPLOSIONS
137	81-812	2-900	312	"	"	1038	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
138	81-812	2-900	312	"	16-5	1052	Yes	Yes	TWO DISTINCT EXPLOSIONS.

No	GAS METER READING	GAS IN Cubic feet		CURRENT IN AMPERES	BREADTH OF FLANGE	GAP		IGNITION		REMARKS.
		EC	SB			LENGTH IN INCHES	DEPTH IN INCHES	INTERNAL	EXTERNAL	
139	312-84662		320	15	1-375	18.5	.052	YES	YES	
	57-362	2-700								
140	8734-87676		312			16	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
	On 376	2-700								
141	70469		313					YES	YES	GAS PIPE PLUG BLOWN OUT
	93383	2-700								
	93702		313							
142	96602	2-700						YES	YES	TWO DISTINCT EXPLOSIONS
	96774		313			15		YES	NO	EXTERNAL MIXTURE EXPLOSIVE
143	99416	2-700								
	99526		312					YES	NO	D ₀ D ₀ D ₀
144	103476	2-700						YES	NO	D ₀ D ₀
	2-735		312			16	"	YES	NO	
145	3-738	2-700								
	3-750		312			16.75	"	YES	YES	
146	8-730	2-700								
	8-762		312			27	.04	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
147	11-662	2-700								
	11-770		312			ALL ROUND	"	YES	YES	
148	4-531	2-776								
	14-884		313			24	.037	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
149	17-064	2-700								
	17-896		313			27	"	YES	NO	D ₀ D ₀
150	20-896	2-700								
	20-896		316			30	"	YES	NO	D ₀ D ₀
151	20-896	2-700								
	20-896		313			40	"	YES	NO	D ₀ D ₀
152	21-914	2-700								
	21-914		313			ALL ROUND	.036	YES	NO	D ₀ D ₀
153	22-914	2-700								
	22-914		303					YES	NO	D ₀ D ₀
154	23-914	2-700								
	23-914		312					YES	NO	D ₀ D ₀
155	24-914	2-700								
	24-914		315		1-75	1-0	.05	YES	NO	D ₀ D ₀
156	25-914	2-700								
	25-914		350			2-0	"	YES	YES	SHORT DISTANCE PIECES BLOWN OUT FROM BETWEEN FLANGES
157	26-914	2-700								
	26-914		350			2-75	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
158	27-914	2-700								
	27-914		350			3-5	"	YES	NO	D ₀ D ₀
159	28-914	2-700								
	28-914		350			3-5	"	YES	NO	D ₀ D ₀
160	29-914	2-700								
	29-914		350			4-25	"	YES	NO	D ₀ D ₀
161	30-914	2-700								
	30-914		350			6-25	"	YES	NO	D ₀ D ₀
162	31-914	2-700								
	31-914		350			7-25	"	YES	NO	D ₀ D ₀
163	32-914	2-700								
	32-914		350			7-25	"	YES	YES	
164	33-914	2-700								
	33-914		350			6-50	"	YES	YES	
165	34-914	2-700								
	34-914		350			6-25	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
166	35-914	2-700								
	35-914		350			ALL ROUND	.060	NO	NO	FUSE CONNECTION FOUND BROKEN
167	36-914	2-700								
	36-914		350			10	.035	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
168	37-914	2-700								
	37-914		350			10	.046	YES	YES	
169	38-914	2-700								
	38-914		350			10	.049	YES	YES	TWO DISTINCT EXPLOSIONS
170	39-914	2-700								
	39-914		350			10	.039	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
171	40-914	2-700								
	40-914		350			11	.054	YES	NO	D ₀ D ₀
172	41-914	2-700								
	41-914		350			12	"	YES	YES	
173	42-914	2-700								
	42-914		350			11-5	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
174	43-914	2-700								
	43-914		350			12-6	"	YES	NO	D ₀ D ₀
175	44-914	2-700								
	44-914		350			14-25	"	YES	NO	D ₀ D ₀
176	45-914	2-700								
	45-914		350			12-25	"	YES	YES	TWO DISTINCT EXPLOSIONS
177	46-914	2-700								
	46-914		350			15-125	"	YES	NO	EXTERNAL MIXTURE EXPLOSIVE
178	47-914	2-700								
	47-914		350			12-25	"	YES	NO	D ₀ D ₀
179	48-914	2-700								
	48-914		350			12-25	"	YES	YES	
180	49-914	2-700								
	49-914		350			12-25	"	YES	YES	
181	50-914	2-700								
	50-914		350			12-25	"	YES	YES	
182	51-914	2-700								
	51-914		350			12-25	"	YES	YES	
183	52-914	2-700								
	52-914		350			12-25	"	YES	YES	
184	53-914	2-700								
	53-914		350			12-25	"	YES	YES	
185	54-914	2-700								
	54-914		350			12-25	"	YES	YES	
186	55-914	2-700								
	55-914		350			12-25	"	YES	YES	
187	56-914	2-700								
	56-914		350			12-25	"	YES	YES	
188	57-914	2-700								
	57-914		350			12-25	"	YES	YES	
189	58-914	2-700								
	58-914		350			12-25	"	YES	YES	
190	59-914	2-700								
	59-914		350			12-25	"	YES	YES	
191	60-914	2-700								
	60-914		350			12-25	"	YES	YES	
192	61-914	2-700								
	61-914		350			12-25	"	YES	YES	
193	62-914	2-700								
	62-914		350			12-25	"	YES	YES	
194	63-914	2-700								
	63-914		350			12-25	"	YES	YES	
195	64-914	2-700								
	64-914		350			12-25	"	YES	YES	
196	65-914	2-700								
	65-914		350			12-25	"	YES	YES	
197	66-914	2-700								
	66-914		350			12-25	"	YES	YES	
198	67-914	2-700								
	67-914		350			12-25	"	YES	YES	
199	68-914	2-700								
	68-914		350			12-25	"	YES	YES	
200	69-914	2-700								
	69-914		350			12-25	"	YES	YES	

No	GAS METER READING	GAS IN CUBIC FEET EC. SB	CURRENT IN AMPERE	BREADTH OF FLANGE	G A LENGTH INCHES	P DEPTH INCHES	IGNITION INTERNAL/EXTERNAL	REMARKS
185	5350-2647	240	15	1.75	15.25	.045	No No	FREE NOT PROBABLY CONNECTED
186	2647-2677	240	"	"	"	.050	Yes No	EXTERNAL MIXTURE EXPLOSIVE
187	2677-2697	240	"	"	14.25	.045	Yes Yes	TWO DISTINCT EXPLOSIONS
188	2697-2717	240	"	"	"	"	Yes Yes	Do Do
189	2717-2737	240	"	"	16.25	"	Yes No	Do Do
190	2737-2757	240	"	"	"	"	Yes No	Do Do
191	2757-2777	240	"	"	"	"	Yes Yes	TWO DISTINCT EXPLOSIONS
192	2777-2797	240	"	"	"	"	Yes Yes	Do Do
193	2797-2817	240	"	"	15.625	"	Yes No	EXTERNAL MIXTURE EXPLOSIVE
194	2817-2837	240	"	"	"	"	Yes Yes	TWO DISTINCT EXPLOSIONS
195	2837-2857	240	"	"	"	"	Yes No	EXTERNAL MIXTURE EXPLOSIVE
196	2857-2877	240	"	"	"	"	Yes No	Do Do
197	2877-2897	240	"	"	"	"	Yes Yes	?
198	2897-2917	240	"	"	"	"	Yes Yes	!
199	2917-2937	240	"	"	13.75	"	Yes Yes	FLAME LEAVING PAST PLUG
200	2937-2957	240	"	"	1.25	.046	Yes No	EXTERNAL MIXTURE EXPLOSIVE
201	2957-2977	240	"	"	1.50	"	Yes No	Do Do
202	2977-2997	240	"	"	3.0	"	No	Do Do
203	2997-3017	240	"	"	3.75	"	No	Do Do
204	3017-3037	240	"	"	4.50	"	No	Do Do
205	3037-3057	240	"	"	5.00	"	No	Do Do
206	3057-3077	240	"	"	5.50	"	No	Do Do
207	3077-3097	240	"	"	5.50	"	No	Do Do
208	3097-3117	350	"	"	"	"	Yes	
209	3117-3137	350	"	"	"	"	Yes Yes	
210	3137-3157	350	3.0	1	1.25	.051	Yes Yes	TWO DISTINCT REPORTS
211	3157-3177	350	3.9	"	"	"	Yes No	EXTERNAL MIXTURE EXPLOSIVE
212	3177-3197	350	3.0	"	"	"	Yes No	Do Do
213	3197-3217	350	5.0	"	"	"	Yes No	Do Do
214	3217-3237	350	7.0	"	"	"	Yes No	Do Do
215	3237-3257	350	12.0	"	"	"	Yes No	Do Do
216	3257-3277	350	3.0	"	"	"	Yes No	Do Do
217	3277-3297	350	3.0	"	"	"	Yes No	Do Do
218	3297-3317	350	4.0	"	"	"	Yes No	Do Do
219	3317-3337	350	14.0	"	"	"	Yes No	Do Do
220	3337-3357	350	3.5	"	"	"	Yes No	Do Do
221	3357-3377	350	3.5	"	"	"	Yes No	Do Do
222	3377-3397	350	14.0	"	"	"	Yes No	Do Do
223	3397-3417	350	14.0	"	"	"	Yes No	Do Do
224	3417-3437	350	14.5	"	1.56	"	Yes Yes	
225	3437-3457	350	12.5	"	"	"	Yes No	EXTERNAL MIXTURE EXPLOSIVE
226	3457-3477	350	14.5	"	"	"	Yes Yes	TWO DISTINCT EXPLOSIONS
227	3477-3497	350	14.5	"	"	"	Yes No	EXTERNAL MIXTURE EXPLOSIVE
228	3497-3517	350	14.5	"	"	"	Yes Yes	
229	3517-3537	350	14.5	"	"	"	Yes Yes	

No	Gas Mixture		Gap	Ignition	Remarks
	Reading	E.C.			
230	50-5136	032	1	No	Some Circuit in E.C.
231	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
232	40-5136	030	1	Yes	
233	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
234	40-5136	030	1	Yes	Do Do
235	40-5136	030	1	Yes	
236	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
237	40-5136	030	1	Yes	Do Do
238	40-5136	030	1	Yes	Do Do
239	40-5136	030	1	Yes	Do Do
240	40-5136	030	1	Yes	Do Do
241	40-5136	030	1	Yes	Do Do
242	40-5136	030	1	Yes	Do Do
243	40-5136	030	1	Yes	Do Do
244	40-5136	030	1	Yes	Do Do
245	40-5136	030	1	Yes	Do Do
246	40-5136	030	1	Yes	Do Do
247	40-5136	030	1	Yes	Do Do
248	40-5136	030	1	Yes	Do Do
249	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
250	40-5136	030	1	Yes	Do Do Do
251	40-5136	030	1	Yes	Do Do Do
252	40-5136	030	1	Yes	Do Do Do
253	40-5136	030	1	Yes	Do Do Do
254	40-5136	030	1	Yes	Do Do Do
255	40-5136	030	1	Yes	
256	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
257	40-5136	030	1	Yes	Do Do
258	40-5136	030	1	Yes	Do Do
259	40-5136	030	1	Yes	Do Do
260	40-5136	030	1	Yes	
261	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
262	40-5136	030	1	Yes	
263	40-5136	030	1	Yes	
264	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
265	40-5136	030	1	Yes	
266	40-5136	030	1	Yes	
267	40-5136	030	1	Yes	EXTERNAL MIXTURE EXPLOSIVE
268	40-5136	030	1	Yes	Do
269	40-5136	030	1	Yes	Do
270	40-5136	030	1	Yes	Do
271	40-5136	030	1	Yes	Do
272	40-5136	030	1	Yes	Do
273	40-5136	030	1	Yes	Do
274	40-5136	030	1	Yes	Do
275	40-5136	030	1	Yes	Do

No	GAL METER READING	GAS IN CUBIC FEET		BREADTH OF FLANGE	CURRENT IN AMPERES	G A P LENGTH IN INCHES		IGNITION		REMARKS
		EC	SB			INCHES	INCHES	INTERNAL	EXTERNAL	
276	66-127-66-27		020	1"	33	1625	058	YES	YES	
277	66-134	2700	026		33			YES	YES	
278	66-179	2700	025					YES	YES	
279	66-224	2700	015		35			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
280	66-269	2700	020		33			YES	NO	Do Do
281	66-314	2700	022		33			YES	NO	Do Do
282	66-359	2700	023		33			YES	NO	Do Do
283	66-404	2700	026		33			YES	NO	Do Do
284	66-449	2700	025		33			YES	YES	
285	66-494	2700	026		33			YES	YES	
286	66-539	2700	026		33			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
287	66-584	2700	032		33			YES	NO	Do Do
288	66-629	2700	026		49			YES	NO	Do Do
289	66-674	2700	034		49			YES	NO	Do Do Do
290	66-719	2700	026		49			YES	NO	Do Do Do
291	66-764	2700	026		49			YES	NO	Do Do Do
292	66-809	2700	025		49			YES	YES	
293	66-854	2700	036		49			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
294	66-899	2700	030		49			YES	NO	Do
295	66-944	2700	044		49			YES	NO	Do Do Do
296	66-989	2700	026		49			YES	NO	Do Do Do
297	67-034	2700	030		49			YES	NO	Do Do Do
298	67-079	2700	036		49			YES	NO	Do Do
299	67-124	2700	033		40			YES	YES	
300	67-169	2700	033		40			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
301	67-214	2700	030		40			YES	NO	Do Do Do
302	67-259	2700	038		40			YES	NO	Do Do Do
303	67-304	2700	030		40			YES	NO	Do Do Do
304	67-349	2700	030		40			YES	YES	!
305	67-394	2700	031		40			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
306	67-439	2700	032		40			YES	NO	Do Do
307	67-484	2700	032		40			YES	YES	
308	67-529	2700	0325		26			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
309	67-574	2700	032		50			YES	NO	Do Do
310	67-619	2700	033		50			YES	YES	!
311	67-664	2700	033		50			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
312	67-709	2700	033		50			YES	NO	Do Do
313	67-754	2700	033		50			YES	NO	Do Do
314	67-799	2700	032		23			YES	YES	
315	67-844	2700	033		23			YES	NO	EXTERNAL MIXTURE EXPLOSIVE
316	67-889	2700	032		23			YES	NO	Do Do
317	67-934	2700	032		23			YES	NO	Do Do
318	67-979	2700	032		23			YES	NO	Do Do
319	68-024	2700	031		23			YES	NO	Do Do
320	68-069	2700	031		16			YES	NO	Do Do
321	68-114	2700	031		16			YES	YES	

No	GAS METER READING	GAS IN CUBIC FT E.C. 58	CURRENT IN AMPERE	BREADTH OF FRANGE	GAP		IGNITION		REMARKS
					LENGTH IN INCHES	DEPTH IN INCHES	INTERNAL	EXTERNAL	
322	16.16-16.636	7.700	16	1"	1.625	.038			EXTERNAL MIXTURE EXPLOSIVE
323	16.240	7.700	16	"	"	"	Yes	Yes	
324	16.007-16.003	7.700	16	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
325	16.146-16.147	7.700	60	"	"	"	Yes	No	Do Do
326	16.242	7.700	60	"	"	"	Yes	No	Do Do
327	16.311-16.312	7.700	60	"	"	"	Yes	No	Do Do
328	16.231	7.700	60	"	"	"	Yes	No	Do Do
329	16.231	7.700	60	"	"	"	Yes	No	Do Do
330	16.231	7.700	60	"	"	"	Yes	No	Do Do
331	16.231	7.700	60	"	"	"	Yes	No	Do Do
332	16.231	7.700	60	"	"	"	Yes	No	Do Do
333	16.231	7.700	60	"	"	"	Yes	No	Do Do
334	16.231	7.700	60	"	"	"	Yes	Yes	
335	16.231	7.700	78	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
336	16.231	7.700	78	"	"	"	Yes	No	Do Do
337	16.231	7.700	78	"	"	"	Yes	No	Do Do
338	16.231	7.700	78	"	"	"	Yes	No	Do Do
339	16.231	7.700	78	"	"	"	Yes	No	Do Do
340	16.231	7.700	78	"	"	"	Yes	No	Do Do
No	GAS METER READING	GAS IN CUBIC FT E.C. 58	CURRENT IN AMPERE	BREADTH OF FRANGE	GAP		IGNITION		REMARKS
					LENGTH IN INCHES	DEPTH IN INCHES	INTERNAL	EXTERNAL	
341	16.231	7.700	20	3.5"	1.605	.061	Yes	No	EXTERNAL MIXTURE EXPLOSIVE DUST PLACED BETWEEN FRANGES
342	16.231	7.700	"	"	"	.059	Yes	No	Do Do Do Do
343	16.231	7.700	"	"	"	"	Yes	No	Do Do Do Do
344	16.231	7.700	15	3.5"	.054	"	Yes	No	ONLY BETWEEN FRANGES SHOWS DUST EXPLOSION FOLLOWING DUST BEING SWIRLED AROUND THE WALL COVERED
345	16.231	7.700	"	"	"	"	No	No	Do Do
346	16.231	7.700	"	"	"	"	Yes	No	Do Do
347	16.231	7.700	"	"	"	"	Yes	No	FOR REMOVAL OF FRANGE PASSED
348	16.231	7.700	"	"	"	"	Yes	No	Do Do Do Do
349	16.231	7.700	"	"	"	"	Yes	No	Do Do
350	16.231	7.700	"	"	"	"	Yes	No	LESS VIOLENT EXPLOSION
351	16.231	7.700	"	"	"	"	Yes	No	Do Do
352	16.231	7.700	"	"	"	"	Yes	No	Do Do
353	16.231	7.700	"	"	"	"	Yes	No	ONLY A SMALL QUANTITY OF DUST REMAINING
354	16.231	7.700	"	"	"	"	Yes	No	DUST ALL CONSUMED
355	16.231	7.700	3.5"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE FEEBLE INTERNAL EXPLOSION
356	16.231	7.700	"	"	"	"	Yes	No	Do Do Do
357	16.231	7.700	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE ONLY FILM ON FRANGES CONDENSED
358	16.231	7.700	"	"	"	"	Yes	Yes	TWO DISTINCT EXPLOSIONS
359	16.231	7.700	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
360	16.231	7.700	"	"	"	"	Yes	No	Do Do Do
361	16.231	7.700	"	"	"	"	Yes	Yes	COVER OF BOX BROKEN BY FORCE OF EXPLOSION
362	16.231	7.700	3.5"	"	"	.056	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
363	16.231	7.700	"	"	"	"	Yes	No	Do Do
364	16.231	7.700	"	"	"	"	Yes	No	Do Do

No	Gas Meter Reading	Gas in Cubic Feet		Coal Dust in Grams	Current in Amperes	GAP		Ignition		Remarks
		EC	SB			Length inches	Depth inches	Internal	External	
365	85.442-85.447 88.617	2.700	.024	3.5	15	1625	.038	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
366	88.618-88.623 91.862	2.700	.026	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
367	91.863-91.868 94.068	2.700	.025	"	"	"	"	Yes	No	D ₀
368	94.069-94.074 97.324	2.700	.026	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
369	97.325-97.330 100.575	2.700	.024	3.5	ADDED	"	"	Yes	No	D ₀ D ₀ D ₀
370	100.576-100.581 103.826	2.700	.025	"	"	"	"	Yes	No	D ₀ D ₀
371	103.827-103.832 107.077	2.700	.027	"	"	"	"	Yes	No	D ₀ D ₀
372	107.078-107.083 110.328	2.700	.026	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
373	110.329-110.334 113.579	2.700	.026	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
374	113.580-113.585 116.830	2.700	.026	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
375	116.831-116.836 120.081	2.700	.027	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
376	120.082-120.087 123.332	2.700	.027	3.5	ADDED	"	"	Yes	No	D ₀ D ₀ D ₀
377	123.333-123.338 126.583	2.700	.027	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
378	126.584-126.589 129.834	2.700	.027	"	"	"	"	Yes	Yes	COVER OF BOX BROKEN BY FORCE OF EXPLOSION
379	129.835-129.840 133.085	2.700	.030	3.5	ADDED	18	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
380	133.086-133.091 136.336	2.700	.029	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
381	136.337-136.342 139.587	2.700	.029	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
382	139.588-139.593 142.838	2.700	.028	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
383	142.839-142.844 146.089	2.700	.029	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
384	146.090-146.095 149.340	2.700	.030	NONE	"	"	"	Yes	Yes	ONLY SMALL QUANTITY OF DUST LEFT
385	149.341-149.346 152.591	2.700	.030	"	"	"	"	Yes	Yes	
386	152.592-152.597 155.842	2.700	.028	"	"	"	"	Yes	Yes	
387	155.843-155.848 159.093	2.700	.026	"	"	"	"	Yes	Yes	
388	159.094-159.099 162.344	2.700	.028	3.5	ADDED	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
389	162.345-162.350 165.595	2.700	.024	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
390	165.596-165.601 168.846	2.700	.024	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
391	168.847-168.852 172.097	2.700	.030	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
392	172.098-172.103 175.348	2.700	.030	"	20	"	"	Yes	No	D ₀ D ₀ D ₀
393	175.349-175.354 178.604	2.700	.033	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
394	178.605-178.610 181.855	2.700	.031	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
395	181.856-181.861 185.111	2.700	.031	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
396	185.112-185.117 188.367	2.700	.029	"	"	"	"	Yes	Yes	
397	188.368-188.373 191.623	2.700	.035	3.5	ADDED	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
398	191.624-191.629 194.879	2.700	.030	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
399	194.880-194.885 198.135	2.700	.031	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
400	198.136-198.141 201.391	2.700	.030	"	"	"	"	Yes	Nil	RED SPARKS AND FLAMES AT GAP
401	201.392-201.397 204.647	2.700	.031	"	"	"	"	Yes	Nil	D ₀ D ₀
402	204.648-204.653 207.903	2.700	.031	"	"	"	"	Yes	Nil	D ₀ D ₀
403	207.904-207.909 211.159	2.700	.032	3.5	ADDED	"	"	Yes	Nil	LESS FLAME.
404	211.160-211.165 214.415	2.700	.030	"	"	"	"	Yes	Nil	D ₀ D ₀
405	214.416-214.421 217.671	2.700	.030	"	"	"	"	Yes	Nil	D ₀ D ₀
406	217.672-217.677 220.927	2.700	.030	3.5	ADDED	25	"	Yes	Nil	FAIRLY VIOLENT, SHORT WHITE FLAME AT OPENING.
407	220.928-220.933 224.183	2.700	.035	"	"	"	"	Yes	Nil	VIOLENT EXPLOSION
408	224.184-224.189 227.439	2.700	.043	"	"	"	"	Yes	Nil	LESS VIOLENT EXPLOSION
409	227.440-227.445 230.695	2.700	.040	"	"	"	"	Yes	Nil	TWO REPORTS HEARD.
410	230.696-230.701 233.951	2.700	.045	"	"	"	"	Yes	Nil	D ₀ D ₀

No	GAS METER READING	GAS IN Cubic Feet		COAL DUST IN GRAMS	CURRENT IN AMPERES	GAP		IGNITION		REMARKS
		EC	SB			LENGTH IN INCHES	DEPTH IN INCHES	INTERNAL	EXTERNAL	
411	100-100	Nil	.042	3.5	25	1.00	.05	Y	Nil	RED FLAMES AT GAP.
412	100-100	Nil	.049	"	11	"	"	YES	Nil	VERY WEAK EXPLOSION.
413	100-100	Nil	.042	"	"	"	"	YES	Nil	
414	100-100	Nil	.042	"	"	"	"	YES	Nil	Violent Explosion
415	100-100	Nil	.047	"	"	"	"	YES	Nil	
416	100-100	Nil	.047	"	"	"	"	YES	Nil	
417	100-100	Nil	.030	3.5	"	"	"	YES	Nil	Violent Explosion Fair Amount of Flame at Gap
418	1-017	Nil	.037	"	"	"	"	YES	Nil	D ₀ D ₀
419	1-061	Nil	.044	"	"	"	"	YES	Nil	Less Violent Explosion Flame Seeds Sharper
420	1-101	Nil	.040	"	"	"	"	YES	Nil	D ₀ D ₀
421	1-125	Nil	.036	"	"	"	"	YES	Nil	Fairly Violent Fair Amount of Flame
422	1-136	Nil	.047	"	"	"	"	YES	Nil	D ₀ D ₀
423	100-100	Nil	.048	"	"	"	"	YES	Nil	D ₀ D ₀
424	1-250	Nil	.025	"	"	"	"	YES	Nil	D ₀ D ₀
425	1-275	Nil	.023	3.5	"	"	"	YES	Nil	D ₀ D ₀
426	1-300	2.700	.025	"	"	"	"	YES	No	EXTERNAL Mixture Explosive
427	1-300-2.650	2.700	.030	"	"	"	"	YES	YES	
428	1-300	2.700	.030	"	"	"	"	YES	No	EXTERNAL Mixture Explosive
429	1-300-2.650	2.700	.030	3.5	"	"	"	YES	No	D ₀ D ₀ D ₀
430	1-300-2.650	2.700	.030	"	"	"	"	YES	YES	Less Violent Ignition Fair Amount of Flame When Box Opened. Back Striking EXTERNAL Mixture Explosive
431	1-300-2.650	2.700	.030	"	"	"	"	YES	No	
432	1-300-2.650	2.700	.030	"	17	"	"	YES	No	D ₀ D ₀ D ₀
433	1-300-2.650	2.700	.030	"	20	"	"	YES	No	D ₀ D ₀ D ₀
434	1-300-2.650	2.700	.030	"	30	"	"	YES	No	D ₀ D ₀ D ₀
435	1-300-2.650	2.700	.030	"	40	"	"	YES	No	D ₀ D ₀ D ₀
436	1-300-2.650	2.700	.030	"	60	"	"	YES	No	D ₀ D ₀ D ₀
437	1-300-2.650	2.700	.035	"	60	"	"	YES	YES	DUST ALL CONSUMED
438	1-300-2.650	2.700	.045	3.5	"	"	"	YES	No	Feeble Internal Explosion EXTERNAL Mixture Explosive
439	1-300-2.650	2.700	.045	"	"	"	"	YES	No	D ₀ D ₀ D ₀
440	1-300-2.650	2.700	.034	"	"	"	"	YES	No	Violent Internal Explosion EXTERNAL Mixture Explosive
441	1-300-2.650	2.700	.044	"	"	"	"	YES	No	Less Violent Internal Explosion EXTERNAL Mixture Explosive
442	1-300-2.650	2.700	.033	"	"	"	"	YES	No	Violent Internal Explosion EXTERNAL Mixture Explosive
443	1-300-2.650	2.700	.034	"	"	"	"	YES	No	D ₀ D ₀ D ₀
444	1-300-2.650	2.700	.034	"	"	"	"	YES	YES	TRACES OF DUST ON-T
445	1-300-2.650	2.700	.030	NONE	60	.875	.050	YES	No	WUBBO CO. H. 1250 EXTERNAL Mixture Explosive
446	1-300-2.650	2.700	.035	"	"	"	"	YES	YES	
447	1-300-2.650	2.700	.030	"	"	"	"	YES	YES	
448	1-300-2.650	2.700	.035	"	"	"	"	YES	No	EXTERNAL Mixture Explosive
449	1-300-2.650	2.700	.034	"	"	"	"	YES	No	D ₀ D ₀ D ₀
450	1-300-2.650	2.700	.033	"	"	"	"	YES	No	D ₀ D ₀ D ₀
451	1-300-2.650	2.700	.033	"	"	"	"	YES	No	D ₀ D ₀ D ₀
452	1-300-2.650	2.700	.026	"	"	"	"	YES	No	D ₀ D ₀ D ₀
453	1-300-2.650	2.700	.027	"	"	"	"	YES	No	D ₀ D ₀ D ₀
454	1-300-2.650	2.700	.026	"	"	"	"	YES	No	D ₀ D ₀ D ₀
455	1-300-2.650	2.700	.027	"	"	"	"	YES	No	D ₀ D ₀ D ₀

No	Gas Meter Reading	Gas in Casing Foot		Current in Amperes	Breakdown of Range	GAP		Ignition		Remarks
		EC	S.B			Length in Inches	Depth in Inches	Int.	Ext.	
156	9501-9503		.025	60	1"	.575	.080	Yes	No	Webbed Cover Used External Mixture Explosive
157	96-112	2-700	.026	"	"	"	"	Yes	No	Do Do Do
158	96-563	2-700	.027	"	"	"	"	Yes	No	Do Do Do
159	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
160	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
161	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
162	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
163	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
164	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
165	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
166	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
167	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
168	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
169	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
170	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
171	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
172	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
173	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
174	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
175	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
176	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
177	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
178	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
179	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
180	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
181	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
182	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
183	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
184	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
185	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
186	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
187	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
188	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
189	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
190	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
191	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
192	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
193	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
194	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
195	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
196	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
197	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
198	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
199	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
200	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do
201	96-572	2-700	.028	"	"	"	"	Yes	No	Do Do Do

No	GAS METER		GAS IN CUM. FEET	CURRENT IN AMPERES	DENSITY OF FLAME	G A P		IGNITION		REMARKS
	READING	EC 58				LENGTH IN INCHES	DEPTH IN INCHES	INT	EXT	
501	19-03-19-11	2700	025	60	1"	5-5	0-40	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
502	22-371	2700	030	"	"	"	"	Yes	No	Do Do Do
503	25-071	2700	030	"	"	5-5	0-40	Yes	No	Do Do Do
504	27-902	2700	030	"	"	"	"	Yes	No	Do Do Do
505	27-922	2700	030	"	"	"	"	Yes	Yes	Do Do Do
506	30-632	2700	030	"	"	"	"	Yes	Yes	Do Do Do
507	30-642	2700	030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
508	32-161	2700	035	"	"	"	"	Yes	No	Do Do Do
509	32-161	2700	036	"	"	6-8	0-40	Yes	No	Do Do Do
510	31-367	2700	034	"	"	"	"	Yes	No	Do Do Do
511	41-240-41-247	2700	034	"	"	"	"	Yes	No	Do Do Do
512	46-337	2700	030	"	"	"	"	Yes	No	Do Do Do
513	47-137	2700	030	"	"	"	"	Yes	No	Do Do Do
514	47-167	2700	030	"	"	"	"	Yes	No	Do Do Do
515	48-167	2700	031	"	"	"	"	Yes	No	Do Do Do
516	49-197	2700	031	"	"	"	"	Yes	No	Do Do Do
517	50-294	2700	030	"	"	"	"	Yes	No	Do Do Do
518	52-628	2700	030	"	"	"	"	Yes	No	Do Do Do
519	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
520	55-328	2700	030	"	"	10-25	0-40	Yes	Yes	Do Do Do
521	55-328	2700	030	"	"	"	"	Yes	Yes	Do Do Do
522	55-328	2700	030	"	"	"	"	Yes	Yes	Do Do Do
523	55-328	2700	030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
524	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
525	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
526	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
527	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
528	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
529	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
530	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
531	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
532	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
533	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
534	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
535	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
536	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
537	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
538	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
539	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
540	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
541	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
542	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
543	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
544	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do
545	55-328	2700	030	"	"	"	"	Yes	No	Do Do Do

No.	GAS METER READING	GAS IN CONCENTRATION		CURRENT IN AMPERES	BEARING OF FLAME	GAP		IGNITION		REMARKS
		EC	S.B.			LENGTH IN INCHES	DEPTH IN INCHES	INT.	EXT.	
546	61.731	2700	+031	18	1"	1.875	.052	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
547	61.762	2700	+031	"	"	1.95	"	Yes	No	DO DO DO FINE EXPLOSIVE 1" BORE BOMBING CASE
548	61.792	2700	+030	"	"	"	"	Yes	No	DO DO DO
549	61.822	2700	+030	"	"	2.00	"	Yes	No	DO DO DO
550	61.852	2700	+030	"	"	"	"	Yes	No	DO DO DO
551	61.882	2700	+030	"	"	2.05	"	Yes	No	DO DO DO
552	61.912	2700	+030	"	"	"	"	Yes	No	DO DO DO
553	61.942	2700	+030	"	"	2.10	"	Yes	No	DO DO DO
554	61.972	2700	+030	"	"	"	"	Yes	No	DO DO DO
555	62.002	2700	+030	"	"	2.25	"	Yes	No	DO DO DO
556	62.032	2700	+030	"	"	"	"	Yes	No	DO DO DO
557	62.062	2700	+030	"	"	"	"	Yes	No	DO DO DO
558	62.092	2700	+030	"	"	"	"	Yes	No	DO DO DO OLD BOX WITH WEARER COVER
559	62.122	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
560	62.152	2700	+030	"	"	2.30	"	Yes	No	DO DO DO DO
561	62.182	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
562	62.212	2700	+030	"	"	2.375	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE OLD BOX AND OLD COVER (FURNED)
563	62.242	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
564	62.272	2700	+030	"	"	2.50	"	Yes	No	DO DO DO DO
565	62.302	2700	+030	"	"	2.54	"	Yes	No	DO DO DO DO
566	62.332	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
567	62.362	2700	+030	"	"	2.63	"	Yes	No	DO DO DO DO
568	62.392	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
569	62.422	2700	+030	"	"	2.75	"	Yes	No	DO DO DO DO
570	62.452	2700	+030	"	"	2.875	"	Yes	No	DO DO DO DO
571	62.482	2700	+030	"	"	3.00	"	Yes	No	DO DO DO DO
572	62.512	2700	+030	"	"	3.125	"	Yes	No	DO DO DO DO
573	62.542	2700	+030	"	"	3.25	"	Yes	No	DO DO DO DO
574	62.572	2700	+030	"	"	3.375	"	Yes	No	DO DO DO DO
575	62.602	2700	+030	"	"	3.5	"	Yes	Yes	BOLT SHEARED
576	62.632	2700	+030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE (OLD BOX AND OLD COVER)
577	62.662	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
578	62.692	2700	+030	"	"	3.625	"	Yes	No	DO DO DO DO
579	62.722	2700	+030	"	"	3.750	"	Yes	No	DO DO DO DO
580	62.752	2700	+030	"	"	3.875	"	Yes	No	DO DO DO DO
581	62.782	2700	+030	"	"	4.0	"	Yes	No	DO DO DO DO
582	62.812	2700	+030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE (NEW BOX COMPLETE (C.S.))
583	62.842	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
584	62.872	2700	+030	"	"	4.25	"	Yes	Yes	(NEW BOX)
585	62.902	2700	+030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE (OLD BOX AND OLD COVER)
586	62.932	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
587	62.962	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
588	62.992	2700	+030	"	"	"	"	Yes	No	NEW BOX COMPLETE (C.S.) BOLT FOUND WITH STRIPPED THREAD
589	63.022	2700	+030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE (NEW BOX WITH OLD COVER)
590	63.052	2700	+030	"	"	"	"	Yes	No	DO DO DO DO
591	63.082	2700	+030	"	"	"	"	Yes	Yes	OLD BOX WITH NEW COVER

No	GAS METER READING	GAS IN CUBIC FEET		CURRENT IN AMPERES	BREADTH OF FLANGE	GAP		IGNITION		REMARKS
		E.C.	S.B.			LENGTH IN INCHES	DEPTH IN INCHES	INTERNAL	EXTERNAL	
592	87-100-87-400 90-160		.030	18	1	4-28	.052	Yes	No	EXTERNAL MIXTURE EXPLOSIVE [OLD BOX WITH RIBBED COVER]
593	92-150-90-160 92-110	2-700	.030	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
594	92-110 95-610	2-700	.020	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
595	95-640 91-310	2-700	.030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE [NEW BOX COMPLETE]
596	91-310 101-230	2-700	.030	"	"	4-375	"	Yes	No	D ₀ D ₀ D ₀
597	101-230 3-960	2-700	.030	"	"	4-500	"	Yes	Yes	TWO DISTINCT EXPLOSIONS
598	3-960 6-690	2-700	.030	"	"	"	"	Yes	Yes	
599	6-690 8-550	2-700	.030	"	"	"	.050	Yes	No	EXTERNAL MIXTURE EXPLOSIVE [OLD BOX WITH OLD COVER]
	8-550 12-260	2-700	.030	"	"	"	"	Yes	No	D ₀ D ₀
601	12-260 15-010	2-700	.030	"	"	"	"	Yes	No	D ₀ D ₀
602	15-010 18-310	2-700	.030	"	"	"	"	Yes	No	D ₀ D ₀
603	18-310 21-610	2-700	.030	"	"	"	.052	Yes	Yes	
604	21-610 23-760	2-700	.030	"	"	PERIPHERY OF BOX	.050	Yes	Yes	CASE STEEL BOX WITH STUDS SET IN FLANGES TO GIVE GAP
605	23-760 26-430	2-700	.030	"	"	"	"	Yes	Yes	COVER AS ABOVE WARPED AT ENDS AND SIDES .01 AND .003 RESPECTIVELY
606	26-430 28-110	2-700	.030	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE. STUDS RASA DOWN TO REDUCE GAP
607	28-110 32-140	2-700	.030	"	"	"	"	Yes	No	D ₀ D ₀
608	32-140 34-510	2-700	.030	"	"	"	"	Yes	No	D ₀ D ₀

E.C. MEANS EXPLOSION CHAMBER.

S.B. MEANS SWITCH BOX.

No	Gas Meter Reading	Gas Pressure (psi)	Flow Rate (gpm)	Time (min)	Bar (psi)	Temp (°F)	Remarks	Remarks
604	34.23-34.95	2.10	0.10	18	175	34.9	1	EXTERNAL MIXTURE EXPLOSIVE, HEATING DURING FILLING OF E.C. Mixture filling of E.C. Mixture filling of E.C.
610	37.30	2.10	0.25					D ₀ D ₀ D ₀ D ₀
611	37.30-37.30	2.10	0.25					D ₀ D ₀ D ₀ D ₀
612	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
613	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
614	37.30-37.30	2.10	0.30					HEATING DURING FILLING OF E.C. Mixture filling of E.C. Mixture filling of E.C.
615	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
616	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
617	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
618	37.30-37.30	2.10	0.30					HEATING DURING FILLING OF E.C.
619	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
620	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
621	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
622	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
623	37.30-37.30	2.10	0.30					EXTERNAL MIXTURE EXPLOSIVE, HEATING DURING FILLING OF E.C.
624	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
625	37.30-37.30	2.10	0.30					HEATING DURING FILLING OF E.C.
626	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
627	37.30-37.30	2.10	0.30					EXTERNAL MIXTURE EXPLOSIVE, HEATING DURING FILLING
628	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
629	37.30-37.30	2.10	0.30					EXTERNAL MIXTURE EXPLOSIVE, HEATING DURING FILLING
630	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
631	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀
632	37.30-37.30	2.10	0.30					EXTERNAL MIXTURE EXPLOSIVE, HEATING DURING FILLING
633	37.30-37.30	2.10	0.30					D ₀ D ₀ D ₀ D ₀

[illegible]

No	Gas Meters Reading	EC	SB	Current in Amps	Barometer inches	Amplitude inches	Grav inches	Conductance in ohms	Resistance in ohms	Ignition inches	Remarks
54	95.600-95.950	29.0	0.30	15	NIL	30	57	1.75	0.50	Yes	EXTERNAL MIXTURE EXPLOSIVE
55	96.000-96.350	29.0	0.30	2	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
56	96.350-96.700	29.0	0.30	2	247	53	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
57	96.700-97.050	29.0	0.30	2.5	247	56	"	"	"	No	EXTERNAL MIXTURE EXPLOSIVE
58	97.050-97.400	29.0	0.30	3.1	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
59	97.400-97.750	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
60	97.750-98.100	29.0	0.30	NIL	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
61	98.100-98.450	29.0	0.30	2	"	57	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
62	98.450-98.800	29.0	0.30	"	"	"	"	"	"	No	EXTERNAL MIXTURE EXPLOSIVE
63	98.800-99.150	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
64	99.150-99.500	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
65	99.500-99.850	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
66	99.850-100.200	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
67	100.200-100.550	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
68	100.550-100.900	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
69	100.900-101.250	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
70	101.250-101.600	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
71	101.600-101.950	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
72	101.950-102.300	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
73	102.300-102.650	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
74	102.650-103.000	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
75	103.000-103.350	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
76	103.350-103.700	29.0	0.30	"	"	60	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
77	103.700-104.050	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
78	104.050-104.400	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
79	104.400-104.750	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
80	104.750-105.100	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
81	105.100-105.450	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
82	105.450-105.800	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE
83	105.800-106.150	29.0	0.30	"	"	"	"	"	"	Yes	EXTERNAL MIXTURE EXPLOSIVE

[illegible]

No	Gas Meter Reading	Gas in Cylinders	Exhaust in 1000 ft	Atmosphere Bar	Temp of Bar	Speed of Revs	G-A Lens in 100 ft	Caliber	Range	AD	SD	Remarks
709	25.000	2500	18	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
710	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
711	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
712	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
713	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
714	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
715	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
716	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
717	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
718	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
719	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
720	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
721	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
722	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
723	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
724	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
725	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
726	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
727	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
728	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
729	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
730	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
731	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
732	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.
733	25.000	2500	25	29.8	56	1	1375	0.5	9.5	187	187	EXTERNAL MIXTURE EXPLOSIVE.

No	Gas Meter Reading	Gas 18 EC SB	Gas 18 16000 14000 12000 10000 8000 6000 4000 2000 0	Approximate But Eye 100 50 0	Time 1 5 10 15 20 25 30 35 40 45 50 55	GA P 1 5 10 15 20 25 30 35 40 45 50 55	Quantity of Gas 1 5 10 15 20 25 30 35 40 45 50 55	Cal Res. 1 5 10 15 20 25 30 35 40 45 50 55	INTERNAL Temp. of Gas 1 5 10 15 20 25 30 35 40 45 50 55	IGNITION 1 5 10 15 20 25 30 35 40 45 50 55	REMARKS.
734	54.30-54.20	520	18 Nil	56	1	5.05-06.3	-	-	56 56	Yes No	EXTERNAL MIXTURE EXPLOSIVE
735	54.20-54.10	520	-	-	-	-	-	-	56 56	Yes	
736	54.10-54.00	520	-	-	-	-	-	-	56 56	Yes	
737	54.00-53.90	520	5	56	-	-	4	4	111	Yes	
738	53.90-53.80	520	-	-	-	-	10	7	146	Yes No	EXTERNAL MIXTURE EXPLOSIVE
739	53.80-53.70	520	-	-	-	-	5	4	126	Yes	
740	53.70-53.60	520	-	-	-	-	8	6	167	Yes	
741	53.60-53.50	520	-	-	-	-	4.5	3.5	119	Yes	
742	53.50-53.40	520	-	-	-	-	3	3	98	Yes No	EXTERNAL MIXTURE EXPLOSIVE
743	53.40-53.30	520	2.7	-	-	-	8	6	167	Yes No	D ₀
744	53.30-53.20	520	Nil	-	-	-	5	4.5	126	Yes No	D ₀
745	53.20-53.10	520	-	-	-	-	4	4	112	Yes	
746	53.10-53.00	520	-	-	-	-	1.5	1.5	78	Yes	
747	53.00-52.90	520	2.7 2.45 5.7	-	-	-	7	6	154	Yes	
748	52.90-52.80	520	Nil	-	-	-	4.5	4	119	Yes No	EXTERNAL MIXTURE EXPLOSIVE
749	52.80-52.70	520	3.7	-	-	-	7	6	152	Yes No	D ₀
750	52.70-52.60	520	Nil	-	-	-	5	4.5	119	Yes No	D ₀
751	52.60-52.50	520	2.7	-	-	-	6	5	126	Yes No	D ₀
752	52.50-52.40	520	-	-	-	-	7	6	151	Yes	
753	52.40-52.30	520	Nil	-	-	-	5	4.5	119	Yes	
754	52.30-52.20	520	-	-	-	-	3	3	98	Yes No	EXTERNAL MIXTURE EXPLOSIVE
755	52.20-52.10	520	-	-	-	-	3	3	98	Yes No	D ₀
756	52.10-52.00	520	-	54	-	10.5-05.7	-	-	54	Yes	
757	52.00-51.90	520	-	-	-	-	-	-	54	Yes	
758	51.80-51.70	520	5	-	-	0.35	-	-	54	Yes No	EXTERNAL MIXTURE EXPLOSIVE

[illegible]

[illegible]

No	Gas Meter Reading	Gas in (lb/ft ³) EC 5.B	Starting Current in Amperes	Breadth of Flange	Shaft Bearing Diameter	Shaft Diameter	Dia. of Bearing Min. Dia. of Shaft	IGNITION		REMARKS	
								INTERNAL	EXTERNAL		
805	17.800-17.550	5.50	18	1.375	20"	2"	.0015	YES	NO	EXTERNAL MIXTURE EXPLOSIVE	PAPER MANDREL BURST BY INTERNAL EXPLOSION
806	71.350	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
807	73.100	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
808	73.650	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
809	74.100	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
810	74.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
811	75.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
812	75.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
813	76.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
814	76.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
815	77.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
816	77.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
817	78.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
818	78.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
819	79.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
820	79.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
821	80.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
822	80.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
823	81.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
824	81.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
825	82.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
826	82.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
827	83.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
828	83.700	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀
829	84.200	5.50	"	"	"	"	"	YES	NO	D ₀	D ₀

No.	Gas Meter Reading	Gas in Tube Feet		Initial Gauge in Times	Revolutions of Flame	White Noise Bearing Diameter	Shaft Diameter	Dia. of Bearing Minus Dia. of Shaft	Ignition Internal	Ignition External	REMARKS	
		F.C.	S.B.									
830	46-150-1,000	—	550	18	1-375	2.0	2.5	1.965	Yes	—	SPARKS AT BEARING. WHITE EMANATION EXTENDING BEYOND BEARING. TEMPERATURE OF SHAFT 88°F.	D ₀
831	47-500	—	550	—	—	—	—	—	Yes	—	—	D ₀
832	48-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
833	4800-10,500	2,500	600	—	—	—	—	—	Yes	No	EXTERNAL MIXTURE EXPLOSIVE	D ₀
834	49-050	2,500	600	—	—	—	—	—	Yes	No	—	D ₀
835	50-000	2,500	600	—	—	—	—	—	Yes	No	—	D ₀
836	51-000	2,500	600	—	—	—	—	—	Yes	No	WHITE DEPOSIT (PROBABLY SMOKE) ON SIDE OF EXPLOSION CHAMBER NINE INCHES EACH BEARING.	D ₀
837	52-050	2,500	600	—	—	—	—	—	Yes	No	STUD OUT SHAFT DRAIN IN	D ₀
838	53-000	2,500	600	—	—	—	—	—	Yes	No	NUMEROUS SPARKS AND SHORT FLAME AT BEARING	D ₀
839	54-000	2,500	600	—	—	—	—	—	Yes	No	BROWNISH FLAME EXTENDING ONE INCH BEYOND BEARING	D ₀
840	55-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
841	56-050	—	550	—	—	—	—	—	Yes	—	—	D ₀
842	57-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
843	58-050	—	550	—	—	—	—	—	Yes	—	—	D ₀
844	59-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
845	60-050	—	550	—	—	—	—	—	Yes	—	—	D ₀
846	61-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
847	62-050	—	550	—	—	—	—	—	Yes	—	—	D ₀
848	63-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
849	64-050	—	550	—	—	—	—	—	Yes	—	—	D ₀
850	65-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
851	66-050	—	550	—	—	—	—	—	Yes	—	—	D ₀
852	67-000	—	550	—	—	—	—	—	Yes	—	—	D ₀
853	68-050	—	550	—	—	—	—	—	Yes	—	—	D ₀

No	Gas Meter Reading	Gas in Cubic Feet	Gas in Cubic Feet S.C.	Ignition in Apparatus	Bearing Diameter	Bearing Length	Shaft Diameter	DIA. OF BEARING MINUS DIA. OF SHAFT	Ignition		Remarks
									Internal	External	
854	46.430-46.200	-	1.20	14	1.375	20" 2.5"	1.955	.045	Yes	-	RED FLAME EXTENDING TWO INCHES BEYOND BEARING
855	46.200	-	1.30	"	"	"	"	"	Yes	-	D ₀ D ₀ D ₀
856	46.175	-	1.75	"	"	"	"	"	Yes	-	D ₀ D ₀ D ₀ TIP OF SHAFT 100°F
857	46.150	-	1.75	"	"	"	"	"	Yes	-	SHAFT, BEARING AND FLANGES OILED. SHAFT AT BEARING AND FLAME VERY SHORT. SHAFT OUTSIDE BEARING COVERED WITH RESIDUE OF OIL.
858	47.125	-	1.75	"	"	"	"	"	Yes	-	D ₀ D ₀ D ₀
859	47.350-47.125	-	1.75	"	"	"	"	"	Yes	No	EXTERNAL MIXTURE EXPLOSIVE
860	47.300	2.500	1.75	"	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
861	47.275	2.500	1.75	"	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
862	47.250	2.800	1.75	"	"	"	"	"	Yes	No	D ₀ D ₀ D ₀ STUD OUT. SHAFT DRAWN INSIDE
863	47.225	2.125	1.75	"	"	"	"	"	Yes	No	SWITCH BALL EXTENDING COMPLETELY AFTER THE FLAME DRAWN INTO SHAFT. FLAME EXTENDING
864	46.300	3.000	1.75	"	"	"	"	"	Yes	No	PRODUCED WHEN FLAME DRAWN IN BETWEEN FLANGES AND BEARING
865	46.275	3.000	1.75	"	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
866	47.250	3.000	1.75	"	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
867	47.225	3.000	1.75	"	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
868	47.200	3.000	1.75	"	"	"	"	"	Yes	No	D ₀ D ₀ D ₀
869	46.175	-	1.75	"	"	"	1.945	.055	Yes	-	SHOWER OF WHITE SPARKS AT BEARING
870	47.150	-	1.75	"	"	"	"	"	Yes	-	WHITE FLAME EXTENDING 3-3 INCHES BEYOND BEARING
871	47.125	-	1.75	"	"	"	"	"	Yes	-	PRECEDED BY RED SPARKS D ₀ D ₀ D ₀
872	47.100	-	1.75	"	"	"	"	"	Yes	-	D ₀ D ₀ D ₀
873	47.075	-	1.75	"	"	"	"	"	Yes	-	D ₀ D ₀ D ₀
874	47.050	-	1.75	"	"	"	"	"	Yes	-	STUD OUT. SHAFT DRAWN INSIDE WITHIN BOX CASING SHATTERED BY FORCE OF INTERNAL EXPLOSION.
875	47.025	-	1.30	"	1.0"	1.625	1.625	.001	Yes	-	BRASS BEARING. NO FLAME
876	47.000	-	1.30	"	"	"	"	"	Yes	-	D ₀ D ₀ D ₀
877	46.975	-	1.30	"	"	"	"	"	Yes	-	D ₀ D ₀ D ₀
878	46.950	-	1.30	"	1.650	"	"	.025	Yes	-	D ₀ D ₀ D ₀

No	GAS METER READING	GAS IN CYLINDER		IGNITION CURRENT AMPERES	BREADTH OF FLAME	BEARING		SHAFT DIAMETER	DIA OF BEARING MINUS DIA OF SHAFT	IGNITION		REMARKS
		EC	SB			Diameter	Length			INTERNAL	EXTERNAL	
879	53.70-54.00	—	+0.30	18	1.0	.650	.625	.625	.025	YES	—	SPARKS AT BEARING
880	53.30-53.50	—	+0.30	—	—	.672	—	—	.047	YES	—	SPARKS AND SHORT BLUE FLAME
881	53.30-53.50	—	+0.30	—	—	.687	—	—	.062	YES	—	SPARKS AND RED FLAME EXTENDING APPROXIMATELY ONE INCH BEYOND BEARING
882	53.20-53.50	5400	+0.30	—	—	.657	—	—	—	YES	No	EXTERNAL MINTURE EXPLOSIVE
883	53.70-54.00	5400	+0.30	—	—	.657	—	—	—	YES	No	D ₀ D ₀
884	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	D ₀ D ₀
885	53.70-54.00	5400	+0.30	—	—	.645	—	—	.070	YES	—	BLUE FLAME EXTENDING APPROXIMATELY TWO INCHES BEYOND BEARING
886	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	EXTERNAL MINTURE EXPLOSIVE
887	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	D ₀ D ₀
888	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	D ₀ D ₀ TWO REPEATS HEARD
889	53.70-54.00	5400	+0.30	—	—	.720	—	—	.095	YES	—	SPARKS FOLLOWED BY BLUISH-GREEN FLAME EXTENDING APPROXIMATELY TWO INCHES BEYOND BEARING
890	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	—	D ₀ D ₀
891	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	EXTERNAL MINTURE EXPLOSIVE
892	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	D ₀ D ₀ D ₀
893	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	D ₀ D ₀ D ₀ TWO REPEATS HEARD
894	53.70-54.00	5400	+0.30	—	—	.735	—	—	.110	YES	—	BLUISH-WHITE FLAME EXTENDING 2-3 INCHES BEYOND BEARING
895	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	—	GREEN FLAME EXTENDING 2-3 INCHES BEYOND BEARING (COPPER FUSE USED)
896	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	EXTERNAL MINTURE EXPLOSIVE
897	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	D ₀ D ₀
898	53.70-54.00	5400	+0.30	—	—	.750	—	—	.125	YES	—	GREEN FLAME EXTENDING APPROXIMATELY 3 INCHES BEYOND BEARING (COPPER FUSE USED)
899	53.70-54.00	5400	+0.30	—	—	.750	—	—	—	YES	—	D ₀ D ₀ D ₀
900	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	EXTERNAL MINTURE EXPLOSIVE
901	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	No	D ₀ D ₀ D ₀
902	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	YES	—
903	53.70-54.00	5400	+0.30	—	—	—	—	—	—	YES	YES	—

APPENDIX 1.

INTRODUCTION:

In 1927, the Committee of The Journal of the Royal Technical College, accepted for publication an account of part of the work. A copy of the reprint is herewith presented, together with a mathematical investigation given by way of extension to this paper. The author acknowledges his indebtedness to Professor James Muir, M.A., D.Sc., A.R.T.C., and Mr. Alex. D. Third, B.Sc., Ph.D., A.R.T.C., for assistance given in the mathematical investigation.

The drop in Temperature produced by the passage of Hot Gases through apertures.

The following rough calculations are given by way of extension to the paragraph at the foot of page 119 of the 4th. Number of the Journal of the Royal Technical College (Paper on Preventing Electrical Apparatus from starting Explosions).

The paragraph is as follows:- "The cooling of hot gases, by passing through a gap, may be regarded as being effected in two ways (i) naturally by the sudden expansion from high pressure to low, and (ii) by passing over the cold surfaces of the metal flanges. The heat absorbed by a cold metal surface from a hot gas passing over it, is proportional to the area of the surface, to the density of the gas, to the speed with which the gas sweeps over the surface and to the difference in temperature between the surface and the hot gas. When a 'gap' is short, deep and of small area the first cause of cooling probably predominates; with long narrow gaps cooling is possibly accomplished mainly by the large metal surface swept over."

In considering the passage of hot gases (such as steam) through nozzles, it is found that the passage of the gas may be regarded as being a "reversible adiabatic expansion". This may be a large assumption if applied to the passage of flames from an explosion but it will be made (combustion is not adiabatic). For the reversible adiabatic expansion of ideal gas, $p v^n = \text{Constant}$ and $p v = RT$, hence if the expansion through the gap be from P_1, T_1 , to P_2, T_2 , $T_2 = T_1 \left(\frac{P_2}{P_1} \right)^{\frac{n-1}{n}}$. Taking n (the ratio of the specific heats of the gas at constant pressure and constant volume) as 1.4 and $r = P_2/P_1$, then $T_2 = T_1 r^{0.286}$. If the gap is a convergent or a parallel one the outlet pressure is always 0.53 of the initial pressure (if the final pressure is not above this). If the passage is divergent, fuel expansion can take place within its boundary.

Should/

Should the velocity of the stream at any point in the gap be required it is given by the equation,

$$r = \sqrt{\frac{2gn}{n-1} P_1 V_1 (1 - r^{\frac{n-1}{n}})}$$

where P_1 is in lbs/sq. ft.

V_1 the specific volume in cub.ft/Lb.

r the pressure ratio at the point (referred to P_1)

g the acceleration due to gravity,

or the flow is given by

$$w = a \sqrt{\frac{2gn}{n-1} \frac{P_1}{V_1} (r^{\frac{2}{n}} - r^{1 + \frac{1}{n}})}$$

where w is the flow in lb. per sec.

and 'a' is the area to flow in sq. ft.

CASE I.

The extreme case of the gap $\frac{1}{10}$ " x $\frac{1}{8}$ " x 1".

The form of the passage is of the same order as in nozzle work, the heat transference through the limited surface of metal is negligible when the quantity of gas flowing is considered; the expansion being regarded as adiabatic we have

$$T_2 = T_1 r^{0.286}$$

$$r = \frac{P_2}{P_1} = 0.53 \text{ so } T_2 = 0.83 T_1$$

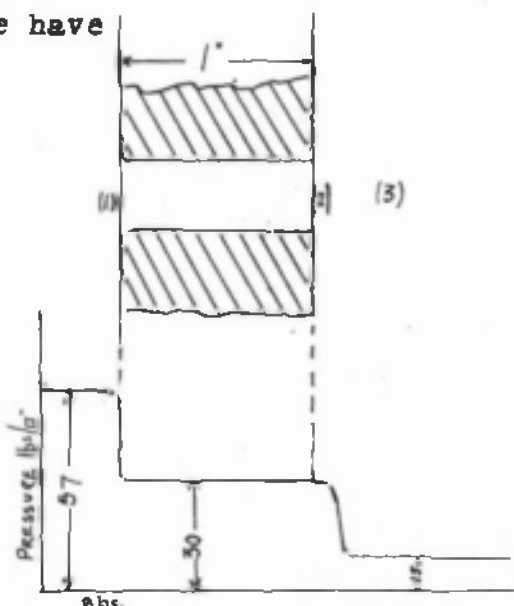
If say $T_1 = 2000^\circ\text{F}$ (absolute)

$T_2 = 1660^\circ\text{F}$ a drop of 340°F

but for this case Fig. 5 p 122 of the Journal paper gives a maximum Pressure $P_1 = 57 \text{ lb/in.}^2$ (abs).

$$\text{So } P_2 = 0.53 \times 57 = 30 \text{ lbs/in}^2$$

$$P_3 = 15 \text{ lbs/in}^2 \text{ (atmospheric.)}$$



There is thus expansion from 30 lbs to 15 lbs/in² after leaving the gap. This expansion takes place very close to the gap and produces further cooling which is probably of importance.

CASE II. Another extreme case - the gap 0.03" x 29" x 1"

Fig. 5, p.122 gives $P_1 = 21 \text{ lb/in}^2 \text{ absol.}$

$$P_2 = P_3 = 15 \text{ lb/in}^2 = (\text{atmos}).$$

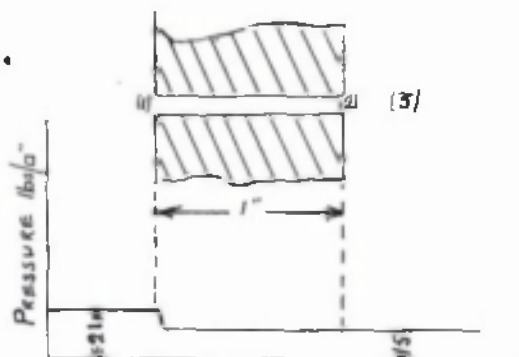
Cooling due to adiabatic expansion.

$$T_2 = T_1 \frac{P_2}{P_1}^{\frac{n-1}{n}}$$

$$= 0.91 T_1$$

If $T_1 = 2000^\circ\text{F}$ (absol) $T_2 = 1800^\circ\text{F}$

A drop of 200°F .



Now consider the heat lost by transmission to the cold metal sides of the gap.

Reynolds' laws of heat transmission, stated in the paragraph quoted at the beginning of this note may be expressed by the equation.

$$H = B \rho v t m \quad [H = A + B \rho v t m - \text{the term } A \text{ for gases at rest being negligible.}]$$

where $t m$ is the mean temperature difference between hot gas and metal; v is the speed of gas, ρ the density, B a constant. H is the heat transmitted per unit area per second.

If w = total weight of gas passing per second and

a = cross-section area of flow, $w = a v \rho$ and

$$H = \frac{w}{a} B t m \quad (\text{Heat transmitted per sq. foot per sec.})$$

For ordinary transmission through boiler tubes $B \doteq \frac{1}{1200}$ to give B in B.T.U. per sq. ft. per sec., w being in lbs/sec, a in sq. ft., $t m$ in $^\circ\text{F}$.

Now T_1 and T_2 being the temperatures of gas on entering and on leaving respectively, then (ignoring natural drop due to expansion) we have

Heat taken from gas = $w.S.(T_1 - T_2)$ where S is the spec. heat of the gas (pressure constant).

$$\text{So } w.S.(T_1 - T_2) = \frac{w}{a} B A t m$$

$$\therefore T_1 - T_2 = \frac{B.A.t m}{S.a.}$$

where A = surface area of metal which is independent of flow of gas as experiments corroborate.

Substituting numbers and taking $S = 0.24$ and the temperature of the hot gases somewhere over 1800°F (absol.) and temperature of metal over 500°F (abs.) so that $t_m = 1300^{\circ}\text{F}$, we get (A and a, both being put in sq. inches.)

$$T_1 - T_2 = \frac{29 \times 2 \times 1300}{1200 \times .24 \times .03 \times 29} = 300^{\circ}\text{F}$$

The total drop in temperature might thus be about 500°F , - 200° due to natural cooling (expansion) and 300° due to transmission to the metal.

Of course the various assumptions made render the calculations of very doubtful value. The experiments might lead one to expect that the drop of 340°F got in Case I should be more nearly equal to the 500°F got in Case II (both gaps were flame-proof), but the further cooling in Case I just after the gas leaves the gap could reasonably be taken to account for this difference, without having regard to a possible difference in the internal temperatures (in one case pressure attained 51 lbs., in the other 21 lbs/in^2 abs.) But 500° seems too small a drop in temperature. If so, then it can only be suggested that calculations based on adiabatic expansion of ideal gases do not apply to gases in a state of combustion.

APPENDIX 11.

BIBLIOGRAPHY.

- Ben & French: Safety devices in Connection with Electrical Machinery and Apparatus in Coal Mines. Trans. Inst. Min. Eng. 1911-12. Vol. XLII, p. 459.
- Clark: Investigation of Explosion Proof Motors - U.S.A. Bureau of Mines. Bulletin 46, 1912.
- Clark: Safety Electric Switches for Mines - U.S.A. Bureau of Mines. Technical Paper. 44, 1913.
- Clark: Permissible Explosion Proof Motors for Mines. Conditions and Requirements for Tests and Approvals - U.S.A. Bureau of Mines. Technical Paper 101, 1915.
- Wheeler: Report on Battery Bell Signalling Systems as regards Danger of Ignition of Firedamp Mixtures by Break Flash at Signal Wires - Home Office Publication 1915.
- Wheeler and Thornton: Report on Electrical Signalling with Bare Wires so far as Regards the Danger of Ignition of Inflammable Mixtures - Home Office Publication 1916.
- Thornton: The Ignition of Coal Gas and Methane by Momentary Electric Arc. - Trans. Inst. Min. Eng. 1912-13 Vol XLIV, p. 145.
- Thornton: The comparative inflammability of Mixtures of Pit Gas and Air Ignited by Momentary Electric Arcs. - Trans. Inst. Min. Eng. Vol. XLVI, pp. 112, 1913-14.
- Thornton: Some Researches on the Safe Use of Electricity in Coal Mines - Journal Inst. Elect. Eng. Vol. 62, p. 481, 1924.
- Thornton & Bowden: The Ignition of Coal Dust by Single Electric Flashes Trans. Inst. Min. Eng., Vol. 39, p.1, part 2, 1910.
- Thornton: The Influence of Presence of Gas on Inflammability of Coal Dust and Air - Colliery Guardian, 19th Sept. 1913.
- Thornton: A New Battery Signalling Bell - Trans. Inst. Min. Eng. Vol. 50, part 1, p. 19, 1915.

Safety in Mines Research Board Papers.

No. 5. Flange Protection (Flame Proof Elect. Apparatus)

No. 20 Electrical Ignition of Firedamp. Alternating
and Continuous Currents Compared.

No. 21. Perforated Plate Protection (Flame Proof Elect.
App.)

Clark & Ilsley: Ignition of Mine Gases by Filaments of Incandescent
lamps. - Bulletin 52. Dept. of the Interior, Bureau
of Mines, U.S.A.

Bulletin 258. U.S.A. Bureau of Mines. On Fasteners, Bolts, Studs,
Screws, Terminals, Lead in Connections. Inter-Compart:
ment Communications, etc.